







# Raport privind Oxidarea Chimică in-situ (ISCO)

Raport final

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### **Introducere privind IMPEL**

Rețeaua Uniunii Europene pentru punerea în aplicare și asigurarea respectării legislației de mediu (IMPEL) este o asociație internațională non-profit a autorităților de mediu din statele membre ale UE, din țările în curs de aderare și din țările candidate la Uniunea Europeană și din țările SEE. Asociația este înregistrată în Belgia, iar sediul său legal fiind în Bruxelles, Belgia.

IMPEL a fost înființată în 1992 drept rețea informală a autorităților europene de reglementare și a autorităților preocupate de punerea în aplicare și respectarea legislației de mediu. Obiectivul rețelei este de a crea impulsul necesar în Comunitatea Europeană pentru a face progrese în ceea ce privește asigurarea unei aplicări mai eficiente a legislației de mediu. Activitățile principale ale rețelei IMPEL se referă la sensibilizarea, consolidarea capacităților și schimbul de informații și de experiențe privind punerea în aplicare, aplicarea și colaborarea internațională în materie de aplicare, precum și la promovarea și sprijinirea practicabilității și a aplicabilității legislației europene de mediu.

În ultimii ani, IMPEL s-a transformat într-o organizație importantă și cunoscută pe scară largă, fiind menționată într-o serie de documente legislative și politice ale UE ca, de exemplu, în cel de-al 7-lea Program de acțiune pentru mediu și în Recomandarea privind criteriile minime pentru inspecțiile de mediu.

Expertiza și experiența participanților din cadrul IMPEL fac ca rețeaua să fie calificată în mod unic pentru a lucra atât la aspectele tehnice, cât și la cele de reglementare ale legislației de mediu a UE.

Informații despre rețeaua IMPEL sunt disponibile și pe site-ul web al acesteia la adresa: www.impel.eu

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Autoritățile competente pentru aprobarea/aplicarea/monitorizarea tehnologiilor de remediere, operatorii industriali, agențiile de protecție a mediului, organismele de protecție a naturii, inspectoratele de mediu, monitorizarea mediului și instituțiile de cercetare, universitățile tehnice, asociațiile de mediu, ONG-urile, companiile și asociațiile de asigurări, consultanții de mediu.

Ca parte a programului de lucru pentru 2020, rețeaua IMPEL a creat acest proiect "Remedierea apei și a terenurilor" (2020/09) ținând cont de criteriile de evaluare a aplicabilității tehnologiilor de remediere.

Proiectul "Water and Land Remediation" are drept punct de plecare definițiile și etapele-cheie ale aplicării tehnologiilor de remediere și se concentrează asupra procedurilor tehnice legate de tehnologiile de remediere. Scopul final al proiectului este de a obține un document care să prezinte criteriile de evaluare a propunerii de aplicare a tehnologiei de remediere, pentru a explica aplicabilitatea, precum și ce trebuie făcut în cadrul testelor de teren și în extrapolarea aplicării acestora la scară largă. Anexa 1 cuprinde o serie de studii de caz, care pot ajuta cititorul să anticipeze orice probleme cu care s-ar putea confrunta și să evalueze dacă soluția oferită se aplică zonei respective, știind că fiecare sit contaminat diferă de altele și că este întotdeauna necesară o abordare specifică pentru fiecare sit în parte.

Obiectivul proiectului de remediere a apei și a solului pentru 2020-2021 a fost să se concentreze pe două tehnologii de remediere, Oxidarea Chimică *In Situ* și Extracția Vaporilor din Sol.

În cele din urmă, proiectul de remediere a apei și a terenurilor intenționează să contribuie la promovarea aplicării tehnologiilor de remediere *in situ* și la fața locului pentru sol și ape subterane și la o mai mică aplicare a tehnologiilor Dig & Dump și Pump & Treat, care sunt tehnici utilizate pe scară largă în Europa, dar care nu sunt sustenabile pe termen mediu și lung. Solul și apa sunt resurse naturale și, atunci când acest lucru este fezabil din punct de vedere tehnic, ar trebui recuperate, nu irosite.

### Mulţumiri

Acest raport a fost revizuit de către o echipă mai largă a proiectului IMPEL și de către echipa de experți IMPEL pentru apă și terenuri, rețeaua COMMON FORUM, rețeaua NICOLE, grupul de lucru EIONET Contaminare și un grup de evaluatori externi.

### Declinarea răspunderii

Această publicație a fost pregătită în cadrul proiectului IMPEL Water & Land Remediation cu sprijinul rețelelor partenere interesate de gestionarea siturilor contaminate. Scris și revizuit de o echipă de autori, documentul de față intenționează să servească drept sursă primară de informații pentru a crea punți de legătură și a extinde cunoștințele între țările și regiunile europene. Această publicație facilitează o înțelegere comună a potențialului tehnologiei specifice de remediere.

Conținutul prezentat aici se bazează pe bibliografia relevantă, pe experiența autorilor și pe studiile de caz colectate. Este posibil ca documentul să nu acopere toate situațiile în care această tehnologie a fost sau va fi aplicată. Studiile de caz (a se vedea anexa) reprezintă contribuții voluntare recunoscute. Echipa de autori nu a avut ca sarcină evaluarea sau verificarea rapoartelor studiilor de caz.

De asemenea, este posibil ca unele țări, regiuni sau autorități locale să fi adoptat legislație, norme sau ghiduri specifice pentru a încadra aplicațiile tehnologiei și aplicabilitatea acesteia.

Prezentul document NU este conceput ca un ghid sau ca un document de referință BAT pentru această tehnologie. Configurațiile pedologice, geologice și hidrogeologice ale siturilor contaminate din Europa prezintă o mare variabilitate. Prin urmare, proiectarea și punerea în aplicare adaptate la fiecare sit în parte este esențială pentru succesul remedierii siturilor contaminate. Astfel, orice recomandare raportată ar putea fi aplicată, parțial aplicată sau neaplicată. În orice caz, autorii, colaboratorii și rețelele implicate nu pot fi considerați responsabili.

Opiniile exprimate în acest document nu sunt neapărat cele ale membrilor individuali ai rețelelor subsemnate. IMPEL și rețelele sale partenere recomandă cu tărie ca persoanele/organizațiile interesate să aplice tehnologia în practică să apeleze la serviciile unor profesioniști cu experiență în domeniul mediului.

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# Definiţii

TERMEN	DEFINIȚIE	SURSĂ	PARAGRAF
'punct de conformitate'	locul (de exemplu, solul sau apele subterane) în care se măsoară criteriile de evaluare și nu trebuie să fie depășite	ISO EN 11074	3.4.5
'controlul investigație sau program de inspecție, testare sau conformității sau al performanței' strategie de remediere a fost implementată în mod corespunzător (de exemplu, toți contaminanții au fost eliminați) și/sau, atunci când a fost adoptată o abordare de izolare, că aceasta continuă să funcționeze la nivelul specificat		ISO EN 11074	6.1.5
'contaminant' <sup>1</sup>	substanța (substanțele) sau agentul (agenții) prezent(e) în sol ca urmare a activității umane	ISO EN 11074	3.4.6
'situl contaminat' <sup>2</sup>	situl în care este prezentă contaminarea	ISO EN 11074	2.3.5
'contaminarea' substanța (substanțele) sau agentul (agenții) prezent(e) în sol ca urmare a activității umane		ISO EN 11074	2.3.6
'eficientă' <sup>3</sup>	'eficientă' <sup>3</sup> <metodă de="" remediere=""> măsoară a capacității unei metode de remediere de a atinge o performanță necesară</metodă>		6.1.6
'emisie'	eliberarea directă sau indirectă de substanțe, vibrații, căldură sau zgomot din surse individuale sau difuze din cadrul instalației în aer, apă sau sol	IED	Art. 3 (4)
'standardul de setul de cerințe care trebuie să fie îndeplinite la un moment dat de un anumit mediu sau de o anumită parte a acestuia, astfel cum sunt stabilite în dreptul Uniunii		IED	Art. 3 (6)
'Coeficientul lui Henry'	coeficientul de partiție între aerul din sol și apa din sol	ISO EN 11074	3.3.12
'metoda de tratare in situ' <sup>4</sup>	· · · · · · · · · · · · · · · · · · ·		6.2.3
'levigare'	dizolvarea și deplasarea substanțelor dizolvate de către apă	ISO EN 11074	3.3.15

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 $<sup>^{1}</sup>_{\text{.}}$  În această definiție nu se presupune că din prezența contaminării rezultă prejudicii.

în această definiție nu se presupune că din prezența contaminării rezultă prejudicii.

<sup>3</sup> În cazul unei metode bazate pe proces, eficacitatea poate fi exprimată în termeni de concentrații reziduale de contaminanți obținute.

<sup>4</sup> Notă: ISO CD 241212 propune ca sinonim: "tehnică (de remediere) in situ" [Notă 1: O astfel de instalație de remediere este amplasată la fața locului, iar acțiunea de tratare a contaminantului este menită să fie aplicată direct pe subsol...] ISO CD 24212 3.1

'poluant'	substanță(e) sau agent prezent în sol (sau în apele subterane) care, datorită proprietăților, cantității sau concentrației sale, provoacă efecte negative asupra funcțiilor solului	ISO EN 11074	3.4.18
'poluare'	introducerea directă sau indirectă, ca urmare a activității umane, de substanțe, vibrații, căldură sau zgomot în aer, apă sau sol care pot fi dăunătoare pentru sănătatea umană sau calitatea mediului, pot provoca daune bunurilor materiale sau pot afecta sau interfera cu facilitățile și alte utilizări legitime ale mediului	IED	Art. 3 (2)
'obiectiv de remediere'	termen generic pentru orice obiectiv, inclusiv cele legate de cerințele tehnice (de exemplu, concentrații reziduale de contaminant, performanțe inginerești), administrative și juridice	ISO EN 11074	6.1.19
'strategia de remediere'	combinație de metode de remediere și de lucrări asociate care vor îndeplini obiectivele specifice legate de contaminare (de exemplu, concentrațiile reziduale de contaminanți) și alte obiective (de exemplu, legate de inginerie) și vor depăși constrângerile specifice ale amplasamentului	ISO EN 11074	6.1.20
'valoarea țintă de remediere'	indicarea performanței care trebuie obținută prin remediere, de obicei definită ca obiectiv legat de contaminare în termeni de concentrație reziduală	ISO EN 11074	6.1.21
'zonă saturată'	zonă a solului în care spațiul poros este complet umplut cu lichid la momentul considerat	ISO EN 11074	3.2.6
'sol'	stratul superior al scoarței terestre situat între roca de bază și suprafață. Solul este compus din particule minerale, materie organică, apă, aer și organisme vii;	IED	Art. 3 (21)
'gaze din sol'  'zonă nesaturată'	gaz și vapori în spațiile poroase ale solului zonă a solului în care spațiul poros nu este complet umplut cu lichid la momentul considerat.	ISO EN 11074 ISO EN 11074	2.1.13 3.2.8
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### 1 INTRODUCERE

Oxidarea chimică *in-situ* (ISCO) este o tehnologie de remediere aplicată frecvent în remedierea siturilor contaminate datorită gamei largi de poluanți care pot fi tratați. Aceasta constă în injectarea de oxidanți chimici, cum ar fi permanganatul, persulfatul și peroxidul de hidrogen în sol, pentru a transforma poluanții în compuși inofensivi, prin reacții de oxidare.

ISCO poate trata cu succes poluanți precum solvenții clorurați, TPH (hidrocarburi totale din petrol), BTEX (benzen, toluen, etilbenzen și xilen), MTBE (metil terț-butil eter), fenoli, PHA (hidrocarburi aromatice policiclice) și clorobenzeni.

Știm deja că oxidarea are loc între acești poluanți și oxidanți, dar trebuie ajustați o mulțime de parametri. Alegerea acestei tehnologii de remediere necesită cunoștințe specifice sitului cu privire la poluanți, la modul în care aceștia sunt distribuiți în sol și în apele subterane, la situația geologică și hidrogeologică a sitului. Fiecare sit are propriul ISCO "adaptat". Nu este neobișnuit să vedem că alegerea unei tehnologii se face după caracterizarea preliminară, fără a avea informații detaliate, în ideea de a economisi timp și de a începe rapid remedierea. Experiența din ultimele decenii în remedierea siturilor ne-a arătat că este necesară prezentarea caracteristicilor proiectului de remediere (CPR) pentru a decide tehnologia potrivită pentru fiecare situație și nu trebuie făcută nicio generalizare cu privire la distribuția contaminanților sau la geologia subsolului. În figura 1.1. sunt prezentate costurile conceptuale ale ciclului de viață al proiectului cu și fără CRP.

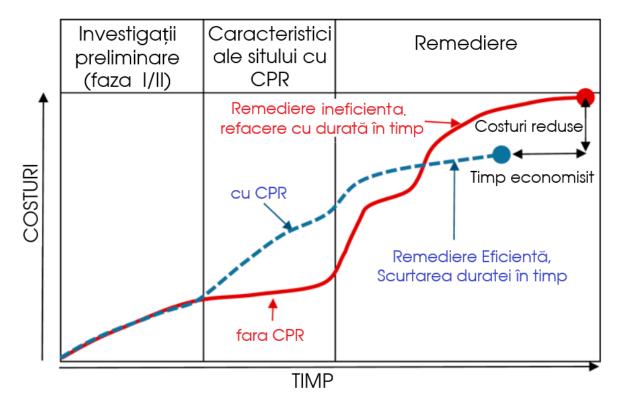


Figura 1.1- Costurile conceptuale ale ciclului de viață al proiectului cu și fără CPR (caracteristicile proiectului de remediere)

În schema de mai sus se arată efectul pozitiv al CPR asupra reducerii timpului și limitării costurilor întregii remedieri, chiar dacă există o creștere considerabilă a costului inițial datorată identificării caracteristicilor proiectului de remediere.

Așadar, principala provocare constă în colectarea tuturor informațiilor utile pentru a facilita procesele de oxidare în situl respectiv. După cum se poate observa în schema din figura 1.2, împărțirea traseului în etape succesive poate fi foarte utilă.

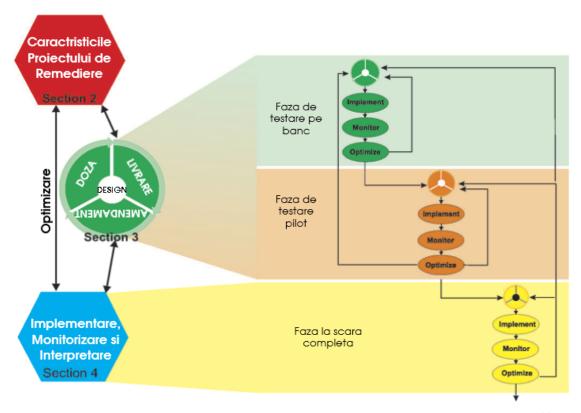


Figura 1.2- Schema de la ITRC - Consiliul Interstatal pentru Tehnologie și Reglementare (https://ois-isrp-1.itrcweb.org/)

De asemenea, ISCO poate fi utilizat în combinație cu alte tehnologii cu diferite niveluri de intensitate, astfel fiind preferabil să fie planificate mai multe scenarii cu performanțe diferite în ceea ce privește componentele de mediu, sociale și economice ale sustenabilității (Figura 1.3). Alternativele de proiectare sunt planificate combinând tehnici de remediere care pot fi aplicate cu o logică spațială (tehnici diferite pe diferite porțiuni ale sitului) sau temporală (succesiune de tehnologii în aceeași zonă), după cum se poate observa în figura 1.4. Intensitatea unui scenariu de tratare variază în funcție de diferitele combinații de abordări de efort de remediere activă și de efort de remediere pasivă. Eforturile de remediere activă se bazează pe utilizarea de energie ridicată și de reactivi chimici, în timp ce eforturile de remediere pasivă implică mecanisme biologice.

De obicei, această abordare integrată generează efecte sinergice asupra întregului proiect de remediere.

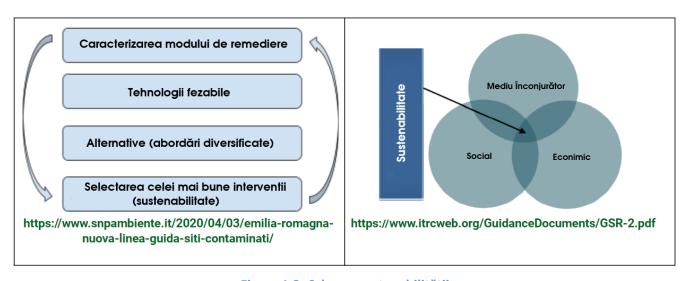


Figura 1.3- Schema sustenabilității

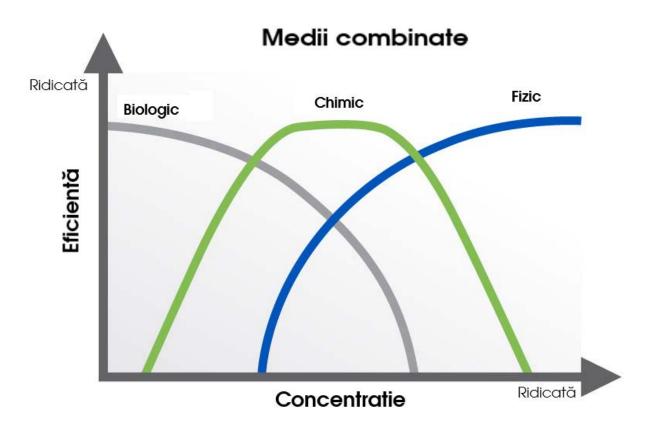


Figura 1.4- Schemă din Tratamentul integrat/Prezentarea generală a remedierii combinate (©Regenesis 2016)

Următoarele capitole vor descrie tehnica și cei mai importanți pași care trebuie parcurși pentru a atinge obiectivele activității de remediere. Informațiile conținute sunt rezultatul unor ani de observații experimentale și de aplicare în teren a cunoștințelor teoretice.

### 2 DESCRIEREA TEHNICII

Tehnicile de remediere *in situ* a solului se ocupă de situațiile de contaminare a solului și a apelor subterane fără a fi necesară excavarea solului sau extragerea apelor subterane. Deoarece nu este necesară nicio excavare, aceste tehnici au un impact mai mic asupra utilizării terenului și pot fi aplicate în diverse locații. De asemenea, compoziția și structura solului sunt mai puțin afectate.

Tehnica ISCO utilizează substanțe chimice numite oxidanți (de exemplu, permanganat, persulfat, peroxid de hidrogen, ozon) pentru a ajuta la transformarea poluanților dăunători în produse secundare mai puțin toxice. Se numește *in situ* deoarece se realizează la fața locului, fără a fi nevoie să se excaveze solul sau să se pompeze apa subterană pentru decontaminare.

Pentru a aplica tehnica ISCO, se injectează în subteran un agent oxidant, care trece prin masa acestuia și provoacă distrugerea chimică (oxidarea) a poluanților, transformându-i în compuși mai mici și mai puțin toxici. Oxidanții sunt aplicați prin metoda selectată (pentru descrierea principalelor metode de aplicare, a se vedea secțiunea 4.1.6). Astfel, amendamentul adăugat ajunge la zona contaminată. În acest caz, accentul se pune atât pe poluanții dizolvați, cât și pe cei nedizolvați.

Odată aplicat, oxidantul este difuzat în acvifer, unde se amestecă și reacționează cu poluanții. În acest scop, secțiunea de filtrare a sondelor sau supapele trebuie să fie de așa natură încât să asigure un tratament eficient al contaminării, pentru a acționa cât mai mult posibil asupra contaminării. În acest caz, accentul este pus atât pe poluanții dizolvați, cât și pe cei nedizolvați. Odată ce oxidantul este pompat în sonde, acesta se difuzează în solul saturat și în apele subterane din proximitate, unde se amestecă și reacționează cu poluanții.

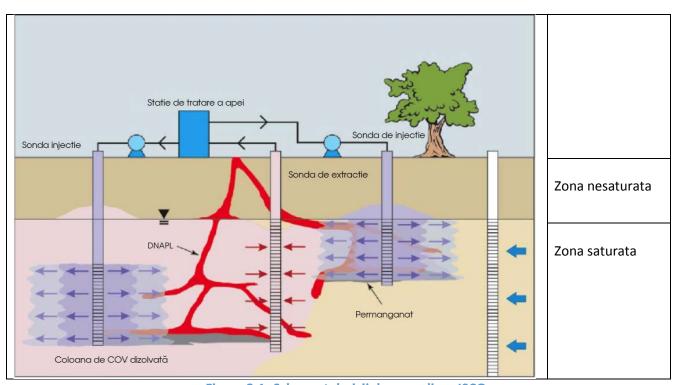
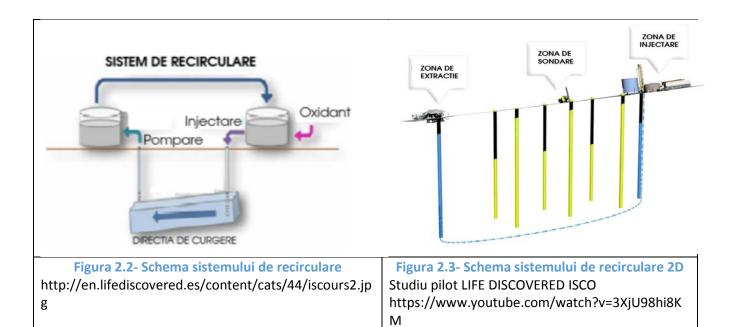


Figura 2.1- Schema tehnicii de remediere ISCO

COV – compus organic volatil, LDFN – lichid dens în fază neapoasă (DNAPL)

Caracteristicile principale ale tehnicii sunt:

- Sunt reduse semnificativ concentrațiile poluanților, ținând cont de obiectivele fixate în strategia de remediere.
- Un produs (agent oxidant) este introdus în sol, distribuit în întreaga masă a solului şi inițiază distrugerea chimică (oxidarea) poluanților în specii mai puțin toxice.
- Structura solului rămâne intactă.



### 2.1 Fazele ISCO

Comportamentul unui poluant în sol și eficiența unei tehnologii de remediere sunt determinate de mai mulți factori care interacționează într-un mod complex și care depind de caracteristicile poluantului în sine și de cele ale solului. Pentru a selecta o tehnologie cu perspectivă bună de succes, este esențial să se țină seama de caracteristicile poluantului și ale sitului poluat.

Pentru punerea în aplicare in situ a tehnicii ISCO, există următoarele opțiuni pentru FAZE, după cum urmează:

- 1. SELECTAREA ZONEI DE APLICARE ȘI A INFRASTRUCTURII DE BAZĂ: Succesul tehnicii depinde de amplasarea optimă a sondelor. De asemenea, în cazul în care nu se cunoaște nicio locație optimă, forajele și sondele de injectare, de extracție și de monitorizare a testului trebuie să fie efectuate în zona pilot selectată.
- 2. 2. INJECTĂRI: După forare, se injectează în sondă o soluție cu un agent oxidant. Soluția respectivă rupe legăturile C-C ale poluanților. Oxidarea chimică a poluanților îi transformă în compuși mai puțin periculoși și mai ușor de tratat.
- 3. RECIRCULAREA: Oxidarea poluanților depinde de timpul de rezidență a oxidantului în sol. Atunci când timpul de contact (oxidant roca de bază) este considerat suficient, soluția este pompată printr-o sondă și injectată din nou, dacă este necesar. Procesul de recirculare se va desfășura până când capacitatea de oxidare a agentului scade (figura 2.4).

- 4. EXTRACȚIA: Remedierea va fi oprită atunci când oxidantul nu mai este eficient și când concentrația de poluant prezintă o tendință de scădere. Apoi, soluția este pompată și tratată într-o stație de tratare adecvată a apei.
- 5. MONITORIZAREA: Pentru a evalua progresul de remediere a tehnicii ISCO (condiții inițiale, medii și finale) și performanța globală a testului, este esențial să se monitorizeze parametri precum potențialul de oxidoreducere, conductivitatea, temperatura, oxidanții și subprodușii de reacție, precum și concentrația poluanților vizați.

Aceste etape pot fi sau nu efectuate în mod secvențial.

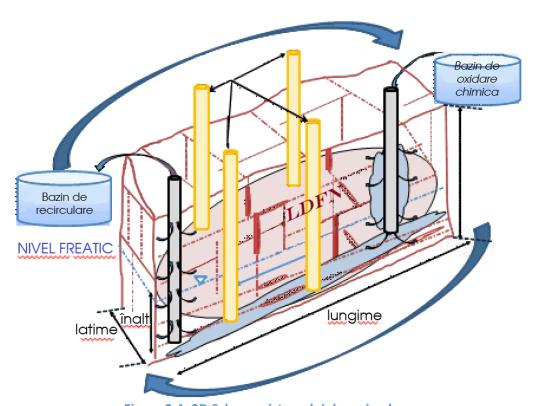


Figura 2.4- 3D Schema sistemului de recirculare



Figura 2.5- Amestecarea agenților înainte de injectare

### 2.2 Caracteristicile DNAPL (lichidului dens în fază neapoasă)

Acronimul DNAPL provine de la *dense liquid in non-aqueous phase*. DNAPL este un lichid mai dens decât apa și este nemiscibil în apă sau nu se dizolvă în apă. Termenul este folosit de ingineri, ecologiști și hidrogeologi pentru a descrie un grup de poluanți prezenți în apele de suprafață, apele subterane sau sol.

Termenul DNAPL include multe substanțe chimice. Unele dintre cele mai importante sunt solvenții organoclorurați, creozotul, reziduurile de gudron de cărbune și pesticidele. Cei mai des întâlniți DNAPL în siturile contaminate sunt solvenții organoclorurați.

În funcție de proprietățile fizice și chimice ale DNAPL, acestea sunt evacuate în cantități semnificative în sol. Ca urmare, solul devine poluat. În general, DNAPL va percola solul până când se va acumula în cele din urmă deasupra unor straturi cu o impermeabilitate ridicată. Capacitatea mare de penetrare și complexitatea mediului natural (eterogenitate) fac ca acest tip de contaminare să fie dificil de localizat. În consecință, este dificil de remediat și reabilitat subsolul.



Figura 2.6- DNAPL

Poluanți: hexaclorbenzen (HCB), Alpha hexaclorciclohexan ( $\alpha$ -HCH), beta hexaclorciclohexan ( $\beta$ -HCH), lindan și pentaclorobenzen https://www.youtube.com/watch?v=3XjU98hi8KM



Figura 2.7- Amestec nemiscibil între DNAPL și proba de apă http://en.lifediscovered.es

Riscurile asociate prezenței acestui tip de poluanți în subsol sunt ridicate.

Consecințele sunt ușor observabile pe termen mediu și lung, în special deoarece:

- Toxicitatea poluanților în DNAPL este mare
- Solubilitatea individuală a poluanților este mică, dar de cele mai multe ori suficientă pentru a depăși pragurile limită în apa potabilă
- Au un potențial de migrare ridicat, atât prin sol, cât și prin apele subterane.

Infiltrarea acestor DNAPL prin sol depinde de natura deversării, de caracteristicile lichidului, cum ar fi densitatea, tensiunea superficială, vâscozitatea și porozitatea. De asemenea, forțele hidraulice au un impact asupra infiltrării. Migrarea DNAPL are loc în mod preferențial prin căile mai permeabile, cum ar fi fracturile dintr-un mediu de rocă consolidată sau de argilă sau straturile foarte permeabile.

Detectarea DNAPL în probele de sol și de apă subterană este dificilă, din cauza culorii (uneori DNAPL este transparent), a concentrațiilor scăzute sau a aspectului eterogen în subsol. Toți acești factori complică caracterizarea sursei de contaminare, care este, de obicei, agravată de prezența amestecurilor de compuși.

DNAPL sunt clasificate în patru mari grupe:

- compuși organici halogenați.
- gudron şi creozot
- bifenili policlorurați (PCB)
- amestecuri şi pesticide.

Majoritatea siturilor afectate de DNAPL conțin compuși organici halogenați, în principal organoclorurați. Utilizarea lor pe scară largă, proprietățile chimice și toxicitatea ridicată sunt principalii factori care accentuează problema.

Cele mai caracteristice proprietăți chimice ale DNAPL sunt:

- densitate ridicată;
- vâscozitate scăzută;

- volatilizare ridicată;
- solubilitate semnificativă în raport cu toxicitatea.

### 2.2.1 Volatilitatea

De asemenea, DNAPL pot fi clasificate în funcție de volatilitate. Compușii organici volatili sunt notați utilizând acronimul COV.

Aceștia sunt compuși organici care au constante Henry și presiuni de vapori ridicate, solubilitate moderată și greutate moleculară mică.

Volatilitatea unui compus este, în general, mai mică la o temperatură de fierbere mai mare (Tf), o constantă Henry mai mare (KH) și o presiune de vapori mai mare (Pvap). Prin urmare, COV au o compoziție chimică favorabilă evaporării în condiții normale de mediu în ceea ce privește temperatura și presiunea. În general, acești compuși au o constantă Henry mai mare de  $10^{-5}$ atm m³/mol și presiuni de vapori mai mari de 1 mm Hg (0,0013 atm).

În ceea ce privește volatilitatea, compușii organici pot fi clasificați după cum urmează:

- volatili (COV);
- semi-volatili (SCOV);
- puţin volatili.

În general, compușii organici halogenați sunt volatili sau semivolatili, PCB-urile și pesticidele sunt semivolatile, iar uleiurile lubrefiante sunt puțin volatile.

Compuși organici	Temperatura de fierbere	Exemplu
volatili (COV)	Tf< 250°C	compușii organici halogenați , PCE și TCE
semi-volatili (SCOV)	250 C <tb<390 c<="" td=""><td>PCB-uri și pesticide, pesticide organoclorurate și alți</td></tb<390>	PCB-uri și pesticide, pesticide organoclorurate și alți
		compuși halogenați.
puţini volatili	Tb> 390°C	Uleiuri lubrefiante

Tabelul 2.1- Volatilitatea principalelor clase de poluanți

### 2.3 Oxidarea poluanților

Oxidarea chimică *in situ* (ISCO) se bazează pe o reacție red-ox în sol între oxidantul injectat și contaminanții prezenți. Oxidantul și orice substanțe auxiliare necesare sunt injectate în sol, unde reacționează cu contaminanții prezenți. Ca urmare, oxidantul este redus, iar contaminanții sunt oxidați și descompuși în specii inofensive care pot fi prezente în mod natural în sol. Această tehnică de remediere este adecvată numai pentru remedierea contaminării organice.

### 2.3.1 Agentul oxidant

Există mai multe forme diferite de oxidanți care au fost utilizate pentru ISCO, însă, cei patru agenți oxidanți cel mai frecvent utilizați sunt următorii:

- Permanganat de potasiu (e.g., KMnO<sub>4</sub>);
- Peroxid de hidrogen/apă oxigenată (H<sub>2</sub>O<sub>2</sub>) şi fier (Fe) (oxidarea Fenton sau oxidarea derivată H<sub>2</sub>O<sub>2</sub>);

- Ozon (O<sub>3</sub>);
- Persulfat (Ex: K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> sau Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub>).

Oxidant	Specii reactive	Formă	Persistență <sup>(1)</sup>	Stagiu de dezvoltare
Permanganat	MnO <sub>4</sub> -	pulbere/lichid	>3 luni	în dezvoltare
Fenton	OH, O <sub>2</sub> -, HO <sub>2</sub> , HO <sub>2</sub> -	lichid	minute - ore	experimental/de urgență
Ozon	O <sub>3</sub> , OH	gaz	minute - ore	experimental/de urgență
Persulfat	'SO <sub>4</sub> <sup>2-</sup>	pulbere/lichid	ore- săptămâni	experimental/de urgență

Oxidanți și reacții Potențialul la electrod (E <sub>h</sub> ) <sup>(2)</sup>					
Permanganat					
$MnO_4^- + 4H^+ + 3e^- \rightarrow MnO_2 + 2H_2O$	1,7 V (ion permanganat)	(1)			
Fenton (reactanți derivați ai H <sub>2</sub> O <sub>2</sub> )					
$H_2O_2 + 2H^+ + 2e^- \rightarrow 2H_2O$	1,8 V (peroxid de hidrogen)	(2)			
2 OH +2H $^{+}$ +2e $\rightarrow$ 2H <sub>2</sub> O	2,8 V (radical hidroxil)	(3)			
$HO_2 + 2H^+ + 2e^- \rightarrow 2H_2O$	1,7 V (radical perhidroxil)	(4)			
$O_2$ +4H <sup>+</sup> + 3e $\rightarrow$ 2H <sub>2</sub> O	-2,4 V (radical superoxid)	(5)			
$HO_2 + H_2O + 2e \rightarrow 3OH$	-0,88 V (anion hidroperoxid)	(6)			
Ozon					
$O_3 + 2H^+ + 2e^- \rightarrow O_2 + H_2O$	2,1 V (ozon)	(7)			
$2O_3 + 3H_2O_2 \rightarrow 4O_2 + 2.0H + 2H_2O$	2,8 V (radical hidroxil)	(8)			
Persulfat					
$S_2O_8^{2-} + 2e^- \rightarrow 2SO_4^{2-}$	2,1 V (persulfat)	(9)			
$SO_4 + e^- \rightarrow SO_4^{2-}$	2,6 V (radical sulfat)	(10)			
(1) Persistența oxidantului variază în funcție de d	condițiile specifice sitului. Rezultatele se bazează pe observații generale				
(2) În paranteză sunt trecute speciile reactive. Po	otențialul de reducere este negativ				

Tabel 2.2- Forma oxidantă, Stabilitatea Etapa de dezvoltare şi potențialul de oxidare pentru agenții folosiți la aplicarea tehnicii ISCO

### 2.3.1.1 Permanganatul de potasiu (KMnO<sub>4</sub>)

Permanganatul persistă perioade îndelungate de timp, fiind posibilă și difuzia în materiale cu permeabilitate scăzută și transportul pe distanțe prin medii poroase.

Reacția directă este reacția de înjumătățire cu 3 electroni pentru oxidarea permanganatului (MnO4-) în majoritatea condițiilor de mediu (pH 3,5 până la 12). Unul dintre subprodușii de reacție este  $MnO_2$ , în intervalul de pH de la 3,5 la 12 acesta fiind un precipitat solid.

 $MnO_4^- + 4 H^+ + 3e^- \rightarrow MnO_2 + 2 H_2O$ 

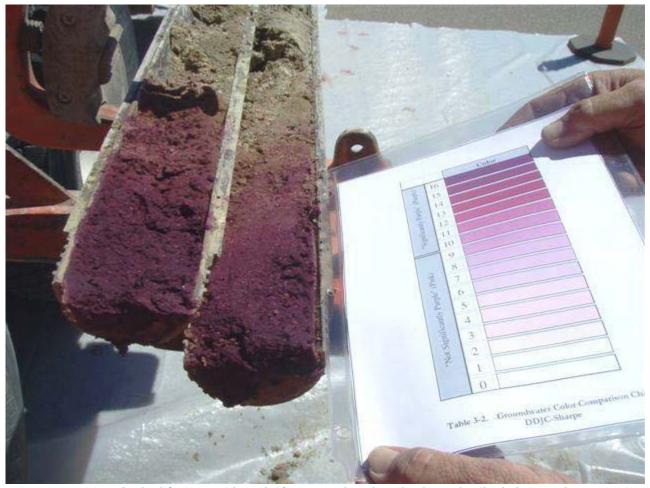


Figura 2.8- Exemplu de difuzie a oxidantului în carote de sol argilos la 90 de zile de la introducerea prin fracturare a suspensiei oxidante de permanganat de potasiu (fotografie obținută cu amabilitatea URS, arhiva Bures)

În condiții acide (pH <3,5), în soluție sau în formă coloidală, Mn poate fi prezent în diferite stări oxidative dependente de condițiile red-ox (Mn $^{+2,+4,+7}$ ).

$$MnO_4^- + 8 H^+ + 5e^- \rightarrow Mn^{+2} + 4 H_2O$$

În plus, în condiții puternic alcaline, pH>12, Mn poate fi prezent sub formă de  $Mn^{+6}$ .  $MnO_4^{-1} + e^- \rightarrow MnO_4^{-2}$ 

Reacțiile de oxidare chimică a poluanților: percloroetilenă (PCE), tricloretilenă (TCE), dicloroetilenă (DCE) și, respectiv, clorură de vinil (VC):

- Percloroetilenă (PCE)  $4KMnO_4 + 3C_2CI_4 + 8 H_2O \rightarrow 6 CO_2 + 4MnO_2 + 4KOH + 12HCI$
- Tricloretilenă (TCE)
   2 KMnO<sub>4</sub> + C<sub>2</sub>HCl<sub>3</sub> → 2 CO<sub>2</sub> + 2 MnO<sub>2</sub> + 2 KCl + HCl

- Dicloroetilenă (DCE)
   8 KMnO<sub>4</sub> + 3C<sub>2</sub>H<sub>2</sub>Cl<sub>2</sub>→6 CO<sub>2</sub> + 8 MnO<sub>2</sub> + 2 KOH + 6 KCl + 2H<sub>2</sub>O
- Clorură de vinil (VC)  $10 \text{ KMnO}_4 + 3C_2H_3Cl \rightarrow 6 \text{ CO}_2 + 10 \text{ MnO}_2 + 7 \text{ KOH} + 6 \text{ KCl} + H_2O$

Dioxidul de carbon ( $CO_2$ ) este un produs secundar rezultat din oxidarea și mineralizarea substanțelor chimice organice și a materiei organice naturale. În studiile pe coloană, reducerea permeabilității și eficiența de spălare au scăzut ca urmare a precipitării  $MnO_2$  (s) și formării de  $CO_2$  (g).

### 2.3.1.2 Peroxidul de hidrogen/ Apa oxigenată (H<sub>2</sub>O<sub>2</sub>)

Reacția Fenton clasică implică în mod specific reacția dintre  $H_2O_2$  și fierul feros (Fe(II)) care produce radicalul hidroxil (OH) și ionii de fier (Fe(III)) și hidroxil (OH-).  $H_2O_2 + Fe(II) \rightarrow Fe(III) + OH + OH$ 

Fe(III) reacționează cu  $H_2O_2$  sau cu radicalul superoxid  $(O_2^-)$   $H_2O_2 + Fe (III) \rightarrow Fe (II) + O_2^- + 2 H^+$ 

Fe(III) reactionează cu radicalul superoxid ( $O_2$ )  $O_2$  + Fe (III)  $\rightarrow$  Fe (II) +  $O_2$  (g) + 2 H<sup>+</sup>

Această secvență generală de reacții continuă să aibă loc până când  $H_2O_2$  este complet consumat. Deoarece  $H_2O_2$  injectat în subsol reacționează cu multe specii chimice, altele decât Fe(II), această tehnologie este adesea denumită peroxid de hidrogen catalizat (CHP).

S-a raportat că  $H_2O_2$  persistă în sol și în acvifer timp de câteva minute până la câteva ore, iar distanțele de transport difuziv și advectiv vor fi relativ limitate. Intermediarii radicali formați cu ajutorul unor oxidanți ( $H_2O_2$ ,  $S_2O_8$  2-,  $O_3$ ), care sunt în mare parte responsabili de diverse transformări ale contaminanților, reacționează foarte rapid și persistă pentru perioade foarte scurte de timp (<1 sec).

### 2.3.1.3 Ozonul (O<sub>3</sub>)

Oxidarea *in situ* cu O<sub>3</sub> presupune injectarea unui amestec de aer și O<sub>3</sub> gazos direct în zonele nesaturate și/sau saturate. Pulverizarea aerului este o tehnologie care a fost investigată în mod riguros și care prezintă multe asemănări cu pulverizarea O<sub>3</sub> și oferă o perspectivă asupra transportului de masă și a mecanismelor de transfer de masă în cazul pulverizării *in situ* a O<sub>3</sub>, care nu a fost investigată în mod riguros în sistemele subterane. Injectarea de aer sub pânza freatică favorizează volatilizarea, furnizează oxigen pentru degradarea aerobă și poate induce amestecarea apei subterane (Johnson, 1998).

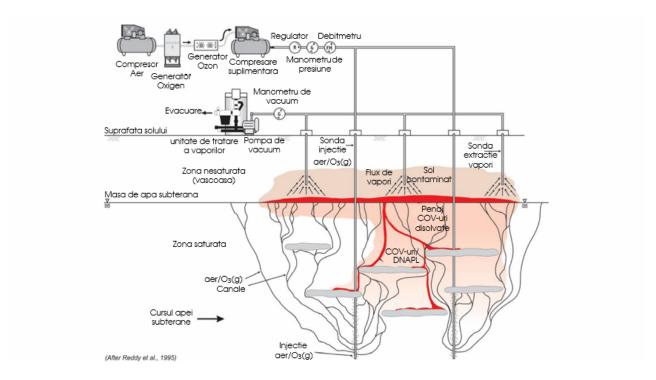


Figura 2.9- Modelul conceptual general al ozonizării in-situ în zona saturată cu extragerea cu vacuum pentru a captura emisiile volatile și ozonul.

### 2.3.1.4 Persulfat de sodiu sau de calciu

Persulfatul este cel mai puternic oxidant din familia peroxigenului, cu un potențial de oxidare de 2,12 volți. După cum este ilustrat mai jos, reacția de oxidare directă a semicelulei pentru persulfat implică un transfer de doi electroni:

$$2 S_2 O_8^{2-} + 2 H^+ + 2e^- \rightarrow 2HSO_4^{-}$$

Cu toate acestea, în majoritatea cazurilor, distrugerea rapidă a poluantului necesită activarea persulfatului pentru a genera radicali sulfat. Radicalii de sulfat sunt agenți oxidanți puternici, cu un potențial de oxidare de 2,6V.

- Persulfatul de sodiu:
  - o Activat în condiții alcaline
  - Activat cu peroxid de hidrogen

Persulfatul activat este catalizat cu peroxidul și baza furnizată de peroxidul de calciu:

 $S_2O_8^{2}$  + activator de peroxid de calciu  $\rightarrow 2SO_4$ •

Persulfatul activat poate rămâne disponibil în subsol timp de luni de zile, oferind o combinație de putere și stabilitate.

Adăugarea de peroxid de calciu oferă mai multe beneficii. În primul rând, acesta conferă alcalinitatea și peroxidul necesare pentru a activa persulfatul folosind chimia de activare. În al doilea rând, atunci când este amestecat cu apă, oferă o sursă de peroxid de hidrogen și hidroxid de calciu cu eliberare lentă pe termen lung. Peroxidul de hidrogen care se formează lent se descompune în oxigen și apă, oferind o sursă de oxigen prelungită pentru bioremedierea ulterioară a hidrocarburilor petroliere. Abordarea utilizată pentru activarea radicalului sulfat a fost utilizarea unui pH ridicat, cu ajutorul peroxidului de calciu.

### 2.4 Tehnica ISCO aplicată în diverse contexte

ISCO ID Țara, organizația și locația	Oxidant	Poluanţi	Area m²	Observations
Israel. Ludan environmental technologies	KMnO <sub>4</sub>	Solvenți clorurați, în special tricloretilenă (TCE). Altele: mangan, crom	300	
<b>Germania</b> . RiskCom GmbH	KMnO <sub>4</sub> Na <sub>2</sub> S <sub>2</sub> O <sub>8</sub> Additivi: Gumă de Guar	PCE/TCE de până la 200.000 μg/L  concentrații de CVOC în sol de > 6 000 mg/kg Concentrații de până la 447.000 μg/L de CVOC total în probele de apă subterană	1000 (estimat)	ISCO care utilizează fracturarea hidraulică (injectată sub presiune) ca metodă preferată de amplasare.
Germania. SENSATEC GMBH. Amplasament în apropiere de Frankfurt am Main, Germania, pe terenul unei foste fabrici de produse chimice care producea solvenți pentru metalurgie, produse chimice de curățare și uleiuri speciale.	Persulfat de potasiu activat prin activare alcalină cu adaos de peroxid de calciu, Modificator de vâscozitate a polimerilor organici	TPH și BTEX în zona nesaturată, cu concentrații de contaminanți de până la 5.000 mg/kg și 344 mg/kg, respectiv 344 mg/kg.  Apa subterană (CHCs) de până la 44.300 µg/L, urmată de TPH (2.000 µg/L) și BTEX (1.800 µg/L).	620	Amplasarea oxidanţilor prin fracturarea solului TSE
Austria. Keller Grundbau Ges.mbH. Sit-ul este situat în centrul orașului Graz, Styria	KMnO <sub>4</sub>	Tetracloroetilena a fost folosită în spălarea chimică de la fața locului. Cele mai mari concentrații de 14000 mg/m³ au fost găsite sub locul de instalare a mașinilor de spălat.	300 (estimat)	
<b>Olanda.</b> Heijmans Infra BV Sit de lângă centrul orașului Uden, Netherlands.	Persulfat de sodiu (Klozur R One). S-a presupus că necesarul de oxidanți este de 3,0 g de	Hidrocarburi clorurate Solvenți clorurați, în special tricloretilenă (TRI) > 16.000 µg/l în zona saturată. În zona nesaturată a fost prezent mai mult de 16.000 mg/kg TRI.	270	

	persulfat/kg			
	de sol.			
Italia. REGENESIS. Regiunea Veneto, Italia Un camion-cisternă s-a răsturnat pe un drum mic din nordul Italiei deversând peste 36.000 de litri de motorină și benzină. Combustibilul a avut impact asupra unui canal, a digurilor de apărare împotriva inundațiilor, a solului și a apelor subterane din imediata vecinătate.	Percarbonat de sodiu și lichid/gel compus în principal din silicat de fier	Sol afectat de TPH și BTEX Ape subterane afectate de MTBE și TPH	Approx. 500	
Italia. ARPA Campania. Compania activează și produce pentru sectoarele apărării, aerospațial și de securitate. În apropierea zonei de confirmare Lago Fusarohttps://www.leonard ocompany.com/	Permanganat de sodiu Soluție de permanganat de sodiu cu o concentrație de 40%.	Soluri: Hidrocarburi: 3500 mg/Kg  Apă subterană: Benzo(a)antracenă: 7,6 µg/L Piren: 29 µg/L Benzo(b)fluorantena: 4,2 µg/L Benzo(g,h,i)perilen: 2,2 µg/L Hidrocarburi aromatice policiclice (sumă): 10 µg/L Tetracloretilenă: 50 µg/L Tricloretilenă: 5,4 µg/L Clorură de vinil: 4,1 µg /L Benzen: 27 µg/L Xilen: 133 µg/L Toluen: 22 µg/L	300 (calculat)	PZSB4 10 3 10 3 10 3 10 3 10 3 10 3 10 3 10
Italia. Golder Associates S.r.l. Stație de distribuție de carburanți, cu depozitarea carburanților în rezervoare subterane, situată în centrul Italiei.	Persulfat de sodiu (Na <sub>2</sub> S <sub>2</sub> O <sub>8</sub> ), activat prin adăugarea de hidroxid de sodiu (NaOH) Peroxid de calciu (CaO <sub>2</sub> ), pentru a spori	Sol nesaturat de adâncime poluat cu benzen 163 mg/kg SS etilbenzen 502 mg/kg SS toluen 648 mg/kg SS xileni 1.472 mg/kg SS hidrocarburi ușoare C≤12 19,509 mg/kg SS hidrocarburi grele C>12 5,742 mg/kg SS MtBE 736 mg/kg SS	800 (calculat)	Same of the control o

Italia. Stantec Până în 2015 a fost un loc de vânzare cu amănuntul a	bioremedier ea.  Persulfat și peroxid de calciu	- Apă subterană, cu benzen 46 μg/l toluen 3.800 μg/l p-xilen 2,619 μg/l hidrocarburi totale (ca n- hexan) 13.000 μg/l MtBE 230 μg/l Contaminarea cu MTBE a fost detectată înainte de demolarea instalației.	1500	
carburanţilor, iar din 2015 este o zonă de parcare. S-a emis ipoteza unei scurgeri de ulei din rezervoare şi/sau din conducte în timpul activităţilor de vânzare.	carora			
Franţa. ARTELIA Fostă benzinărie care a fost dezafectată și se află în proces de încetare a activităţii. Impactul asupra solului și apelor subterane ca urmare a unui incident - scurgere de hidrocarburi.	Permanganat de sodiu 20%.	Concentrațiile din sol: TPH C5-C10: 250 până la 1 500 mg/kg BTEX: 80 până la 820 mg/kg  Concentrații maxime în apele subterane: TPH C5-C10: 52 000 până la 48 500 ug/l BTEX: 43 000 până la 96 980 ug/l		
Italia. Arcadis Italia s.r.l. Stație de carburanți cesionată, situată într-o zonă de câmpie din nordul Italiei. Activitatea amplasamentului consta în distribuția de produse petroliere pentru transport, cu depozitarea temporară a substanțelor în rezervoare subterane.	Persulfat (soluție de apă 20%) și un activator (peroxid de calciu) care crește pH-ul.	Probele de apă subterană au indicat prezența benzenului (10 μg/L), a hidrocarburilor totale (1.000 μg/L) și a EtBE (1.000 μg/L).  Prezență în solul saturat a ETBE (0,5 mg/Kg).	450	
Italia. Mares S.r.l. Situat pe malul sudic al lacului Maggiore. O stație de benzină unde s- au desfășurat activități de comercializare a produselor petroliere pentru autovehicule, realimentarea autovehiculelor, vânzarea de lubrifianți și	Complex oxidant pe bază de persulfat de sodiu activat cu peroxid de calciu.	TPH şi BTEX Probele de apă subterană au indicat prezenţa MTBE	200 (estimat)	

schimbarea uleiului autoturismelor.				
Germania. Züblin Umwelttechnik GmbH Sit industrial, a prezentat o contaminare masivă a apelor subterane din Keuper.	Soluţie de NaMnO₄ 40%.	Apele subterane au prezentat un maxim clar de COV cu concentrații între 30 și 50 mg/l.	Întreaga zonă de contamin are 20.000 m², Sursa de contamin are 5.000 m²	The state of the s

### 3 STUDIUL DE FEZABILITATE

Deoarece ISCO este o tehnologie de remediere foarte versatilă, aplicarea trebuie să fie adaptată la fiecare sit în parte. Proiectarea unei remedieri durabile înseamnă, de asemenea, că aspectele de mediu, sociale și economice trebuie combinate pentru a se ajunge la cea mai bună soluție posibilă pentru respectivul sit. Astfel, este esențial să se compare mai multe soluții fezabile și să se identifice cea mai sustenabilă dintre acestea.

Pentru a dobândi informațiile necesare, trebuie parcurse următoarele etape:

- Definirea obiectivelor ISCO in cadrul proiectului de remediere;
- Aplicabilitatea tehnicii ISCO prin:
  - Screening iniţial;
  - Screening detaliat.

#### 3.1 Definirea objectivelor

Primul pas în verificarea fezabilității tratamentului cu oxidanți chimici este definirea obiectivelor întregului proiect de remediere. Definirea obiectivelor trebuie să descrie nivelurile de concentrație care trebuie atinse și orice factor limitativ, inclusiv resursele economice și încadrarea în timp.

Obiectivul remedierii cu oxidanți poate fi definit din punctul de vedere al valorilor de screening (ţinte de remediere, de exemplu: MCL), sau un nivel de concentrație intermediar, identificat ca parte a unei abordări integrate de remediere, bazată pe diferite mecanisme de acţiune (fizice, chimice și biologice). De exemplu, pentru a maximiza eficiența unei acţiuni de remediere, ISCO poate fi aplicat după tratamentul cu agenţi chimici tensioactivi sau de desorbţie, ori poate fi utilizat ca primă etapă pentru a reduce concentraţia de contaminanţi şi pentru a-i face compatibili cu aplicarea unei bioremedieri.

Exemple de obiective pentru ISCO:

- reducerea masei de contaminant în zona de tratare (de exemplu, cu 90%);
- atingerea unui nivel de contaminare specificat (ţintă de remediere) pentru tratamentul post-ISCO;
- atingerea unui nivel de contaminare specificat (ţintă de remediere) la unul sau mai multe puncte relevante de conformitate.

### 3.2 Aplicabilitatea tehnicii ISCO

Diagrama bloc din Figura 3.1 este utilă ca screening inițial, atunci când trebuie să se adopte decizia de a face o remediere utilizând tehnica ISCO.

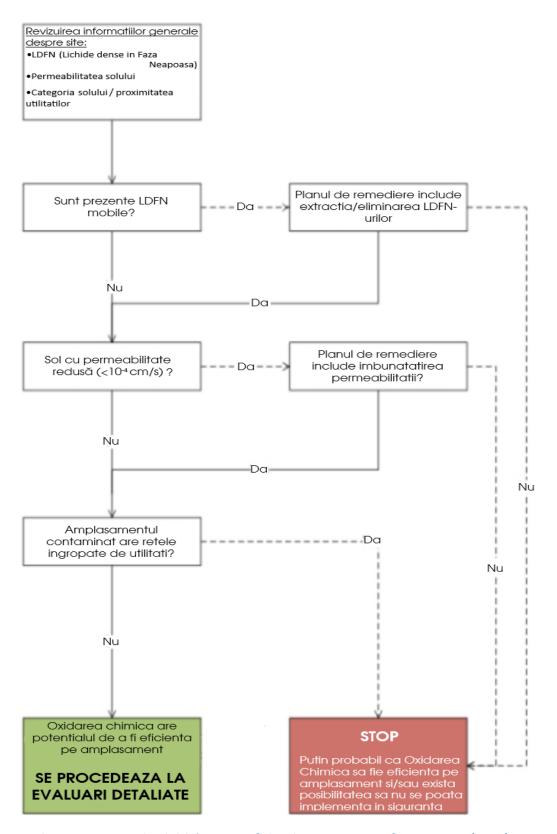


Figura 3.1- Screening inițial pentru eficientizarea ISCO, conform US EPA (2004)

Un screening inițial pentru evaluarea fezabilității tehnicii de remediere ISCO include:

- consumul de oxigen
- caracteristicile hidrogeologice și litostratigrafice;
- prezența infrastructurii subterane.

### 3.2.1 Consumul de oxigen

Prezența NAPL-urilor în faza mobilă determină o cerere excesivă de oxidant, ceea ce ar putea compromite fezabilitatea, din cauza creșterii cantității de oxidant și a numărului de injectări necesare, care conduce la rezultate nefavorabile ale unei analize a efectelor costurilor, după cum se arată în diagrama din figura 3.2.

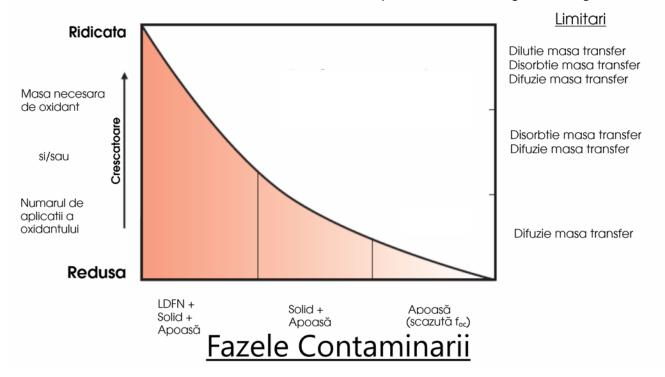


Figura 3.2- Impactul fazelor contaminanților, al transferului de masă și limitările privind masa de oxidant și/sau a numărului de aplicații de oxidant necesare pentru ISCO

Schema din tabelul 3.1 analizează situațiile posibile și descrie diferite strategii de aplicare a unui ISCO eficient: în cazul primei și celei de-a doua opțiuni din tabel (NAPL mobile; bazine continue de NAPL), trebuie să se utilizeze mai întâi o altă tehnologie.

Natura contaminantului	Aplicabilitatea ISCO	Considerații
NAPL mobile  Bazine continue de NAPL	Posibilă, dar provocatoare	Necesară doză foarte ridicată de oxidant sau co-solvent/surfactant
NAPL reziduale  Zone discontinue de NAPL	Da, dar provocatoare	Necesară doză ridicată de oxidant sau co-solvent/surfactant

subterană				
>10 mg/L				
Concentrații scă subterană <1 mg/L	zute în	ара	Da, poate fi costisitoare	Costurile depind de necesarul de oxidant și de mărimea penei

Tabel 3.1: Aplicabilitatea generală a tehnicii ISCO (ITRC, 2005)

Testele efectuate folosind  $KMnO_4$  ca oxidant arată că condițiile ideale de aplicare a ISCO sunt îndeplinite la valori SOD/TOD (cererea de oxidant din sol/cererea totală de oxidant) sub 30 g/Kg sol uscat. Schemele din tabelele 3.2 și 3.3 raportează aplicabilitatea ISCO la cererea de oxidant din sol și la cererea totală de oxidant în g/Kg sol uscat, inclusiv poluantul, precum și la fracțiunea de carbon organic din sol.

SOD/TOD (g/Kg sol uscat)	Aplicabilitatea ISCO
<30	aplicabilă
>30	de investigat

Tabel 3.2: Relația dintre rata cererii de oxidant din sol și cererii totala de oxidant și aplicabilitatea ISCO

foc (%)	Aplicabilitatea ISCO
<0,3	aplicabilă
0,3 <foc<3< td=""><td>de investigat</td></foc<3<>	de investigat
>3	nerecomandată

Table 3.3: Relația dintre fracția de carbon organic din sol și aplicabilitatea ISCO

### 3.2.2 Caracteristici litostratigrafice și hidrogeologice ale sitului

Permeabilitatea și rata corespunzătoare de curgere a apelor subterane afectează distribuția oxidantului în acvifer și, prin urmare, succesul ICSO (a se vedea tabelul 3.4). O permeabilitate ridicată înseamnă, de obicei, un transport ridicat de oxidant. O permeabilitate scăzută reduce raza de influență (ROI), adică zona afectată de oxidanți; în acest caz, rețeaua de injectare trebuie îngroșată sau sunt necesare presiuni de injectare ridicate, folosind, de exemplu, hidrofracturarea în prezența unor aditivi adecvați.

Permeabilitatea (m/s)	Aplicabilitatea ISCO
>10-4 m/s	excelentă
10-5 ⇔ 10-4 m/s	aplicabilă
<10-5 m/s	nerecomandată

Tabel 3.4: Aplicabilitatea tehnicii ISCO în funcție de permeabilitate

Cu toate acestea, dacă viteza este prea mare, este necesar să se ia în considerare dacă timpul de contact între oxidant și poluant este suficient pentru a realiza reacția de oxidare și pentru a efectua tratamentul.

De asemenea, succesul ISCO depinde de adâncimea pânzei freatice (a se vedea tabelul 3.5). Intervalul optim pentru aplicarea ISCO în zona saturată este între 3 m și 15 m adâncime. La o adâncime a pânzei freatice mai mică de 3 m, este posibilă aflorirea pânzei freatice; în schimb, aplicarea pentru valori ale grosimii acviferului mai mari de 15 m este necesar a evalua considerentele economice.

Adâncimea pânzei de apă freatică (m)	Aplicabilitatea ISCO
<3	de investigat
3 ÷ 15	excelentă
>15	de luat în considerare

Tabel 3.5: Aplicabilitatea tehnicii ISCO în funcție de adâncimea pânzei de apă freatică

Grosimea stratului subsolului (m)	Aplicabilitatea ISCO
<15	aplicabil
>15	de luat în considerare

Tabel 3.6: Aplicabilitatea tehnicii ISCO în funcție de grosimea stratului de sol

Aplicarea tehnicii ISCO în zona vadoasă prezintă dificultăți legate de propagarea agenților oxidanți și de reactivitatea acestora cu solul.

### 3.2.3 Prezenta infrastructurii

Aplicarea tratamentelor *in situ* poate fi limitată de prezența infrastructurii îngropate și/sau a utilităților subterane; acestea pot fi deteriorate de activitățile de injectare, atât din cauza reactivității produselor, cât și a volumelor și presiunilor mari necesare pentru a dispersa reactivilor.

De asemenea, structurile îngropate pot afecta eficiența injectării din cauza prezenței unor potențiale căi preferențiale care ar putea devia reactivul și invalida operațiunea. De asemenea, prezența barierelor îngropate poate să limiteze eficiența intervenției deoarece acestea pot întârzia sau împiedica contactul cu contaminanții țintă.

În timpul studiului de fezabilitate este necesar să se efectueze investigații (geofizice, geoelectrice) care să ofere informații despre prezența infrastructurii drept suport pentru proiectarea executivă a intervenției.

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asemenea, să limiteze eficiența intervenției, deoarece acestea pot întârzia sau împiedica contactul cu contaminanții țintă.

În timpul studiului de fezabilitate este necesar să se efectueze investigații (geofizice, geoelectrice) care să ofere informații despre prezența infrastructurii ca suport pentru proiectarea executivă a intervenției.

### 3.3 Screening-ul secundar

În această fază, pe măsură ce se verifică condițiile descrise în prima fază de screening, este necesară o a doua fază mult mai detaliată. Trebuie evaluată influența altor factori, cum ar fi: pH-ul, alcalinitatea și salinitatea (concentrația de clorură). Variațiile valorilor pH-ului pot afecta transportul metalelor și ionilor în soluție, care pot reacționa cu radicalii produși de sistemul de oxidare, ceea ce ar putea scădea eficiența acestuia asupra poluanților vizați.

Salinitatea (Clorură mg/L)	Aplicabilitatea ISCO
<1000	aplicabilă
>1000	de investigat

Tabel 3.7: Aplicabilitatea tehnicii ISCO în funcție de salinitate

Alcalinitatea (mg/L as CaCO <sub>3</sub> )	Aplicabilitatea ISCO
<1000	aplicabilă
>1000	de investigat

Tabel 3.8: Aplicabilitatea tehnicii ISCO în funcție de alcalinitate

Factor	Detalii de luat în considerare
Tipul agentului oxidant	- capacitatea de oxidare a contaminanților primari problematici (COC)
	- capacitatea de oxidare a contaminanților secundari
	- capacitatea generală de oxidare a produșilor
	- capacitatea metodei de a funcționa în funcție de carbonul organic
	fracționat (FOC) din sitului
	- capacitatea metodei de a funcționa în funcție de pH-ul sitului
	- capacitatea metodei de a funcționa în funcție de alcalinitatea sitului
	- capacitatea metodei de a funcționa în funcție de clorura din sit
	- capacitatea metodei de a funcționa în funcție de distribuția masei
	de COC pe amplasament
Metodele de implementare	- adaptabilitate la mediul sitului
	- adaptabilitatea tehnicii de livrare la conductivitatea hidraulică a
	sitului
	- adaptabilitatea la eterogenitatea sitului
	- capacitatea de acţiune asupra contaminării din adâncime
	- capacitatea de acțiune asupra densității contaminanților
	- perturbarea activităților de la suprafața sitului
	- perturbarea activităților din subsol
Oxidanții și activatorii de reacție	- permanganatul

	<ul> <li>ozonul (incluzând ozonul individual şi ozonul activat cu peroxid de hidrogen</li> <li>peroxidul de hidrogen (inclusiv activarea cu fier/acidă, activarea cu chelaţi de fier sau procesul fără activare – cataliza mineralelor)</li> <li>percarbonatul</li> <li>persulfatul (inclusiv activarea alcalină, activarea termică, activarea</li> </ul>
	cu fier/acidă, activarea cu chelaţi, activarea cu peroxid sau procesul fără activare – cataliza mineralelor)
Metodele de injectare	<ul> <li>injectarea directă cu sondă de împingere</li> <li>sonde verticale de injectare</li> <li>sonde orizontale de injectare</li> <li>sonde verticale cu recirculare</li> <li>amestecul solului</li> <li>fractură hidraulică la imersarea amendamentului specific ISCO</li> <li>fractură pneumatică la imersarea amendamentului specific ISCO</li> <li>injectare prin șanț</li> <li>aplicare la suprafață/ în galerii de infiltrare</li> </ul>

Tabel 3.9- Factorii și detaliile de luat în considerare

### 3.4 Tratabilitatea contaminanților

Contaminanții provin din diferite clase de substanțe chimice, fiecare având proprietăți proprii, motiv pentru care prezintă o sensibilitate diferită la tratamentul de oxidare. Tabelul 3.10 prezintă potențialul de oxidare al diferiților contaminanți.

Foarte oxidabil	potential oxidabil
cloroetenă	cloroetan
clorobenzen	clorometan și bromometan
BTEX	explozivi
hidrocarburi aromatice policiclice (PAH)	pesticide
fenoli	N-Nitrozodimetilamină (NDMA)
MTBE	cetone
alcool	РСВ
1-4 dioxină	dioxine-furani

Tabel 3.10: Potențialul de oxidare al diferiților contaminanți

### 4 TESTARE LA FAȚA LOCULUI / ÎN LABORATOR

Dacă ISCO a fost identificat ca parte a unui proiect general de remediere, pasul care urmează studiului de fezabilitate este proiectarea remedierii utilizând tehnica ISCO. Așa cum s-a descris în capitolul introductiv, din faza de proiectare va face parte o serie de activități care includ un studiu aprofundat al modelului conceptual al sitului (RDC) și, atunci când este necesar, teste de laborator sau teste de teren la scară pilot.

### 4.1 Aspecte de proiectare

Principalele aspecte care trebuie evaluate în proiectarea remedierii ISCO sunt:

- alegerea tipului de oxidant;
- cantitatea de oxidant;
- alegerea sistemului de injectare.

### 4.1.1 Alegerea tipului de oxidant

La alegerea dintre posibilii oxidanți compatibili cu contaminanții se iau în considerare aspectele prezentate în continuare. Eficiența unui sistem de oxidare într-un anumit context depinde de diverși factori, cum ar fi: cinetica reacției, densitatea oxidantului, geologia, hidrogeologia, concentrația contaminanților și necesarul de oxigen al apei subterane/acviferului, denumită în general cerere naturală de oxidant (NOD). Caracterul adecvat al agenților oxidanți în funcție de acești factori a fost descris în secțiunile următoare.

### 4.1.1.1 Cinetica reacției

Acesta descrie distrugerea unui contaminant în timp. În situația în care concentrația oxidantului este mult mai mare decât concentrația compusului care urmează să fie oxidat, reacția urmează o cinetică de ordinul întâi. În consecință, viteza de reacție poate fi măsurată prin intermediul duratei medii (durată medie de viață).

Timpul de înjumătățire este timpul necesar reacției pentru a înjumătăți concentrația contaminanților. Durata de înjumătățire depinde de tipul de oxidant utilizat și de combinațiile de contaminanți prezente în subsol. Oxidarea chimică este fezabilă numai în cazul în care rata de oxidare a contaminantului este mai mare decât rata de interacțiune dintre oxidant și cererea de oxidant a acviferului.

De asemenea, cinetica reacției este influențată de procesele de dispersie, desorbție, dizolvare și difuzie care afectează atât transportul agenților oxidanți, cât și transportul contaminanților prin subsol.

Oxidanții chimici sunt insolubili în lichidele în fază neapoasă (NAPL), în timp ce oxidarea contaminanților are loc numai în fazele apoase. Prin urmare, trebuie să aibă loc mai întâi transferul de masă al contaminanților în faza apoasă, urmat de procesul de oxidare. Rata de eliminare a masei de contaminanți este strict legată de dizolvarea NAPL, un proces lent în comparație cu oxidarea. Pentru o distribuție uniformă a oxidantului se sugerează o densitate a oxidantului cât mai apropiată de densitatea contaminantului, pentru a promova aceleași căi de difuzie pentru ambii compuși.

### 4.1.1.2 Geologia și hidrogeologia

Transportul agentului oxidant în zona saturată se datorează în principal curgerii apelor subterane, legii lui Darcy și dispersiei. Difuzia joacă un rol esențial în cazul unui debit slab al apelor subterane sau în cazul introducerii unor produse deosebit de concentrate.

Putem distinge între trei tipuri de litologii: permeabilitate scăzută, moderată și ridicată. În tabelul 4.1 este prezentată adecvarea oxidanților în funcție de categoria de permeabilitate.

Litologie	Permanganat de sodiu/ potasiu	Peroxid de hidrogen	Percarbonat de sodiu	Persulfat de sodiu	Ozon
Foarte permeabil	+++	+++	+++	+++	+++
Puţin permeabil	+		-/+	+	Fără date
Moderat permeabil	++		+	++	Fără date

Tabel 4.1 – Alegerea oxidantului în funcție de categoria de permeabilitate

### 4.1.1.3 Necesarul de oxigen al apei subterane și a acviferului (NOD)

Distanța de transport a oxigenului în zonele necontaminate ale acviferului depinde nu numai de cererea totală de oxigen, ci și de următoarele variabile:

- rata de reacție a substanțelor cu substanțele nevizate;
- debitul apelor subterane;
- densitatea soluției.

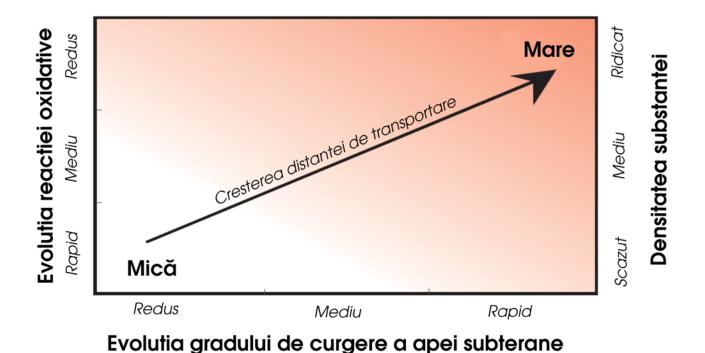


Figura 4.1 – Rata reacției de oxidare/ densității soluției, în funcție de viteza de curgere a apei subterane

<sup>-/+</sup> chestionabil, + adecvat, ++ foarte adecvat, +++ foarte recomandat

### 4.1.1.4 pH

ISCO poate avea un impact considerabil asupra pH-ului solului, fie deoarece oxidantul poate fi asociat cu posibila producere de protoni sau deoarece poate fi asociat cu posibila producere de ioni hidroxil direct în timpul reacției. Amploarea efectului pH-ului depinde de capacitatea de tamponare a solului și, în consecință, de concentrația de carbonați. Prin urmare, concentrația de carbonați influențează cinetica reacției. În tabelul 4.2 este prezentată adecvarea oxidanților în funcție de pH-ul subsolului.

рН	Permanganat de sodiu/ potasiu	Peroxid de hidrogen	Percarbonat de sodiu	Persulfat de sodiu	Ozon
<5	+++	+++		+++	+++
5-6	+++	+++	+	+++	+++
6-7	+++	++	++	+++	+++
7-8	+++	+	+++	+++	++
8-9	+++	-	+++	+++	++
>9	++		+++	+++	+

Tabel 4.2 Alegerea oxidantului/pH

### 4.1.1.5 Fracția de materie organică (f<sub>oc</sub>)

Atunci când se alege tipul de oxidant, este important să se evalueze reactivitatea produsului cu substanțele organice care nu sunt vizate (materia organică din subsol), pentru a crește SOD. În tabelul 4.3 este prezentată adecvarea oxidanților în funcție de fracția de carbon organic din subsol.

f <sub>oc</sub>	Permanganat de sodiu/ potasiu	Peroxid de hidrogen	Percarbonat de sodiu	Persulfat de sodiu	Ozon
>3%			-	+	
1-3%	-	-	+	++	-
0,3-1%	++	++	+++	+++	++
0,1-0,3%	+++	+++	+++	+++	+++
<0,1%	+++	+++	+++	+++	+++

Tabel 4.3 – Alegerea oxidantului în funcție de conținutul de carbon organic din subsol (foc)

<sup>--</sup> cu siguranță neadecvat, + adecvat, ++ foarte adecvat, +++ foarte recomandat

<sup>--</sup> cu siguranță neadecvat, + adecvat, ++ foarte adecvat, +++ foarte recomandat

### 4.1.1.6 Concentrația contaminantului

Concentrația de contaminanți este, de asemenea, un aspect care trebuie luat în considerare în alegerea agentului oxidant. Este necesar să se utilizeze oxidanți foarte reactivi, în timp ce în zona penei de poluare se sugerează să se aleagă reactanți mai puțin reactivi pentru a maximiza aria de influentă. În tabelul 4.4 este prezentată adecvarea oxidanților în funcție de concentrația contaminanților.

Concentrația contaminantului	Permanganat de sodiu/ potasiu	Peroxid de hidrogen	Percarbonat de sodiu	Persulfat de sodiu	Ozon
foarte mică					+
mică	++	++	++	++	++
moderată	+++	+++	+++	+++	+++
ridicată	++	+++	++	+++	+
foarte ridicată		++	+	++	-

Tabel 4.4 Alegerea oxidanților în funcție de concentrația contaminantului

### 4.1.1.7 Compatibilitatea oxidanților cu mediul

Cinetica reacției, concentrația oxidantului, pH-ul și temperatura acviferului, concentrația de contaminanți și cererea de oxigen din sol (SOD) fac parte din setul de variabile care determină "longevitatea" oxidantului, adică persistența oxidantului atunci când este aplicat pe substratul care urmează să fie tratat. Acest aspect este de o importanță fundamentală, deoarece afectează raza de influență (ROI) pe care oxidantul o poate atinge, atunci când este încă activ.

După cum s-a menționat în capitolul introductiv, tehnica ISCO este o metodă care rareori poate fi implementată de una singură ca tehnologie de remediere, în special în cazul unor limite de reglementare stricte. În mod obișnuit, este necesară o remediere combinată. Aceasta implică o etapă ulterioară care ar putea fi o bioremediere îmbunătățită sau accelerată.

O abordare ecologică și eficientă pentru tratarea acestor constituenți este utilizarea unui compus pasiv cu eliberare controlată pentru a stimula biodegradarea *in situ*. Bioremedierea este eficientă la mineralizarea intermediarilor formați în timpul oxidării, care altfel ar rămâne ca produși secundari nedoriți. Bioremedierea poate fi ultima etapă rentabilă în atingerea obiectivului general al unui proiect de restaurare a apelor subterane.

Prin urmare, în selectarea oxidanților, este necesar să se ia în considerare cu atenție numai agenții oxidanți care nu sunt agresivi față de microorganismele din subsol.

În cazuri specifice, este necesar să se verifice dacă produșii secundari ai reacției nu agravează condițiile hidrochimice ale apelor subterane, în special în cazul în care este prezent un receptor sensibil și/sau dacă

<sup>-</sup> neadecvat, + adecvat, ++ foarte adecvat, +++ foarte recomandat

resursa de apă subterană are utilizări speciale. Exemple de subproduse generate sau de substanțe mobilizate de reacția de oxidare sunt: sulfați, mangan, crom și alte metale grele.

Prezența structurilor subterane, a conductelor sau a sistemelor de canalizare poate constitui o constrângere importantă în timpul selecției agentului de oxidare. Nu se recomandă injectarea de volume mari de produs în apropierea fundațiilor. Aceeași concluzie este valabilă și pentru utilizarea de oxidanți care necesită un pH scăzut în apropierea rezervoarelor subterane, a conductelor sau a utilităților delicate.

### 4.1.2 Cantitatea de oxidant

Pentru a determina cantitatea de reactiv necesară pentru o oxidare chimică la fața locului, este necesar să se identifice cererea totală de oxigen (TOD) necesară pentru tratarea specifică locului. TOD include cererea de oxigen pentru a oxida contaminanții țintă și oxigenul necesar substanțelor acceptante de electroni "non-țintă" conținute în subsol (NOD/SOD).

#### 4.1.2.1 Contaminanții de interes

Cererea de oxidant legată de contaminanții de interes (CoC) trebuie evaluată în toate fazele posibile:

- faza dizolvată;
- faza adsorbită;
- faza liberă;
- lichide în fază neapoasă (NAPL);
- faza de vapori (zona vadoasă).

Pentru a determina cererea de oxigen, trebuie mai întâi să se evalueze masa totală a fiecărui tip de contaminant prezent în subsol. Ulterior, trebuie să se estimeze lățimea, lungimea și adâncimea zonei sursă. În cele din urmă, în funcție de tipul de sol (pietriș, nisip, nămol sau argilă), trebuie efectuată o evaluare cantitativă a volumului, densității și volumului porilor solului contaminat.

Valoarea masică a fazei dizolvate poate fi calculată prin analizarea concentrațiilor de contaminanți prezenți în sondele de monitorizare. Cererea de oxigen legată de faza absorbită, pe de altă parte, poate fi estimată fie direct din analiza probelor de sol colectate *in situ*, fie indirect, prin calcule integrale ale masei stoichiometrice a contaminanților. Aceasta depinde de densitatea materialului acvifer, de fracția de carbon organic (foc) și de coeficientul de partiție a carbonului organic în apa poroasă a contaminantului (Koc). Densitatea solului și valorile foc pot fi estimate în funcție de tipul de sol, în timp ce valoarea Koc poate fi obținută din literatura de specialitate sau din baze de date online.

Aprecierea masei NAPL în fază liberă este adesea complexă. În acest sens, API și EPA din SUA au elaborat diverse metode de calcul.

#### 4.1.2.2 Matricea

În mod evident, reactivii injectați în subsol vor reacționa cu substanțele organice și anorganice prezente în mod natural în subsol. Deoarece în anumite cazuri cantitatea de oxigen necesară poate fi semnificativă, trebuie acordată o atenție deosebită cerințelor de bază ale oxidanților care se bazează pe reacții catalitice sau pentru care se folosesc alți reactivi ca stabilizatori. Un exemplu de acest tip de sistem este peroxidul de hidrogen catalizat ISCO. Peroxidul de hidrogen va forma rapid complexe de suprafață și va reacționa cu metalele tranziționale, cum ar fi fierul, pe suprafețele minerale. Un alt factor care trebuie luat în considerare în procesele pe termen lung este potențialul proceselor de transport de a transporta componente reactive suplimentare în zona de tratare, așa cum s-a menționat în secțiunea 4.1.1.

#### 4.1.2.3 Determinarea cererii de oxidant

Există două metode de calcul a cererii de oxidant:

- Printr-un sistem bazat pe carbonul organic total (TOC) şi pe necesarul chimic de oxigen (COD);
- Prin intermediul fracției molare.

Cantitatea de oxidant utilizată pentru reacție trebuie să fie mai mare decât cererea teoretică de oxidant pentru a asigura o cantitate suficientă de reactant în scopul menținerii cineticii de ordinul întâi.

### 4.1.3 Adăugarea amendamentului

Principalele aspecte care trebuie luate în considerare în proiectarea injectarei de reactivi sunt:

- Eterogenitatea litostratigrafică care condiționează alegerea metodei de injectare şi structura. Metoda de injectare directă permite o mai mare versatilitate în distribuția reactivului prin ajustarea intervalelor verticale și orizontale de injectare pe baza diferitelor permeabilități ale straturilor care urmează să fie tratate. Astfel, se evită ca reactivul să fie distribuit în principal în straturile mai permeabile; o situație care accentuează fenomenul de ricoșeu. Rezoluția spațială orizontală și verticală a valorii ROI trebuie planificată odată cu activitățile de stabilire a caracteristicilor proiectului de remediere, în funcție de eterogenitatea litostratigrafică a sitului.
- Rezultatele injectărilor de testare în timpul experimentelor desfășurate la scară pilot. Se recomandă efectuarea de teste de injectare ca parte a implementării la scară pilot pentru a obține informații privind valorile presiunii de injectare și volumele de reactiv aplicabile pentru fiecare strat omogen.
- Rezultatele studiilor cu trasori (de exemplu, litiu și fluoresceină) care pot fi utilizate pentru a sprijini activitățile la scară pilot.

### 4.1.4 Volumele de reactiv care urmează a fi injectat

Pentru a efectua un tratament eficient, trebuie injectată o cantitate suficientă de oxidant în spațiul poros al solului pentru a garanta cinetica de ordinul întâi a reacţiei.

Volumul de reactiv care trebuie injectat se calculează pe baza porozității efective a masei de sol care urmează să fie tratat. În prezența unei geologii eterogene, este recomandabil să se estimeze valorile porozității efective pentru fiecare strat în parte, de preferință pe baza unei analize granulometrice prin sedimentare.

Este necesar să se injecteze un volum care să echivaleze cu 10% până la 50% din porozitatea efectivă. Procentul de spațiu gol care trebuie tratat direct prin injectare depinde de ROI-ul proiectat, deoarece este de așteptat ca partea rămasă a microporilor să fie atinsă de reactiv prin advecție.

Un studiu de implementare la scară pilot permite dobândirea de informații detaliate privind cinetica reacțiilor care guvernează transferul de masă al oxidantului prin advecție și desorbție. Acest lucru permite estimarea numărului de injectări necesare, a intervalului de timp dintre injectări și a dozei optime de oxidant pentru fiecare injectare.

### 4.1.5 Accesibilitatea zonei de intervenție

- În cazul în care operațiunea de tratare implică zone cu activități în desfășurare sau cu acces public (de exemplu, drumuri, zone școlare etc.), trebuie luate în considerare și costurile asociate cu ocuparea temporară a acestor zone.
- În acest caz, este necesar să se evalueze dacă numărul de injectări necesare face ca instalarea de sonde de injectare fixe (sondecu supapă) să fie benefică din punct de vedere economic.

### 4.1.6 Tehnologii de injectare

Cele mai utilizate tehnologii pentru injectarea reactivului în acvifer sunt următoarele:

- Injectarea prin tehnologia de împingere directă Injectarea de reactiv în acvifer se realizează cu ajutorul unor tije de oțel goale cu fantă și cu ajutorul unor pompe speciale cu piston care permit atingerea unor presiuni ridicate (> 50 bar);
- Sonde cu supapă Acestea sunt puncte de injectare fixe, formate dintr-o țeavă de PVC, instalate prin
  carotaj continuu și etanșarea cavității cu beton. Conducta este echipată cu grupuri de 4 găuri pe același
  plan la o distanță de 30-50 cm. Supapele sunt acoperite cu un manșon elastic care acționează ca o
  supapă de reținere;
- Piezometre existente Injectarea se face folosind secțiunea de filtrare a piezometrelor sigilată.

Fiecare tehnologie are avantaje și dezavantaje. Metoda de împingere directă permite variația poziției punctelor de injectare pentru fiecare campanie. Acest lucru face posibilă consolidarea semnificativă a grilei de injectare și, în consecință, garantarea unei probabilități mai mari de contact cu contaminantul care urmează să fie tratat. O distanță mai mică permite, de asemenea, reducerea presiunii de injectare, cu un risc mai mic de fracturare a matricei și, prin urmare, o eterogenitate a tratamentului și posibilitatea ca produsul să se ridice de-a lungul tijei de injectare. Limita tehnică a tehnologiei, în ceea ce privește adâncimea de injectare, este de 30-35 de metri. Utilizarea tehnologiei de împingere directă devine nerentabilă atunci când sunt necesare mai mult de 5-6 campanii de injectare.

Tehnologia care oferă puncte fixe de injectare (sonde) are următoarele avantaje:

- Adâncimea injectării până la 100 m;
- Presiuni de injectare ridicate, până la 90 bari;
- Posibilitatea de a utiliza amestecuri foarte vâscoase;
- Un control mai mare al intervalul de injectare pe verticală;
- Rentabilitatea tratamentului, în cazul în care este necesar un număr mare de injectări (>5-6);
- Impact mai redus pentru tratamente în zone cu activități în desfășurare.

Utilizarea piezometrelor existente are avantajul economic al reutilizării materialului deja prezent în zona de tratament. Cu toate acestea, în cele mai multe cazuri, nu permite o distribuție omogenă în zona de tratament, deoarece piezometrele au fost proiectate cu în scopuri diferite. Tratamentul cu piezometre existente poate fi totuși inclus într-un proiect care integrează diferitele tehnologii de injectare, pentru a maximiza eficiența globală a intervenției.

Imaginile următoare prezintă metodele de injectare cu sonde și cu împingere directă.

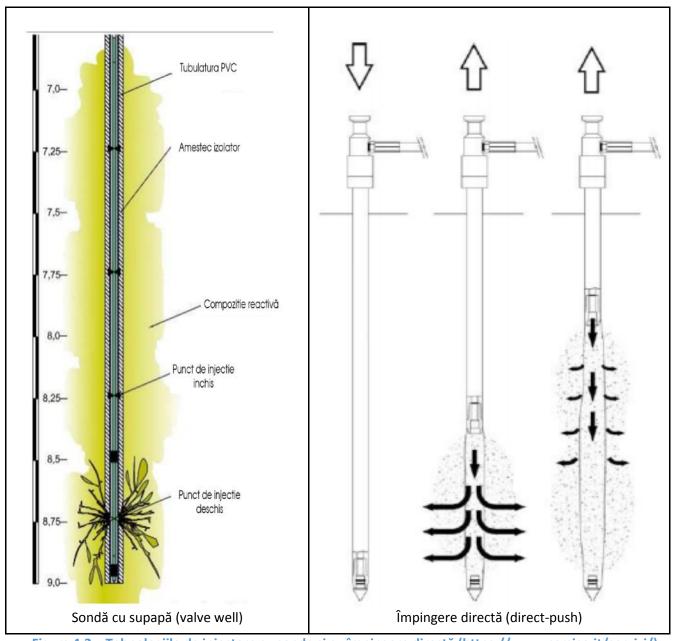
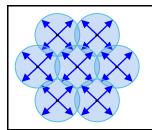


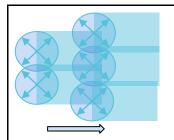
Figura 4.2 – Tehnologiile de injectare cu sonde și cu împingere directă (https://www.carsico.it/servizi/)

Mai multe tipuri de strategii de injectare sunt descrise în figurile următoare.



#### Retea

Este cea mai versatilă metodă pentru a asigura o distribuție omogenă a produsului în zona de interes.



GWFLOW

#### Injectare și drift/infiltrare

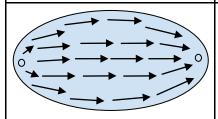
Acest sistem profită de curgerea apelor subterane pentru a răspândi produsul sub gradient prin advecție. Această abordare poate fi utilizată în zonele în care viteza de curgere este mare și litologia este relativ omogenă.

Agentul oxidant este aplicat în sol prin intermediul sondelor de infiltrare, fără ajutorul acțiunilor mecanice sau al presiunilor. În cazul unui sistem pasiv, este necesar să se țină seama de capacitatea de infiltrare a solului, de adâncimea pânzei freatice, de viteza de curgere a apelor subterane și de durata de viață a oxidantului. Capacitatea de infiltrare a solului este strâns legată de tipul de sol. Atunci când se elaborează configurația de infiltrare, este important să se estimeze capacitatea de infiltrare cât mai exact posibil. Valorile prezentate în tabelul de mai jos reprezintă o estimare aproximativă a capacității de infiltrare. Testele de infiltrare a solului pot ajuta la determinarea unei estimări mai precise a capacității de infiltrare a solului. Ar trebui să se realizeze că nu există niciun motiv pentru a testa capacitatea de infiltrare a solurilor care au deja o permeabilitate scăzută.

Utilizarea unei metode de aplicare indirectă necesită un debit de apă subterană mai mare de 0,05 m/zi. În cazul în care debitul de apă subterană este mai mic de 0,05 m/zi, <u>aplicarea metodei de infiltrare</u> pasivă este complexă, deoarece oxidantul va fi distribuit insuficient.

Infiltrarea este o tehnică de injectare valabilă numai dacă sunt îndeplinite următoarele conditii:

- Oxidantul trebuie să rămână activ în subsol suficient de mult timp pentru a oxida contaminanții.
- Regula de bază:
  - o timpul de înjumătățire a oxidantului în mediu este mai mare decât dublul timpului de reacție
  - o oxidantul trebuie să rămână stabil în sol suficient de mult timp pentru a obține o rază de influență mare.



### Recircularea

Această strategie constă în injectarea oxidantului într-un punct și extragerea simultană a apei subterane într-un alt punct.

Utilizarea acestei strategii se limitează, de obicei, la siturile cu transmisivitate relativ ridicată.

Metoda este o combinație între un sistem de pompare și tratare și ISCO. Acest proces are avantajul de a crea un gradient hidraulic ridicat în zona contaminată și, în consecință, o zonă de influență mai mare.

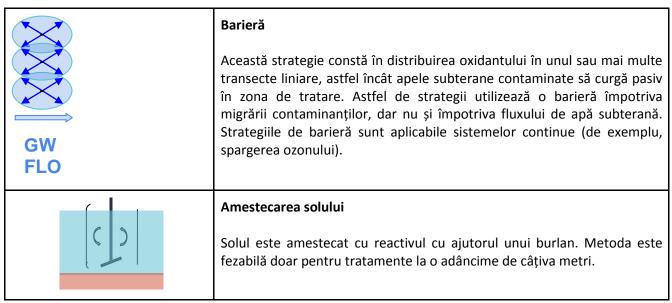


Figura 4.3- Tipuri de strategii de injectare

parametru	filtrare	împingere directă	recirculare	infiltrare	amestecarea solului
> 10 <sup>-5</sup> m/sec	+++	+++	+++	+++	+++
10 <sup>-6:</sup> ÷ 10 <sup>-3</sup> m/sec	++	+++	+	++	+++
10 <sup>-7:</sup> ÷ 10 <sup>-8</sup> m/sec	-	-		-	+++
<10 <sup>-8</sup> m/sec					+++

Tabel 4.5- Aplicabilitatea tehnologiilor de injectare în funcție de conductivitatea hidraulică

-- cu siguranță neadecvat, - neadecvat, + adecvat, ++ foarte adecvat, +++ foarte recomandat

parametru	filtrare	împingere directă	recirculare	infiltrare	amestecarea solului
< 5 m bgl	+++	+++	+++	+++	+++
5 <sup>:</sup> ÷ 10 m bgl	+++	+++	+++	-	+++
10 <sup>:</sup> ÷ 25 m bgl	+++	++	+++		-
25 <sup>:</sup> ÷ 50 m bgl	++	+	++		
> 50 m bgl	++		++		

Tabel 4.5- Aplicabilitatea tehnologiilor de injectare în funcție de adâncimea zonei de tratament

<sup>--</sup> cu siguranță neadecvat, - neadecvat, + adecvat, ++ foarte adecvat, +++ foarte recomandat

În *Dal Santo* și *Prosperi (2020)* sunt prezentate argumentele pro și contra pentru fiecare metodă aplicabilă, conform tabelului 4.6.

METODA	APLICABILITATE	PRO	CONTRA
Împingere directă	Pentru aplicarea tuturor tipurilor de produse	Distribuție bună în acvifer dacă este proiectat cu o plasă adecvată. Nu afectează funcționalitatea sondelor rețelei de monitorizare	Puncte nerepetabile. Este necesar un sistem de geoprobe pentru a repeta injectarea.  În timpul aplicării într-un acvifer fin, în puţine cazuri se poate înregistra o ascensiune a reactivului în spaţiul inelar
Tuburi cu supapă	Tuburi cu supapă  Pentru aplicarea tuturor tipurilor de produse		Costuri suplimentare pentru instalarea rețelei de injectare prin tuburi cu supapă

Piezometrele existente	Pentru aplicarea tuturor tipurilor de produse	Nu există costuri suplimentare pentru construirea punctelor de injectare	Locația și intervalul de filtrare al sondelor sunt deja definite.  Injectarea poate afecta funcționalitatea rețelei de monitorizare prin ocluzii parțiale și formarea de produși secundari în coloana de sonde.  Injectarea nu este complet controlată prin utilizarea filtrării sondelor pentru distribuirea reactivului și nu a tuburilor de supapă.
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Tabel 4.6- Argumente pro și contra pentru fiecare metodă aplicabilă (conform Dal Santo și Prosperi, 2020)

### 4.2 Teste de laborator și teste pilot

Decizia de a utiliza testele la scară de laborator și/sau la scară pilot ar trebui evaluată în funcție de complexitatea și dimensiunea amplasamentului.

Investiția în costurile de obținere a informațiilor ar trebui să fie echilibrată de reducerea incertitudinilor care ar putea duce la neatingerea obiectivelor de tratare ISCO. Procesul de obținere de informații din activități la scară de laborator sau pilot este iterativ și se dezvoltă pe baza necesității de a maximiza eficacitatea și eficiența intervenției globale.

### 4.2.1 Teste de laborator

Informațiile care pot fi obținute din testele de laborator sunt:

- informații privind cinetica reacției, formarea de produși intermediari de reacție (inclusiv gaze) și căldura produsă;
- cererea de oxigen a contaminanților dizolvați sau saturați în sol;
- cererea de oxigen a matricei solului;
- mobilizarea potențială a metalelor;
- capacitatea tampon a solului;
- efectele potențiale asupra permeabilității (de exemplu, formarea de MnO<sub>2</sub>);
- substanțele oxidante care fac ca reacția de oxidare să fie mai eficientă;
- informații pentru calcularea razei de influență (ROI).

În general, testele de laborator nu sunt reprezentative pentru condițiile din teren din cauza problemelor de scară și a eterogenității condițiilor hidrogeologice, a cineticii de reacție și a altor caracteristici fizice sau chimice

care nu pot fi obținute în laborator. În ciuda acestor limitări, rezultatele testelor de laborator pot oferi o evaluare inițială, la nivel de screening, a eficacității potențiale a produsului reactiv/comercial asupra contaminanților din zona care urmează să fie tratată. Cunoștințele dobândite pot fi utilizate pentru a proiecta și a pune în aplicare un test pilot. Testele de laborator ar trebui să fie concepute pentru a îndeplini obiectivele prestabilite și nevoile specifice de proiectare.

#### 4.2.2 Teste Pilot

Testele pilot sunt intervenții de tratament la scară mică, cu aceeași schemă de proiectare prevăzută pentru tratamentul întregii zone.

Setul de activități care urmează să fie efectuate în cadrul testului pilot vizează reducerea incertitudinii asociate cu prezența a numeroase variabile legate de eterogenitatea sitului, de prezența constrângerilor structurale și de performanța așteptată în ceea ce privește reducerea contaminării. Obiectivele testului pilot sunt, prin urmare, evaluarea:

- fezabilității tehnice a ISCO;
- compatibilității cu limitele bugetare (ca parte a unei intervenții globale de remediere);
- datelor de proiectare a intervenţiei, în ceea ce priveşte procesul şi performanţa.

Zona de testare trebuie să fie identificată ținând cont de obiectivele tratamentului cu oxidant. Cea mai eficientă utilizare a oxidării chimice are loc acolo unde concentrația contaminanților țintă este cea mai mare, adică în zona sursei. În cazul în care strategia de intervenție include și "tratarea penei de poluare", este necesar să se planifice injectarea de reactivi pentru a evita atât riscul de a nu distinge fenomenele normale de "ricoșeu", cât și infiltrarea contaminării din zonele de deasupra – zonele de gradient.

Informațiile care urmează să fie colectate în timpul fazei pilot ar trebui să verifice eficiența din punctul de vedere al fezabilității, al procesului și al performanței proiectului de remediere. Prin urmare, în timpul fazei pilot, poate apărea necesitatea de a reevalua fazele anterioare și de a dobândi cunoștințe suplimentare în ceea ce privește caracterizarea.

Colectarea de informații în cadrul testului pilot se referă în principal la datele de proces (selectarea reactivului oxidant și aplicarea acestuia în zona de tratare țintă) și la datele de performanță (reducerea contaminării și a efectelor secundare). Pe baza feedback-ului obținut, trebuie verificată necesitatea de a dobândi noi informații (caracterizarea proiectului de remediere) și/sau de a reevalua fezabilitatea tehnologiei și/sau a abordării de intervenție.

### 4.2.3 Monitorizarea procesului

Succesul tehnicilor de remediere *in situ* este puternic condiționat de aplicarea corectă a reactivului în zonele care urmează a fi tratate. Monitorizarea procesului are ca scop controlul parametrilor tehnici legați de activitățile de injectare, precum și a răspunsurilor zonei de tratament în ceea ce privește perturbarea parametrilor fizico-chimici așteptați. În cazul în care datele dobândite în timpul și după activitățile de injectare evidențiază situații neprevăzute anterior de proiectul de intervenție, este necesar să se repete etapele descrise mai sus, pentru a asigura o intervenție eficientă "la scară reală".

### 4.2.4 Monitorizarea preformației

nivelul apei subterane	Creșterile neobișnuite ale nivelului apelor subterane fac posibilă verificarea prezenței oricăror căi preferențiale de circulație a lichidelor în matricea solului.
presiunea de injectare	Presiunile de injectare mai mari decât cele preconizate pot fi cauzate de permeabilitatea scăzută a substratului care urmează să fie tratat. O creștere a presiunii pentru a compensa rezistența matricei poate produce o distribuție necontrolată a reactivului din cauza fracturării. Prin urmare, este necesar să se dobândească cunoștințe suplimentare. Presiuni de injectare mai mici decât cele prevăzute, eventual asociate cu o creștere a debitului, pot fi cauzate de prezența unor căi preferențiale (de exemplu, conducte de cabluri, canalizări).
parametrii fizico-chimici	Valorile neașteptate ale conductivității, temperaturii, pH-ului, potențialului red-ox și ale oxigenului dizolvat sugerează prezența unor căi preferențiale sau un ROI insuficient.

Tabel 4.6 Cei mai importanți parametri de proces

#### 4.2.4.1 Indicatori

Se pot identifica diferite tipuri de indicatori de performanță pentru a măsura scăderea progresivă a contaminării, de exemplu:

- concentrația de contaminanți indicator utilizat pentru a compara cu limitele de reglementare (MCL) sau pentru a evalua tranziția către alte tehnologii (Ex.: bioremediere, MNA). Concentrația poate fi evaluată din punct de vedere spațial, utilizând hărți cu izoconcentrații și din punct de vedere temporal, prin calcularea tendinței cu ajutorul testelor statistice (Ex.: Mann Kendall);
- rata de epuizare a masei indicator utilizat pentru a demonstra gradul de eficiență a tratamentului.
   Evaluarea masei distruse poate fi obținută prin calcularea bilanțului masei totale. Pentru a realiza o evaluare riguroasă a masei (inclusiv a NAPL), trebuie să se preleveze și probe de sol saturat. O altă modalitate mai puțin riguroasă, care subestimează totuși masa reală, se bazează pe variația masei dizolvate.
- flux masic indicator utilizat pentru a demonstra reţinerea contaminantului în zona sursă.

#### 4.2.4.2 Rețeaua de monitorizare

Locația punctelor de monitorizare din zona testată trebuie să fie planificată cu scopul de a măsura performanța tratamentului și, prin urmare, obiectivele intervenției ISCO. Pot fi identificate următoarele zone:

- tratament zona afectată de tratament pe baza ROI al fiecărui punct de injectare;
- tranziție zona afectată de efectele geochimice produse de reactiv;
- penajul zona din pana de poluare cu contaminare reziduală;
- punctele în care s-au efectuat injectări pot fi utilizate în scopuri de monitorizare, doar în anumite cazuri, deoarece ar putea furniza informații nerealiste.

Punctele de monitorizare din zona sursei sunt utilizate pentru a verifica raza de influență (ROI).

Numărul de piezometre necesare depinde de obiectivele remedierii:

- pentru a verifica reducerea masei contaminantului în zona sursă, sunt suficiente piezometrele din zona tratată;
- pentru a evalua conformitatea cu limitele de reglementare (MCL), este necesar să se prevadă puncte în zona penei de poluare;
- pentru a evalua persistența efectelor secundare datorate tratamentului în ceea ce privește efectele asupra concentrației în apele subterane a subproduselor (Ex.: sulfați, Mn) și/sau mobilizarea poluanților (Ex.: metale grele), este necesar să se prevadă puncte de control în zonele de tranziție geochimică.

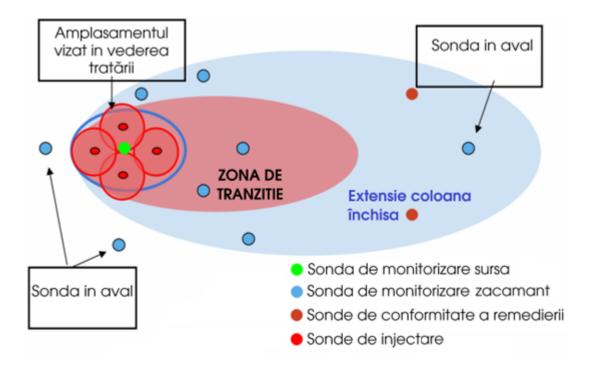


Figura 4.4: Poziționarea sondelor de monitorizare

### 4.2.4.3 Frecvența

Frecvența monitorizării trebuie să permită o înțelegere a evoluției efectelor tratamentului. În primele etape ale remedierii, frecvența trebuie să fie foarte mare, în timp ce în perioadele ulterioare poate fi redusă pe baza evaluării datelor colectate.

Aspectele tipice care trebuie monitorizate în scopul evaluării performanței tratamentului sunt următoarele:

- parametrii care permit verificarea longevității reactivului, de exemplu, pH, potențialul red-ox (ORP), oxigenul dizolvat DO;
- fenomenul de "ricoșeu" legat de mecanismele de transfer de fază (desorbție și dizolvare) ale contaminantului;
- perioada de înjumătățire a concentrației de poluant legată de cinetica de reacție.

Unele criterii pentru planificarea frecvenței de monitorizare sunt:

- viteza de deplasare a pânzei freatice;
- cinetica de reacție a produsului oxidant.

### 5 MONITORIZAREA

### 5.1 Tipuri de teste

Înainte de a selecta și de a aplica un agent de oxidare prin tehnica ISCO, este necesar să se cunoască în detaliu condițiile hidrogeologice ale sitului și geochimia subsolului pentru a decide tipul, metoda și cantitatea de agent aplicat. Deoarece aceste condiții pot fi foarte diferite, monitorizarea este esențială pentru o aplicare reușită a tehnicii ISCO. Prin urmare, înainte de realizarea remedierii propriu-zise, se recomandă să se efectueze:

- Teste de laborator. Scopul acestor teste este de a evalua eficiența unui anumit tip de reactiv pe un eșantion de material de sol din sit și de a calcula consumul acestuia.
- Teste de urmărire. Scopul acestor teste este de a exclude existența unor căi preferențiale nedorite prin care reactivul ar putea să se scurgă. Prin urmare, aceste teste trebuie să caracterizeze direcția și viteza reală de curgere a apelor subterane și de transport al contaminanților și reactivilor. În acest scop, pot fi utilizate unele tipuri de fluoresceină, LiCl etc. Rezultatele testului de urmărire trebuie să furnizeze datele necesare pentru a preciza sistemul de intensificare a procedurii de remediere, care va consta în infiltrarea reactivilor ISCO în cadrul campaniilor de aplicare și, dacă este cazul, în aplicarea soluțiilor de suport PAL (detergenți anionici). Uneori, o direcție dorită de deplasare a reactivului infiltrat poate fi manipulată prin pomparea unor foraje selectate și infiltrarea pe partea cealaltă. După începerea testului, probele vor fi prelevate la un interval de timp corespunzător condițiilor hidrogeologice, de exemplu, o dată pe zi timp de 5 zile într-un sol nisipos, dar semnificativ mai des în cazul solurilor cu permeabilitate mai mică.
- Teste semi-operaționale la fața locului. Scopul acestor teste este de a evalua tehnica ISCO în timpul funcționării. Testele se efectuează pe un foraj selectat, pe un interval de timp de aproximativ 1 lună. Ca urmare, se poate adapta doza de agenți de oxidare, detergenți și parametri precum cantitatea de agent de oxidare, metoda și frecvența de dozare.

Este recomandabil să se combine ISCO cu alte intervenții de remediere *in situ* în zona saturată, utilizând metode de remediere hidraulică și tehnologii de spălare de sprijin cu agenți tensioactivi (PAL). Aplicația PAL este concepută pentru a elimina faza liberă, în timp ce aplicațiile ISCO sunt localizate în afara zonelor sursă, către părțile periferice ale complexului, în direcția de curgere a apelor subterane, pentru a curăța poluanții dizolvați. Infiltrarea poate fi realizată prin foraje verticale, foraje orizontale, pereți reactivi și sonde de presiune.

Selectarea locației sondelor de monitorizare trebuie să fie compatibilă cu poziția sondelor de infiltrare și a punctelor fierbinți de contaminare - la intrarea apelor subterane în sit (sonde de referință) și la ieșirea din sit (sonde de monitorizare), precum și a obiectelor pompate și a sondelor din penele de poluare.

Se pot utiliza diverse metode pentru a echilibra cantitatea totală de contaminanți din subteran și cantitatea de contaminanți degradată:

- Stabilirea echilibrului dintre un contaminant degradat și modificarea cantității totale de contaminanți dintr-un sit. În cazul metodelor de remediere in situ, se pot efectua modificări ale concentrațiilor de produse de degradare de la un capăt la altul pentru un echilibru al contaminantului degradat. În cazul degradării in situ a hidrocarburilor clorurate, bilanțul contaminantului degradat poate fi realizat în condiții adecvate pe baza modificărilor concentrațiilor de cloruri. Cu toate acestea, utilizarea clorurilor exclude destul de des concentrațiile adiacente ridicate ale acestora în apele subterane.
- Stabilirea echilibrului unui contaminant degradat pe baza unei cantități de substanțe consumate.

- Stabilirea echilibrului unui contaminant degradat pe baza unei modificări a concentrațiilor de produși de degradare.
- Efectuarea bilanţului unui contaminant degradat pe baza unei modificări a raportului izotopic C12/13 şi Cl 35/37. Aceasta este cea mai recentă şi probabil cea mai precisă modalitate de bilanţ a substanţelor organice descompuse in situ. Metoda se bazează pe monitorizarea modificărilor în compoziţia izotopică a C12/13, ca urmare a degradării in situ a unei contaminări cu hidrocarburi. Recent, a început să fie utilizat şi raportul izotopic Cl35/37. Probabil, acesta este cel mai promiţător mod de a realiza bilanţul hidrocarburilor organice degradate in situ.

### 5.2 Tipuri de monitorizare

### 5.2.1 Monitorizarea operațional-tehnologică

Scopul este de a monitoriza concentrația reactivului și mișcarea acestuia în subteran, precum și funcționalitatea dispozitivelor.

În timpul remedierii, se monitorizează concentrațiile de poluanți și reactivi din sit în sondele disponibile și se evaluează în permanență dacă remedierea se realizează corect. Rezultatele sunt evaluate periodic și prezentate în rapoarte anuale, cu recomandări.

### 5.2.2 Monitorizarea continuă și în etapa finală

Scopul este de a evalua dacă obiectivele remedierii au fost atinse cu succes.

Este posibil să se înceapă dovedirea atingerii parametrilor țintă de remediere numai în momentul dispariției substanței suport (reactiv) și a efectelor care decurg din prezența acesteia în subsol (reactiv inactiv/nereacționat).

Există mai multe abordări de bază care pot fi utilizate pentru a evalua atingerea parametrilor țintă ai remedierii:

- Obiectivul de remediere este atins atunci când concentrațiile din zona de interes nu depășesc obiectivele de remediere (ca valori de screening). Această abordare reprezintă o toleranță zero pentru depășirea obiectivelor de remediere și conduce la rezultate optime de remediere. Cu toate acestea, ea poate duce la costuri de remediere excesive, în special în cazul unor condiții naturale complicate, în care nu se poate defini cu precizie, în condiții tehnice și economice acceptabile, o locație și o cantitate de contaminare în subteran sau o imposibilitate de atingere a obiectivelor.
- Obiectivul de remediere este atins atunci când concentrațiile pe majoritatea obiectelor de remediere și/sau de monitorizare dintr-o zonă de interes nu depășesc obiectivele de remediere. De exemplu, se depășește cu 20% o valoare fixă specificată în funcție de tipul de contaminant. Această metodă reprezintă o abordare statistică admisibilă până la un anumit grad de toleranță în ceea ce privește depășirea parametrilor țintă de remediere. În acest caz, punctele de monitorizare sunt considerate orientative și sunt distribuite în mod reprezentativ în întreaga zonă de interes, astfel încât atingerea parametrilor țintă de remediere să poată fi evaluată în mod obiectiv, în special în raport cu zona inițială a penei de contaminare. Rezultatele monitorizării sunt apoi prelucrate și interpretate din punct de vedere statistic. Punctele care reprezintă locuri cu valori extreme și punctele care reprezintă cea mai mare parte a spațiului zonei de interes sunt tratate diferit.

- Parametrul țintă de remediere este atins după eliminarea/stabilizarea unei anumite părți a contaminantului. Această abordare prevede o evaluare bazată pe efectuarea uni bilanț a cantității de contaminare înainte și după încheierea remedierii.
- Parametrul țintă de remediere este atins atunci când riscul de contaminare prezent pentru mediu a
  fost redus la cel mai scăzut nivel acceptabil, cu o intervenție de remediere acceptabilă și justificabilă
  din punct de vedere tehnic și economic. Această abordare permite să se pună capăt intervenției de
  remediere atunci când contaminarea reziduală nu prezintă un risc sporit pentru mediu și, în același
  timp, eliminarea completă a acesteia ar necesita o intervenție de neacceptat din punct de vedere
  tehnic și economic.

În cazul în care parametrii țintă de remediere sunt atinși, trebuie adăugate alte etape de monitorizare.

### 5.2.3 Monitorizarea post-remediere

Scopul este de a demonstra durabilitatea parametrilor țintă de remediere atinși. În acest caz, sarcina este pur specifică condițiilor din zona de interes. Sustenabilitatea parametrilor țintă atinși în urma remedierii poate fi demonstrată numai prin monitorizarea pe termen lung a unor puncte de monitorizare selectate în mod corespunzător (în puncte critice și la ieșirea apelor subterane din sit). În majoritatea locațiilor, se poate aștepta o creștere ulterioară a concentrațiilor de contaminanți monitorizați după încheierea intervenției active.

Indicatorii monitorizați în mod obișnuit sunt pH-ul, temperatura și conductivitatea în apele subterane, reactivul utilizat, contaminantul și, nu în ultimul rând, produșii de descompunere. Prelevarea de probe trebuie să se facă într-o manieră dinamică. Unii contaminanți, cum ar fi hidrocarburile clorurate, se descompun în produși de descompunere (percloro - clorură de vinil) care sunt mai toxici decât contaminantul în sine. Este posibil ca acești produși de descompunere toxici să fie remanenți.

Perioada de monitorizare trebuie să fie suficient de lungă, adesea 3-5 ani, și depinde de condițiile hidrogeologice, de mărimea sitului și de cantitatea de contaminanți din subteran. Perioada de timp necesară monitorizării trebuie să includă, de asemenea, posibilitatea unui efect de ricoșeu, adică o creștere a concentrațiilor de contaminanți după ce remedierea a fost considerată completă. De regulă, agentul oxidant reacționează cu fracția dizolvată a contaminanților din apele subterane. Cu toate acestea, sursele de contaminare secundară bazate pe fundul colectorului sub forma unei faze libere de contaminant (DNAPL – lichid dens în fază neapoasă), sau într-o zonă nesaturată de unde ajunge în colector prin spălare din cauza precipitațiilor sau chiar situată în afara sitului vor induce, după un interval de timp, o nouă creștere a concentrațiilor în apele epurate. Cea mai mare creștere ar putea apărea în cazul în care remedierea a eliminat doar parțial poluarea și în care rămâne o fază liberă în subsol. Din punct de vedere hidrogeologic, perioada de monitorizare post-sanitară ar trebui să depindă de rata de curgere și de migrație a contaminării, astfel încât întreaga zonă a penei de contaminare inițială și împrejurimile sale să fie monitorizate în primii ani după finalizarea remedierii.

#### 5.2.4 Prelucrarea analizei de risc actualizate după finalizarea remedierii

Raportul final de remediere și raportul de monitorizare post-sanitare ar putea fi urmat de o analiză de risc actualizată care va fi pregătită pe baza monitorizării continue, finale și post-remediere. Nu se preconizează

nicio altă activitate de natură tehnică în cadrul prelucrării analizei de risc actualizate. Analiza actualizată va evalua riscurile care decurg din poluarea reziduală de pe amplasament.

La aplicarea metodelor *in situ* de reducere a contaminării pot apărea și unele probleme tehnice. De exemplu, este necesar să se verifice prezența unor căi preferențiale de răspândire a reactivilor - scurgeri de utilități depozitate sub nivelul apelor subterane, care pot drena apele subterane și pot evacua soluțiile aplicate de reactivi în afara zonei de remediere. O scurgere a reactivului rezidual în stația de epurare a apelor (uzate) și, ulterior, într-un sistem de apă de suprafață poate cauza probleme precum o contaminare a sondelor adiacente cu reactivul respectiv.

### 6 CONCLUZII

ISCO este un grup de tehnologii de remediere în evoluție continuă, care include multe tipuri de oxidanți și care implică adesea o chimie complexă. ISCO ar putea fi considerată o abordare agresivă. Este adesea selectată drept tehnologie de remediere atunci când unul dintre criteriile cheie este timpul limitat. Cu toate acestea, pentru a crește eficiența și durabilitatea remedierii, ISCO trebuie să fie evaluată ca parte a unei abordări integrate, constând într-o succesiune de tehnologii. Oxidarea chimică este o tehnologie de intervenție utilizată în principal în zona saturată (apele subterane) și pentru zonele sursă, în timp ce aplicarea în partea superioară, în solul nesaturat și în mediul saturat de apă din cadrul zonelor unde pătrunde pana de poluare trebuie să fie evaluată cu atenție.

În oricare dintre cazuri, evaluarea fezabilității unei intervenții ISCO trebuie să fie efectuată ținând cont de obiectivele necesare pentru tratare, indiferent dacă este inclusă într-o intervenție care constă într-o combinație de tehnologii sau dacă ISCO este prevăzută ca activitate de sine stătătoare. Localizarea contaminantului în subsol poate oferi o primă orientare a evaluării fezabilității, dar, în scopul creșterii probabilității de succes și a eficienței unui tratament cu oxidanți chimici, trebuie luați în considerare următorii factori-cheie:

- modelarea precisă a caracteristicilor hidrogeologice, pentru a asigura distribuţia eficientă a agenţilor oxidanţi şi pentru a calcula raza de influenţă, în funcţie de eterogenitatea zonei care urmează să fie tratată;
- caracterizarea geochimică adecvată pentru a calcula consumul de oxigen de către substanțele de tratare care nu sunt vizate (cererea naturală de oxigen);
- caracterizarea 3D a contaminării asociate cu caracteristicile litostratigrafice pentru verificarea zonelor de acumulare a contaminanților și a zonelor de dispersie a acestora.
- evaluarea mai multor alternative de intervenție în faza de pre-proiectare, elaborate printr-o abordare integrată, în scopul identificării secvenței de tehnologii care să maximizeze eficiența pe parcursul întregului proces de remediere;
- efectuarea de teste de laborator și/sau teste pe teren pentru a reduce incertitudinea în faza de proiectare a intervenției;
- efectuarea de monitorizări la scară reală pentru a verifica obiectivele de remediere.

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# Annex 1

## In Situ Chemical Oxidation — Case studies

IMPEL Project no. 2020/09





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## 2. Site background

## 2.1 History of the site: Challenges and Solution

Site located in Veneto Region, Italy

A fuel tanker truck over-turned on a small road in northern Italy, spilling over 36,000L of diesel and petrol. The fuel impacted a canal, flood defences, soils and groundwater in the immediate vicinity.

The accidental event happened the 25th August 2017.

Emergency oil spill response was carried out, with impacted soils and the road surface removed and replaced. An underground pipeline was flushed out and sorbent booms were placed in the adjacent canal to catch and remove the oil.

A site investigation was completed concurrently with the oil-spill response in order to identify the subsurface contamination, build an initial Conceptual Site Model (CSM) and develop plans for remediation. MTBE, petroleum hydrocarbons (TPH) and BTEX were found to be within the soil – concentrated within the capillary fringe.

The groundwater was also found to be impacted and requiring remediation. A remedial options appraisal was completed, considering technical feasibility, sustainability, time and cost and a combined in situ chemical oxidation (ISCO) and enhanced aerobic natural attenuation (ENA) approach was chosen.

Main challenges of the site are related to:

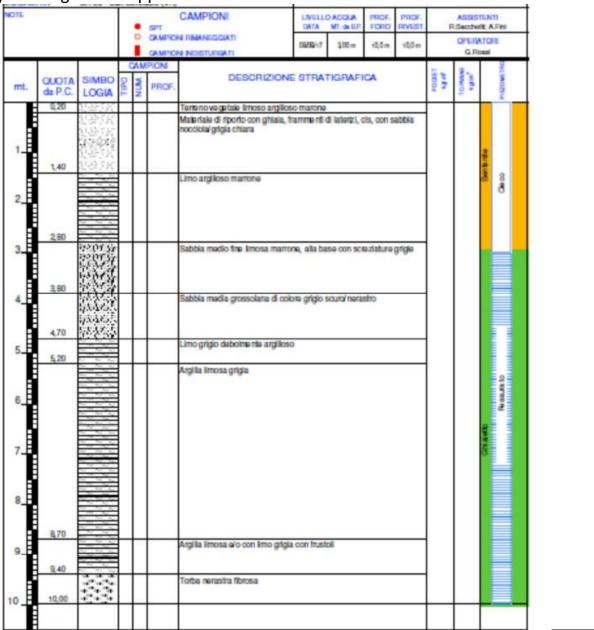
- Urgency to complete remediation and allow area to go back to original conditions
- Public areas, no services available,
- No presence of fences, no surveillance
- Presence of MTBE (highly mobile) in a recent pollution event poses risk for rapid formation of plume of big size
- Different matrices interested: vadose zone soil, soil in capillary fringe, groundwater





## 2.2 Geological and hydrogeological setting

- Intercalation of fine sands with silts
- Unconfined aquifer with groundwater table at 2.5 m BGL
- Bottom of the aquifer at 5-6 m BGL (clay)
- Unknown specific data on conductivity and porosity
- Hydraulic gradient approx 0.5%







## 2.3 Contaminants of concern

- Soil impacted with TPH and BTEX
- Groundwater impacted with MTBE and TPH
- Targets for soils: CSC residential areas:
  - o C>12: 50 mg/kg
  - o C<12: 10 mg/kg
  - o B: 0.1
  - o T: 0.5
  - o EB: 0.5
  - o X: 0.5
- Targets for groundwater: CSC:
  - $\circ$  TPH: 350  $\mu$ g/I
  - o MTBE: 40 μg/l
- Exceedings in soil in table below
- Exceedings in groundwater <1 mg/l for both TPH and MTBE

Campione	DATA	IDROCARBURI PESANTI C>12	IDROCARBURI LEGGERI C<12	BENZENE	TOLUENE	ETIL BENZENE	XILENI	
THE RESIDENCE SAME		mg/kg						
FS1 VASCA	29/08/2017	1564	41	< 0,01	0,03	0,23	3,05	
FS2 VASCA	29/08/2017	1703	52	0,06	3	1	2	
M5C3 (2-2,4m)	04/09/2017	11327	361	0,21	5,54	2,23	3,41	
M4C3 (2-2,8m)	04/09/2017	5094	39	< 0,01	0,02	0,04	0,41	
PZ7C (2,0-3,0)	07/09/2017	118	8	< 0,01	< 0,01	< 0,01	0,01	
PZ6C (2,0-3,0)	06/09/2017	246	15	< 0,01	< 0,01	< 0,01	0,03	
CSC Tab. 1 Col	lonna B	750	250	2	50	50	50	
CSC Tab. 1 Col	lonna A	50	10	0,1	0,5	0,5	0,5	





### 2.4 Regulatory framework

- In Italy, CSC values define potentially contaminated sites. These are table limits.
- You can run risk assessment to find CSR: risk based threshold values, which can be less stringent as CSC and define site specific goals
- In this case, due to the limited size of the site, risk assessment has not been performed. Therefore targets for the remediation equal the national wide table limits CSC, specified at point 2.3
- A remediation plan needs to be submitted to the competent local authorities.
- Once the remediation plan has been submitted, the Municipality needs to call a meeting for its discussion, together with other technical and administrative authorities.
- If the project is approved, the proponent needs to pay a guarantee and then can start the works within the timeframe defined in the approval

## 3. Laboratory-scale application in field

### 3.1 Laboratory scale application

- Laboratory testing was not required and has not been performed
- Lab testing is seldom required by clients or authorities in Italy, and they are rarely performed
- Lab testing rarely can be useful for scaling up on site, and frequently is not representative, as it is difficult to simulate site conditions on a lab scale.
- If needed, a field pilot test, of small size, can provide at approximately the same cost more reliable and representative information.





# 4. Pilot-scale application in field

## 4.1 Main treatment strategy

- No pilot activity has been performed in this site
- This is because of the limited size of the site, and also for necessity of arriving to closure as soon as possible
- Therefore the strategy, the dosing and the activities have been designed based on previous experience on similar sites.





## 5. Full-scale application

### 5.1 Main Reagent

- General strategy was the use of ISCO coupled with EAB on both fringe soil and groundwater
- The strategy was selected after a multicriterial analysis comparing different strategies, taking into account logistics, timing, efficiency, consolidation of the technique, costs.
- The selection has been made thanks to the fact that no installation of active plants was needed, which would have been difficult to install and maintain on a public area without surveillance, the ease of use and the minimization of site activities
- RegenOx® is the ISCO agent selected. It is a patented formulation with catalyzed sodium percarbonate. Main reasons for selecting this specific reagent have been: ease of use, it is less dangerous compared to other ISCO agents (accidental contact with workers does not cause major issues), it is perfectly compatible with any kind of material (doesn't cause corrosion), and has a Strong desorbing effect (which was used in this case). Is also perfectly compatible with ORC oxygen release compound, which made it possible to co-inject together.
- Two different ways of application, at a distance of few days: first a direct application into excavation: product applied inside the excavation using the excavator, and mixing with saturated soil and groundwater. This caused an immediate desorbing effect (thanks to desorbing properties of RegenOx®), and direct recovery of LNAPL. At the end ORC was directly applied to excavation.
- Total size of excavation: 70 m<sup>2</sup>. Dosage: RegenOx® Part A (based mainly on sodium percarbonate) 220 kg; RegenOx® Part B (catalyst, based on iron silicate): 110 kg. ORC (calcium peroxide) 125 kg.
- Secondly, application by direct push has been made in the areas surrounding the excavation. It has been co-applied again RegenOx® + ORC, in capillary fringe and groundwater.
- It has been applied on a regular grid with distance of 3 meters,
- Total of 16 injection points, treatment over a layer of 2 meters (from 2 to 4 m BGL)
- Dosage per single point: RegenOx® Part A: 18 kg; RegenOx® Part B: 18 kg; ORC-Advanced 25 kg.
- The RegenOx® has been dissolved in water, forming a solution of 380 litres per





point. Usual dilution factors used for this reagent is 4-8% of RegenOx® Part A mass in water. After complete dilution, RegenOx® Part B is added (it is already a liquid/gel)

- Immediately after this, ORC powder has been put in water, and mixed, forming a slurry, for a volume of 125 litres per point.
- All field works have been performed in 1 week time.

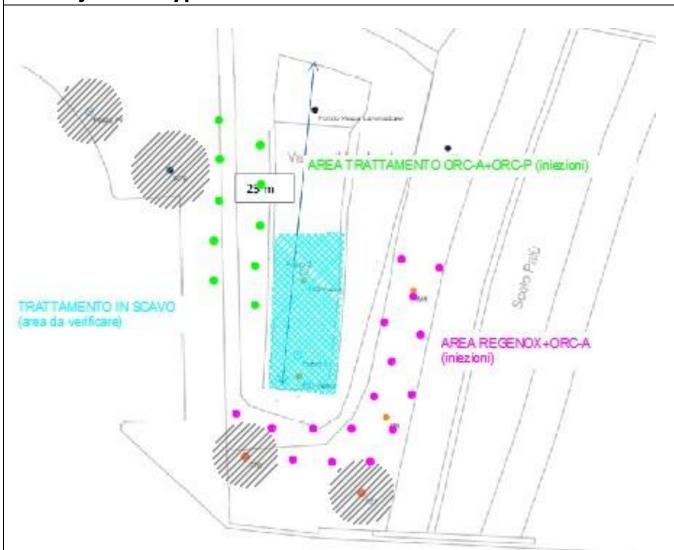
### 5.2 Additives

- RegenOx® is a bicomponent ISCO agent
- In order to make reactive the sodium percarbonate (RegenOx® Part A), it is needed to have a catalyst (RegenOx® Part B).
- Usual dosages for RegenOx® Part B range from 50% to 100% of RegenOx® Part A. In this case it has been applied 50% in the excavation and 100% in direct push
- RegenOx® Part B is a liquid/gel composed mainly by iron silicate. Once in groundwater, it creates a matrix/surface on which both the oxidizer and the contaminants are attracted. This mechanism increases the probability and the velocity of direct contact between oxidizer and contaminants





## 5.3 Injection type



- 2 ways of application: direct application into excavation and direct push injection
- For direct push, regular grid of 3 x 3 meters distance. There was no direct verification of radius of influence, but has been selected this interdistance based on experience and observance in similar sites.
- Layer from 2 to 4 m BGL. Groundwater level is approx at 2.5 meters. So this layer covers fringe soil, periodical fluctuation zone of groundwater, and the first 1.5 meters of aquifer. Not all aquifer treated, as LNAPL tend to accumulate on first part.
- Just one single injection campaign performed. This is not very common for RegenOx®, most frequently we perform 2-3 campaigns at a distance of 1 month, to manage rebound. In this case the majority of the mass was MTBE, a hydrophilic





contaminant, which doesn't sorb that much to saturated soil, so 1 campaign has been considered sufficient.

- See previous paragraphs for dosing
- No fracturing used. Has been injected at relatively low pressure (2-4 bars). High pressure fracturing can cause formation of preferential pathways and lack of treatment in areas which ISCO agent can't access.

### 5.4 Radius of influence

- No direct measurement or calculation of radius of influence on this site
- The interdistance selected was 3 meters, estimating a ROI of approx 1.7-1.8 meters, therefore allowing for some overlapping between ROI in the treatment area
- This has been selected based on experience acquired on similar sites.
- Typical interdistances used for RegenOx® range from 3 to 4-5 meters. In this case the minimum value has been used, due to the relatively low permeability of the soil

### 5.5 Process and performance monitoring

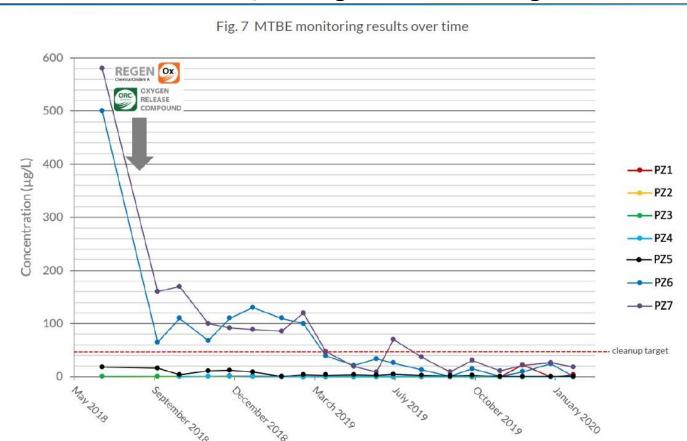
- pH, redox, dissolved oxygen, temperature have been measured on site using multiparametric survey (field measurement)
- Parameters measures once per month for a period of 2 years, the same day as groundwater sampling for contaminants of concern
- Especially pH, redox and dissolved oxygen have been helpful in understanding the ongoing of the treatment
- Also monitoring of metals included, together with contaminants of concern.
   Same frequency and duration (once per month for 2 years)
- Analyzed in laboratory
- Metals searched: iron, manganese, total chromium, chromium VI. No variations have been noted that could be related to the treatment.





## 6. Post treatment and/or Long Term Monitoring

## 6.1 Post treatment and/or Long Term Monitoring

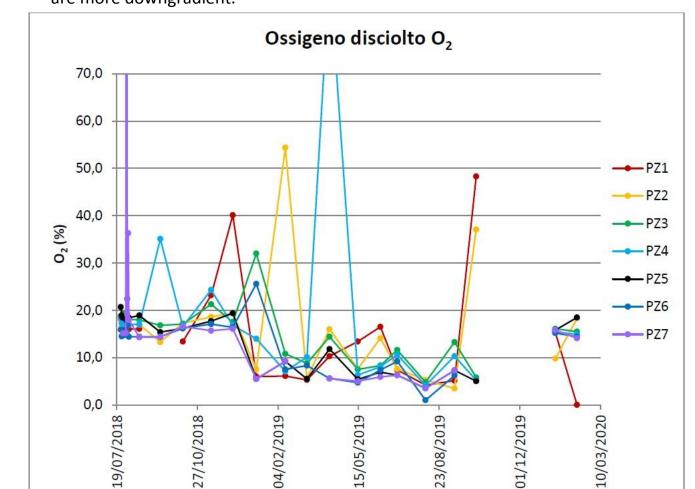


- Contaminants of concern monitored each single month for 2 years after application.
- Contaminants monitored: TPH, BTEX, MTBE, ETBE
- After the 2 years monitoring, formal compliancy of the site observed (reduction of contaminants of concern, and observance of no rebound in the following period).
   This was achieved in February 2020.
- After that, there is an additional post-operam monitoring period (still ongoing) of 2
  year, with analysis every 3 months, to confirm that no increase of concentrations is
  observed.
- Also soil in capillary fringe has been tested for compliance after treatment. This has been performed through 4 soil borings, and analysis for compliancy of CSC, which was achieved in all 4 points.
- For groundwater, main contaminant was MTBE. TPH, originally present in groundwater above CSC, was already below CSC before ISCO application, probably thanks to the primary removal of source (excavation) made as emergency measurement (MISE)





 For groundwater, Pz5 Pz6 PZ7 are the wells inside the treatment area. The others are more downgradient.



## 7. Additional information

### 7.1 Lesson learnt

- Very effective and rapid treatment. This is much faster compared to usual timing on treatment of groundwater in Italy, due to usually slow bureaucratic process
- Only 2 years between contamination event and formal achievement of compliancy of the site.
- Quick process has been achieved thanks to management of some parts in parallel (emergency activities and investigation)
- Also direct involvement and open discussion with local authorities was crucial for getting authorization on time
- Velocity of the process was crucial for not allowing formation of a bigger plume.
- Area accessibility was difficult, being present canals, tanks and private





- surrounding areas. Therefore the treatment areas have been adjusted accordingly, but this hasn't affected the treatment efficiency
- No other parameters measured apart from the ones already mentioned.

### 7.2 Additional information

- Experience is very important, and is usually acquired thanks to management of many sites
- Field pilot test is highly recommended in any case, but it could be avoided for small sites like this one
- Dosing and design can't be based only on stoichiometry. Anyway, stoichiometry needs to be based on total contaminant mass (dissolved phase, sorbed phase to soil, NAPL), and not all of them are always directly known. For example in Italy saturated soil is never analyzed, and this is where the majority of the mass usually stands. This means that the mass of contaminant can be an imprecise estimation.
- Apart from stoichiometry, other factors on which to be based are distribution of the reagent, and minimum dosage required.
- Before getting in charge for an ISCO design, it needs to be evaluated if the technology is feasible. This needs to be done taking into consideration: geology, concentrations, targets, depth, accessibility of the area.
- The selection of the specific reagent can't be based only on reactivity, but needs to take into account longevity, distribution and ease of use. There are general rules and outlines, but is preferable to make these evaluations site-specific.

### 7.3 Training need

- I think the most useful thing is to get many examples of treatments done, in order to have an idea of how an average treatment should look like
- Too many times I see treatments performed using unrealistic designs, meaning interdistance between points too wide, wrong application method (i.e. gravity feeding of wells), very low quantities of amendments. In some cases there are examples of distances that could not be considered applicable in any case.
- Workshops and webinars are probably the most effective ways for training
- Visit to some sites where application is ongoing also is a very useful instrument to have a good idea of what is being done.

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## 2. Site background

## 2.1 History of the site: Challenges and Solution

The site is a gas station, with an adjacent private property in a city of northern Italy, where at least starting from 1959, the marketing of petroleum products for motor vehicles, refuelling of motor vehicles, sale of lubricants and oil change of cars have been carried out. A contamination of TPH and BTEX affecting soil and groundwater (with also LNAPL) was found there in 2006. Thus a groundwater and unsaturated soil remediation plant was installed using MPVE technology. The project approved by the local authorities provides, where the remediation interventions through MPVE have not reached the identified remediation objectives within the set time frame, a Second Remediation Phase through the possible application of ISCO technology. So ISCO was chosen in order to remediate the presence of MTBE in groundwater outside the site.







### 2.2 Geological and hydrogeological setting

The site is located on the southern shore of Lake Maggiore, in a sub-flat area. The Quaternary deposits constituting the subsoil of the study area are characterized by fine sands and silty sands of fluvial and lake origin.

The area in question is located in an area characterized by the presence of alluvial, current fluvial and fluvio-glacial deposits with little or no surface alteration layer. The gas station area hosts a water table with an average subsidence of 3.5 m b.g.s. and outflow facing Lake Maggiore towards the east quadrant.

### 2.3 Contaminants of concern

As anticipated the historical contamination affected both soil and groundwater, with BTEX, TPH and MTBE as CoCs.

After the first phase of the remediation the groundwater samples showed the presence of MTBE, downgradient outside the site, with concentrations historically ranking up to about 1000 micrograms per liter.

## 2.4 Regulatory framework

In Italy the environmental regulatory system is regulated by Legislative Decree No. 152/06 and for fuel stations by the Ministerial Decree No. 31/15. The target value for MTBE is set equal to 40 micrograms per liter. For the implementation of ISCO technology with subsequent injections of chemical reagents in groundwater (as well as for the implementation of any remediation plan) the approval by local authorities is needed.





## 4. Pilot-scale application in field

## 4.1 Main treatment strategy

ISCO technology is a technique that involves injecting an oxidant into the subsoil to chemically treat polluting organic compounds and transform them into harmless substances.

The execution of the field test had a dual purpose: to verify the applicability of the chemical oxidative treatment against residual contaminants present in the groundwater (MTBE) and ascertain the path of the oxidizing solution in the subsoil, in order to dimension the interventions planned for the second phase of remediation.

The solution used is composed of an oxidizing complex based on sodium persulfate activated with calcium peroxide.

The chemical reactions caused by the use of this specific compound are:

- direct chemical oxidation in the short term;
- biological degradation in the long term.

Sodium persulfate breaks down in water generating persulfate anions  $(S_2O_8^{2-})$ , creating a strongly oxidizing alkaline environment.

The persulfate oxidation reactions involves the transfer of 2 electrons and is influenced by the concentration of anions, pH and oxygen.

In order for the contamination to degrade, the persulfate anion must be activated in order to generate the sulfate radical. The activated persulfate increases its oxidizing power, as the radicals are molecular fragments with an extremely reactive unpaired electron.

As for the biological action in the long term, the generic degradation of hydrocarbon compounds is the work of sulfur-reducing bacteria.

## 4.2 Additives

The activation energy of the persulfate is provided by calcium peroxide, which also has the function of regulating alkalinity (restoring a basic environment) and slowly releasing hydrogen peroxide and calcium hydroxide, with formation of hydrogen peroxide.

Hydrogen peroxide breaks down into oxygen and water, playing the role of a source of oxygen necessary for the decomposition of hydrocarbons.

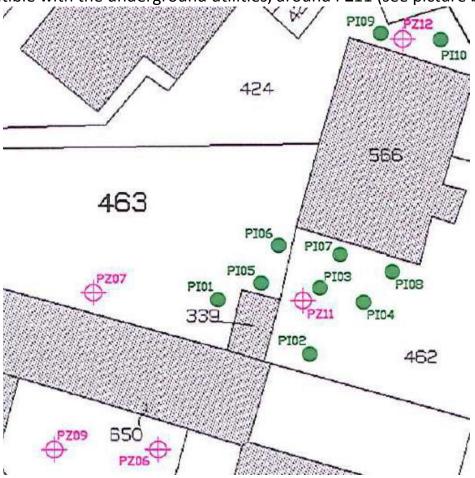
The redox potential of sodium persulfate is 2.12 V, and it is the strongest oxidant of the peroxide family.





### 4.3 Injection type

The pilot test was performed by injecting in the subsoil an oxidizing solution, consisting of the commercial product diluted to approximately 10% with water, at 10 injection points (PIO1 to PI10): 2 spaced 5 m each other near PZ12 and 8 spaced 5 m each other in a grid, compatible with the underground utilities, around PZ11 (see picture below).



The injection took place using a direct-push technique, which involved driving a 1" hollow shaft into the subsoil, from whose terminal filter tip the oxidizing solution was injected under pressure at pre-established depths.

During the test, a solution consisting of about 2.7 m<sup>3</sup> of water mixed with 300 kg of the product was introduced into the aquifer for each of the 10 injection points.

Along the 10 verticals 7 sub-injections were carried out, proceeding in ascent from the bottom upwards in steps of 1 meter, i.e. from the bottom of the hole, located about 9 m b.g.s., up to the capillary fringe, about 3 m b.g.s.

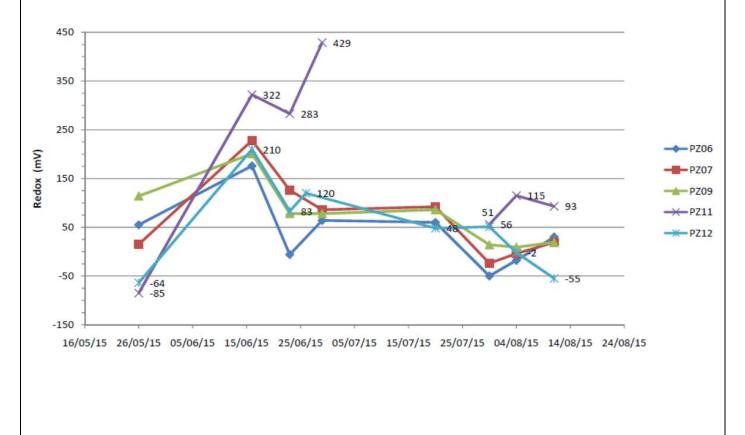
The total quantity of oxidizing solution used at the end of the test was approximately 27 m<sup>3</sup> of water and 3 tons of the commercial product.





#### 4.4 Radius of influence

The oxidant solution injected into the aquifer immediately generated positive redox potentials in all the monitoring points considered with values, gradients and longevity directly proportional to the distance between the monitoring points and the intake area, with effects observed also in PZ07 at about 10 m from the nearest injection point and in PZ06 at about 15 m from the nearest injection point (see the picture below with ORP values observed after the injection of oxidant solution). Moreover, the infiltration and drainage capacity of the oxidizing solution was not affected by the fine particle size that characterizes the subsoil in question (sandy silt and silty sand). To confirm this, in the multiple injection phase by direct-push, with a distance between the injection points of about 5 m, there were no problems of soil super saturation and it was therefore possible to inject all the quantity of oxidizing mixture expected, so such distance between the injection points was able to guarantee an overlap of ROI.







### 4.5 Control parameters

The monitoring of the chemical-physical parameters of the groundwater took place, on a network of 5 monitoring wells, with periodic frequency (approximately every 7 days), by measuring the pH and redox potential with a multiparameter probe directly in well at 3 increasing depths with respect to the free surface of the aquifer (at -1, -2 and -3 m below groundwater level), or on the ground level with field probe and flow cell for the water collected at -1 m depth compared to the free surface of the water table. For the measurements carried out with a multi-parameter probe, it was also possible to record further parameters such as temperature, electrical conductivity, dissolved oxygen (expressed in mg/l and in %) and salinity.

It should be noted that after the injections of the oxidant solution into the aquifer it was not possible to measure the oxygen parameter dissolved in the water (mg/l and %) due to the possible aggressiveness of the product towards the measurement sensor.

The measures of chemical-physical parameters took place at the following time intervals:

- T0 baseline time (13 days prior to the first campaign),
- T1 time (4 days after the first injection),
- T2 time (11 days after the first injection),
- T3 time (17 days after the first injection),
- T4 time (38 days after the first injection),
- T5 time (48 days after the first injection),
  T6 time (53 days after the first injection),
- T7 time (60 days after the first injection).

The test included the analytical determinations on the whole piezometric network involved in the test, of the following parameters:

- Benzene, Ethylbenzene, Toluene, p-Xylene,
- Total hydrocarbons (such as n-hexane),
- MTBE,
- Lead,

in the following time intervals:

- T0 baseline time (13 days prior to the first campaign),
- T3 time (17 days after the first injection),
- T4 time (38 days after the first injection),
- T7 time (60 days after the first injection).





## 5. Full-scale application

#### 5.1 Main Reagent

No changes from pilot test

#### 5.2 Additives

No changes from pilot test

## 5.3 Injection type

In detail, the injection of an activator/buffer based on calcium peroxide in the hydrogeological valley area of the site was carried out by placing a solution in the subsoil, consisting of activator diluted 10% with water, at 10 injection points, named from PIO1 to PI10. In the points where the injected reagent was absorbed with difficulty, in order to allow complete absorption of the same, i.e. in correspondence with points PIO1, PIO4 and PI10, new perforations were made as close as possible to the points of origin (i.e. PIO1bis, PIO4bis, PI10bis), see picture below. The injection took place using a direct-push technique which involved driving a 1" hollow rod into the subsoil, from whose terminal filter tip the solution was injected under pressure at predetermined depths.

During the activity, a solution consisting of 0.9 m³ of water mixed with 100 kg of activator was introduced into the aquifer for each of the 10 injection points. Along the 10 verticals 7 sub-injections were carried out, proceeding in ascent from the bottom upwards in steps of 1 meter, i.e. from the bottom of the hole, located about 9 m b.g.s., up to the capillary fringe, about 3 m b.g.s.

The injection of the oxidant solution with sodium persulfate in the PV area was performed by placing in the subsoil the solution, diluted to 7.5% with mains water, at 6 injection points, named by PIO1 to PIO6, see picture below.

The injection took place using a direct-push technique which involved driving a 1" hollow rod into the subsoil, from whose terminal filter tip the solution was injected under pressure at predetermined depths. The perforations were preceded from vacuum-digging pushed up to 1.5 m depth b.g.s. to verify the presence of any underground services.

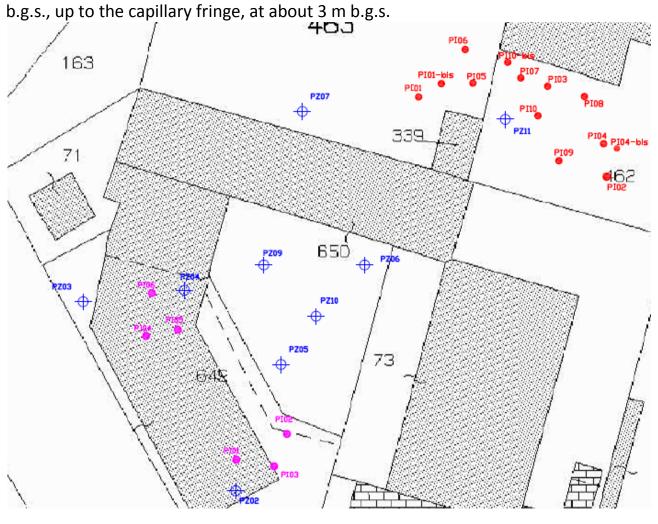
During the activity, a solution consisting of approximately 1.7 m<sup>3</sup> of water mixed with





140 kg of sodium persulfate was introduced into the aquifer for each of the 6 injection points.

Along the 6 verticals 7 sub-injections were carried out, proceeding in ascent from the bottom upwards in steps of 1 meter, i.e. from the bottom of the hole, located about 9 m



## 5.4 Radius of influence

Used the same interaxis of pilot scale.





## 5.5 Process and performance monitoring

The process monitoring of the second phase of remediation lasted more than 2 years. Here you may find the parameters, methods and frequencies.

Parameter	Method	Frequency
рН	Multiparameter probe	Twice a month for the first
		2 months, then monthly
Temperature	Multiparameter probe	Twice a month for the first
		2 months, then monthly
ORP	Multiparameter probe	Twice a month for the first
		2 months, then monthly
DO	Multiparameter probe	Twice a month for the first
		2 months, then monthly
Conductivity	Multiparameter probe	Twice a month for the first
		2 months, then monthly
Groundwater level	Interface meter	Twice a month for the first
		2 months, then monthly
BTEX, TPH, MTBE	Laboratory analysis	Twice a month for the first
		2 months, then monthly

# 6. Post treatment and/or Long Term Monitoring

### 6.1 Post treatment and/or Long Term Monitoring

Post treatment and long term monitoring parameters are the same of the process and performance monitoring parameters. The results were periodically sent to local authorities described in technical reports. The persistence of MTBE in groundwater brought to the necessity of a third phase remediation plan.





### 7. Additional information

#### 7.1 Lesson learnt

In the case study several challenges were encountered during the years. After the first phase of remediation, during which the LNAPL was recovered, the residual contamination, mainly MTBE in groundwater, was recalcitrant to the ISCO technology for several causes. Firstly, the remediation grid of injection points was located within the site property boundaries, because the surrounding private property did not allow the installation of any other remediation device. Secondly, the fine grained soil presumably in some case did not permit the reagents address properly the contamination.

#### 7.2 Additional information

The keystone issue for a successful remediation is to gain a right conceptual site model, with a proper definition, in terms of extent, soil texture and presence of preferential flow pathways of the underground contamination source, in order to find adequate technology to properly address and remediate the CoCs.

## 7.3 Training need

E-learning/webinars in order to firstly understand the theoretical fundamentals of the technology (in terms of successful design and monitoring), but especially to be shown, through case studies, all the possible problems you can deal with during projecting, applying and monitoring the technology (lessons learnt by not perfect experiences).

## **Glossary of Terms**

Term (alphabetical order)	Definition
ВТЕрХ	Benzene, Toluene, Ethylbenzene, p-Xylene
LNAPL	Light Non-Aqueous Phase Liquid
MPVE	Multi Phase Vacuum Extraction
MTBE	Methyl tert-butyl ether
TPH	Total Petroleum Hydrocarbon
VOC	Volatile organic compounds (VOCs)

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# 2. Site background

## 2.1 History of the site: Challenges and Solution

The site is a divested fuel station located in a flat area of northern Italy. The activity of the site was in the distribution of petroleum products for transport with temporary storage of the substances inside underground tanks. Site was divested and tanks removed 2 years before remediation start.

ISCO technology has been evaluated as applicable to the site due to the medium-low lithology, and the type of groundwater contamination, difficult to treat with other systems.

In this context ISCO technology could reach remediation goals faster than other technologies.

## 2.2 Geological and hydrogeological setting

Site sub-soil consists of sandy filled soil from ground level to 3 m, then sandy-silt layer from 3 to 5 m and clayey-silt from 5 to 7m b.g.l. Groundwater depth is approximately 2.5 meters below ground surface in a medium-low permeability (k = 1x10-6 m/s) and low gradient.







#### 2.3 Contaminants of concern

Groundwater samples indicated presence of Benzene (10  $\mu g/L$ ), Total Hydrocarbons (1,000  $\mu g/L$ ) and EtBE (1,000  $\mu g/L$ ) in internal area of the site, in tanks excavation area. Soil investigations after tank removal and excavation show no exceedance of regulatory limits, but the presence in saturated soil of ETBE (0.5 mg/Kg). Remediation target for groundwater were defined with Sanitary and Environmental Risk Assessment. There are no remediation targets in internal area. At site boundary (POC's) is required to reach regulator limits for groundwater. In POC's PM2 and PM7 ETBE exceed the limit of 40  $\mu g/L$ .

### 2.4 Regulatory framework

Remediation targets for groundwater were defined with Sanitary and Environmental Risk Assessment. There are no remediation targets in internal area. At site boundary (POC's) is required to reach regulator limits for groundwater. In POC's PM2 and PM7 ETBE exceed the limit of  $40~\mu g/L$ .

The scope of remediation is to reach laws regulatory? limits in groundwater at POC's and decrease CoC concentrations in internal area in order to maintain POC's compliance.

ISCO Remediation strategy was detailed in a Remediation Design Document, approved by Regulators, that included preliminary laboratory test results.





## 3. Laboratory-scale application in field

## 3.1 Laboratory scale application

Laboratory batch tests were performed in order to evaluate:

- 1) Reagent effectiveness for ETBE concentrations decreasing
- 2) Potential for heavy metals mobilization

The test samples were prepared by mixing 100 g site soil, 500 mL groundwater with ETBE concentration of 1,000  $\mu$ g/L and 1.8 g of sodium persulfate and calcium peroxide mixture. Blank samples (100 g site soil, 500 mL groundwater with ETBE concentration of 1,000  $\mu$ g/L) was prepared too.

Test results shows ETBE decreasing by 28% after 3 days, 57% after 7 days and 77% after 14 days.

CrVI (not detected in blank sample) increase to 26.8  $\mu$ g/L after 14 days. No potential for other metals mobilization was showed.





# 4. Pilot-scale application in field

### 4.1 Main treatment strategy

As described in literature, ISCO technology using persulfate activated by calcium peroxide is applicable at contamination detected in groundwater (at POC's ETBE, in internal area ETBE, Benzene and Hydrocarbons). Laboratory pilot test confirm good effectiveness of reagent for ETBE treatment.

Injections are compatible with the medium-low permeability (the mixture to inject is soluble) of the saturated matrix. Due to medium-low permeability it was decided to inject the reagent with tubes with valves (fixed manchette tubs) operating at high pressure.

Because the compliance of soil samples no other remediation and system was needed. Remediation strategy provides a first 6 month phase (pilot test) in internal area of site and a full-scale phase extended to POC's to be define after pilot test.

The product chosen for injection is a mixture with persulfate and an activator (calcium peroxide) that increase pH.

The mixture supports a two-fold mechanism for treating contaminants of concern. The reagent delivers one of the strongest chemical oxidants for short-term ISCO, and also provides electron acceptors (oxygen and sulphate) for longer-term biological oxidation. Persulfate is the strongest oxidant within the peroxides family, with an oxidation potential of 2.12 volts. As illustrated below, the direct oxidation half-cell reaction for persulfate involves a two-electron transfer:  $2S_2O_8^{2^-} + 2H^+ + 2e^- \rightarrow 2HSO_4^{--}$ 

However in most cases, rapid destruction of the contaminant of concern requires that the persulfate be activated in order to generate sulphate radicals.

Sulphate radicals are powerful oxidizing agents, with an oxidation potential of 2.6 volts. Activated persulfate is catalyzed with the peroxide and base provide by the calcium peroxide:

 $S_2O_8^{2-}$  + calcium peroxide activator  $\rightarrow 2SO_4$ •

Activated persulfate can remain available in the subsurface for months providing a combination of power and stability.

The calcium peroxide provides several benefits. First, it imparts the alkalinity and peroxide needed to activate the persulfate using activation chemistry. Second, when mixed with water it provides a long-term, slow release source of hydrogen peroxide and calcium hydroxide.

The hydrogen peroxide that is slowly formed decomposes to oxygen and water, providing an extended oxygen source for subsequent bioremediation of petroleum hydrocarbons.





### 4.2 Additives

The approach used to activate the sulphate radical was elevating the pH, using calcium peroxide.

The calcium peroxide provides several benefits. First, it imparts the alkalinity and peroxide needed to activate the persulfate using activation chemistry. Second, when mixed with water it provides a long-term, slow release source of hydrogen peroxide and calcium hydroxide.

The hydrogen peroxide that is slowly formed decomposes to oxygen and water, providing an extended oxygen source for subsequent bioremediation of petroleum hydrocarbons.

### 4.3 Injection type

Injection was executed in internal area of the site in 2 tubes equipped by valves (fixed manchette tubs) between 2.5 (groundwater level) to 5 m b.g.l. in sandy-silt layer. Injection points location was at different distance from monitoring wells (3m, 7m and 10m the nearest ones) in order to evaluate the ROI.

It was performed one injection of oxidant dosage of 175 Kg (20% water solution) for each point.

After 8 months monitoring would be start the full scale remediation.

#### 4.4 Radius of influence

Radius of influence (ROI) provided for injection points: 3 meters. It was calculated on empirical methods





### 4.5 Control parameters

The measured parameters were pH, redox potential, temperature, dissolved O₂, electrical conductivity (field instrumentation) BTEX, total Hydrocarbons, ETBE, metals (Cr, Cr VI, As, Cd, Fe, Mn, Hg, Ni, Pb, Cu, Zn) and Sulphates.

Monitoring frequency:

- 1st week all points chemical-physical parameters (with field instrumentation)
- 2nd week all points chemical-physical parameters
- after 1 month all points groundwater analysis and chemical-physical parameters
- after 2 months all points groundwater analysis and chemical-physical parameters
- after 4 months all points groundwater analysis and chemical-physical parameters
- after 6 months all points groundwater analysis and chemical-physical parameters

## 5. Full-scale application

#### 5.1 Main Reagent

With respect to the pilot test it was confirmed the reagent (mixture of sodium persulfate auto activated with calcium peroxide). The dosage was confirmed in internal area and reduced by 40% near site boundaries in order to limit temporary effects of CrVI mobilization.

#### 5.2 Additives

No changes from pilot to full scale application.





## 5.3 Injection type

1 injection campaign was performed in tubes equipped by valves between 2.5 (groundwater level) to 5 m b.g.l. in sandy-silt layer (like pilot test).

Basing on pilot test results full scale was performed using a triangular injection grid, with 4.5 m spacing. (21 injection points in a  $450 \text{ m}^2$  area). Oxidant dosage of 175 Kg (20% water solution) for each point in internal area. Dosage was reduced by 40% for each of 6 injection point near site boundary.

#### 5.4 Radius of influence

Radius of influence was calculated considering at what distance the monitoring wells were interested by injection effects during field pilot test. Pilot test ROI = 3m was confirmed.

## 5.5 Process and performance monitoring

The process monitoring is provided for 1 year.

The measured parameters are the same of pilot test: pH, redox potential, temperature, dissolved O<sub>2</sub>, electrical conductivity (field instrumentation) BTEX, total Hydrocarbons, ETBE, metals (Cr, Cr VI, As, Cd, Fe, Mn, Hg, Ni, Pb, Cu, Zn) and Sulphates.

#### Monitoring frequency:

- 1st week all points chemical-physical parameters (with field instrumentation)
- 2nd week all points chemical-physical parameters
- after 1 month all points GW analysis and chemical-physical parameters
- after 2 months all points GW analysis and chemical-physical parameters
- after 4 months all points GW analysis and chemical-physical parameters
- after 6 months all points GW analysis and chemical-physical parameters
- after 8 months all points GW analysis and chemical-physical parameters
- after 10 months all points GW analysis and chemical-physical parameters
- after 12 months all points GW analysis and chemical-physical parameters





# 6. Post treatment and/or Long Term Monitoring

### 6.1 Post treatment and/or Long Term Monitoring

No long term monitoring is provided after monitoring plan described at 5.5.

### 7. Additional information

#### 7.1 Lesson learnt

Monitoring of injection treatment show in field pilot test a first temporary phase (1 months) of CoC desorption from saturated soil and CrVI mobilization in groundwater (2-6 months) due to pH and redox increase. After that both CoC decrease and reach remediation goal and CrVI return to pre-injection level.

It was possible to define these effects both spatially and temporally due to the presence of a dense network of monitoring wells and frequent control campaigns.

The experience gained during pilot test was fundamental for the design of the full scale phase. Due to the precise technical information described, Regulators have approved the full-scale remediation without any prescription.

#### 7.2 Additional information

The injection points and monitoring wells were drilled with continuous core drilling. It can allow to verify in the field the presence of layer with higher contamination, and for consequence is possible to evaluate increasing oxidant dosage in these levels.

## **Glossary of Terms**

Term (alphabetical order)	Definition		
CoC	Contaminant of Concern		
ROI	Radius of influence		

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## 2. Site background

### 2.1 History of the site: Challenges and Solution

The site is located at the heart of Graz, Styria. From 1946 onward the area had various usages (Dyeing workshop, benzene laundry). Starting in 1958 Tetrachlorethylene was used in chemical laundry on site. For various reasons this TCE intruded into the subsoil, causing contamination on the site and neighbouring public space.

The planned remediation scheme consisted of an excavation with offsite treatment and horizontal well systems to treat contaminated groundwater in public space. After the remediation a residential building is planned.

The lateral support and the remediation of a contaminated subsoil zone below an existing building proved to be challenging. The first mainly due to constraints on available space, making the usage of larger drilling rigs for bored piles impossible. The later because excavation was not possible. HaloCrete® (HC) – an adaption of the jet grouting\*1 technique for in situ remediation – was used as a solution to both problems.

\*1 Jet grouting is a technique were a high-pressure jet – originating perpendicular from a rotating drilling rod - erodes soil material. The jet normally consists of a cement/water slurry. During retraction of the drilling rod this leads to the formation of columns in the subsoil. Working parameters are defined to securely achieve predefined diameters.

Normally this technique is used for underpinnings or lateral support works in geotechnics.

## 2.2 Geological and hydrogeological setting

The geological situation can be described (simplified) in the following way: Manmade fills of various thickness ( $^{\sim}3m$ ) lie over a horizon of fine sands. Below that, the aquifer consisted of sandy, silty gravels. At approx. 7m bgl silts constitute the aquiclude. Groundwater table can be found at around 6m bgl, with a gradient of 0,8%. Permeability was estimated to be around  $5x10^{-4}m/s$  for the gravels.





#### 2.3 Contaminants of concern

Tetrachlorethylene was found to be the main contaminant. Concentration data was given by the environmental planner, with highest concentrations of 14000 mg/m<sup>3</sup> found below the installation site of the washing machines.

Residual PCE in phase was deemed possible.

Most of the ISCO measures were conducted in a zone of approx. 3000 mg/m<sup>3</sup>

## 2.4 Regulatory framework

No special approval was needed.

As the ISCO operation was only a comparatively small part of the remediation no special target values were given. Lacking exact (on spot), in-situ measured concentrations it was agreed to analyse the columns for their content of TCE and compare it with estimated concentrations.

I'm not aware of the specific regulatory framework in place (federal country ("Bundesland") specific) and defined target values. These topics were taken care of by the overall planner.





## 3. Laboratory-scale application in field

## 3.1 Laboratory scale application

Due to time constraints – we've only been involved late in the project – we could only conduct batch tests together with our partners from the AIT (Austrian Institute of Technology).

We analysed for NOD of soil as well as two prospective geotechnical binders (ordinary Portland cements) needed for statical reasons. The soil samples were taken from different depth levels.

As oxidizing agents  $KMnO_4$  and  $NaMnO_4$  were tested, mainly for handling considerations (powder vs. liquid). Hereby no significant difference was observed after 24h. These tests were conducted on simulated column material, i.e. contaminated (site) soil samples + cement + oxidising agent

A target concentration of 20gKMnO<sub>4</sub>/kg column was recommended. This was based on the assumption of residual phase on site. In later discussions with the planner this value was reduced taking into considerations local variances and homogenization effects during the jet grouting process.

## 4. Pilot-scale application in field

### 4.1 Main treatment strategy

For this project no pilot-scale application was conducted. The feasibility of jet grouting had been proven in a research project ("HaloCrete" partly funded by the Austrian authorities)

HaloCrete was selected because it solved both structural (lateral support of excavation) and remediation (below buildings) challenges. KMnO<sub>4</sub> was then selected because it can be easily introduced into the overall jet grouting process. It was added at the mixing plant for the cement slurry in granular form. From there operations were conducted as usual.

The only difference to standard applications was the accumulation of two different backflow slurries. One being from uncontaminated soil zones and the other from contaminated zones containing KMnO<sub>4</sub>.

Works were planned to be finished after four weeks.





# 5. Full-scale application

## 5.1 Main Reagent

KMnO<sub>4</sub>, no changes to lab test



A plan view of the site. Bright red indicates sources of contamination (sewer, washing machines), red and yellow circles are planned HaloCrete columns. Column spacing was 1.2m to secure statical required overlapping, column diameter 1.5m. Drilling depths of up to 9m bgl







A picture of the site. Works are conducted below former washing machines. Backflow deeply covered purple by  $\mathsf{KMnO}_4$ 

## 5.4 Radius of influence

Due to the jet grouting installation process, the radius was "pre-defined" and measured/controlled in-situ.





### 5.5 Process and performance monitoring

Apart from standard quality assurance (for jet grouting applications) no additional controls were required.

Control and monitoring of chemical parameters were not in the scope of Kellers work. The final proof of success on ISCO works was a direct TCE-concentration measurement on samples taken from core drillings at different depths.

### 7.3 Training need

This relatively new approach of using jet grouting as a means of delivery for ICSO reagents must be made more public in general.

Taking various boundary conditions into consideration it can be a feasible and economic approach for in-situ remediation.

What comes to mind are otherwise deep excavations in need of lateral support, source zones difficult to address with conventional injection techniques and synergistic effects with construction requirements. HaloCrete columns can be used statically like any other jet grouting body.

#### 7.4 Additional remarks

I'm aware that this project differs widely from "ordinary" ISCO project, especially as ISCO was only part of a combined solution. Insofar I couldn't give an answer to every question in this survey as not all of them are applicable to our approach.

Nonetheless I hope that this contribution widens the perspective on techniques and possibilities already available for ISCO (or ISCR) applications.

# **Glossary of Terms**

Term (alphabetical order)	Definition		
m bgl	m below ground level		

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## 2. Site background

## 2.1 History of the site: Challenges and Solution

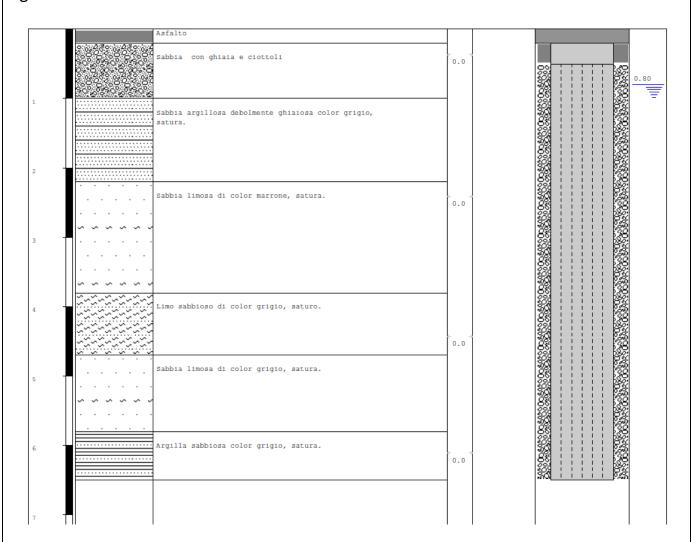
The site was a fuel retail site until 2015, its area is  $1500 \text{ m}^2$  and it is located at 20 m above sea level in Northern Italy. Since 2015 it has been a parking area. MTBE contamination was detected during the preliminary environmental investigation carried out in order to prepare the complete demolition of the plant. It was hypothesized an oil leakage from tanks and/or from lines during the selling activities. ISCO technology was selected in order to manage the residual contamination. At first, in fact, the contamination was treated with EAB filter socks as emergency measures and with EAB product injection as per RAP. The planned RAP second injection was made with ISCO and not with EAB product, as assumed before, just to obtain a more effective contamination reduction and to close the environmental case. ISCO technology was selected in order to remediate groundwater and facing the difficulty in reaching the tight legislative target of 40 µg/l for MTBE.





## 2.2 Geological and hydrogeological setting

The site is characterized by alluvial plain sedimentation: silty-sand (see below "sabbia") with clay-silt lenses, 0.5 to 1 m thick (see below "limo" and "argilla"). The groundwater level varies from 0.80 to 1.5 m bgl. The maximum depth reached by the drilling is 6.5 m bgl.



The local groundwater flow direction is W-E, the gradient 0.01, the hydraulic conductivity is medium (about  $10^{-5}$  m/s. This value is just an estimate, hydraulic tests were not performed)





## 2.3 Contaminants of concern

The maximum concentration measured during the planned monitoring for groundwater and for soil are displayed in the following table. These concentrations have been used as input parameters for the remediation design

CONTAMINANTS OF CONCERN (COCs)						
	GW	Soil				
Constituent	<u>(mg/L)</u>	<u>(mg/kg)</u>				
MTBE	1.45	0.087				
DRO	1	43.25				

According to the historical data set, there are three monitoring wells with exceedances, all the other have total hydrocarbon and MTBE under the law limits. Here below the concentrations measured in the period from 2016 to 2017.

Denominazione		28/07/2016	14/11/2016	23/12/2016	30/01/2017	23/02/2017	06/04/2017	23/05/2017	DLgs 152/06 All 5 Tab 2
Parametro	U.M.								
Piombo	μg/L	< 0.1							0.1
COMPOSTI ORGANICI AROMATICI									
Benzene	μg/L	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1	1
Etilbenzene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1	50
Stirene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1	25
Toluene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1	15
p-Xilene	μg/L	< 1	< 1	< 1	94	< 1	< 1	< 1	10
ALTRE SOSTANZE									
Idrocarburi totali (n-esano)	μg/L	171	101	39	1068	30	43	42	350
MTBE (Metilterzbutiletere)	μg/L	<u>172</u>	<u>52.5</u>	<u>159</u>	<u>154</u>	<u>72.7</u>	<u>73.6</u>	<u>83.6</u>	40*
ETBE (Etilterzbutiletere)	μg/L	-	8.1	6	13.3	3.8	8.7	5.9	40*
Piombo tetraetile	μg/L	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01	0.1*

D	Π	05/08/2016	14/11/2016	23/12/2016	30/01/2017	23/02/2017	06/04/2017	23/05/2017
Denominazione	_	05/08/2016	14/11/2016	23/12/2016	30/01/2017	23/02/2017	06/04/2017	23/03/2017
Parametro	U. M.							
Piombo	μg/L							
COMPOSTI ORGANICI AROMATICI								
Benzene	μg/L	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1
Etilbenzene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
Stirene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
Toluene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
p-Xilene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
ALTRE SOSTANZE								
Idrocarburi totali (n-esano)	μg/L	< 30	43	45	< 30	< 30	< 30	< 30
MTBE (Metilterzbutiletere)	μg/L	<u>1057</u>	<u>375</u>	<u>516</u>	<u>294</u>	<u>1310</u>	33	<u>1195</u>
ETBE (Etilterzbutiletere)	μg/L	-	9.6	9.2	14	31.1	10.5	13.1
Piombo tetraetile	μg/L	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01





Denominazione		28/07/2016	14/11/2016	23/12/2016	30/01/2017	23/02/2017	06/04/2017	23/05/2017
Parametro	U.M.							
Piombo	μg/L	< 0.1						
COMPOSTI ORGANICI AROMATICI								
Benzene	μg/L	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1	< 0,1
Etilbenzene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
Stirene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
Toluene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
p-Xilene	μg/L	< 1	< 1	< 1	< 1	< 1	< 1	< 1
ALTRE SOSTANZE								
Idrocarburi totali (n-esano)	μg/L	< 30	< 30	50	< 30	< 30	< 30	< 30
MTBE (Metilterzbutiletere)	μg/L	<u>80.6</u>	48.8	18.4	< 0,5	< 0,5	32.1	80.7
ETBE (Etilterzbutiletere)	μg/L	-	4.9	1.5	3.5	2.6	2.6	6.6
Piombo tetraetile	μg/L	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01	< 0,01

The clean-up goals are 40  $\mu$ g/I for MTBE and 350  $\mu$ g/I for total hydrocarbon expressed as n-hexane into groundwater.

## 2.4 Regulatory framework

In Italy, according to the D.Lgs 152/06 and DM 31/15, when a potential contamination is assumed or detected the site becomes a contaminated site and the owner or the "involved subject" has to inform the public authorities. The site must be characterized in order to define the conceptual model of the contamination. Then a risk analysis could be carried out in order to define site specific concentration limits. If the concentrations are above the concentration limits, defined by law or by site specific analyses, the site has to undergo a remediation.

In order to apply a chemical product in the ground the Public Authorities have to approve the Remedial Action Plan. In the case shown here the Authorities also allow to put filter socks as "emergency plan" stage and not as RAP, is it not common in all the Italian territories. Most regions allow the use of chemical compounds only under a RAP approval.

Here below the site history of the site related to the regulatory framework.

- May, 2015: execution of the preliminary investigation for the decommissioning of the fuel retail station
- June, 2015: transmission of the notification according to Dlgs.152/06 and D.M. 31/15;
- July, 2015: decommissioning of the plant and starting of emergency activities (removal of the portion of soil surrounding the removed tanks, purging of water from the excavation and soil sampling from the walls and bottom of the





#### excavations);

- November, 2016: installation of socks for EAB
- April, 2017: replacement of socks for EAB
- July, 2017:RAP transmission
- December, 2017: PA approval of RAP.
- April, 2018: EAB product injection
- April, 2019: ISCO injection
- January, 2020: first of 4 planned quarterly groundwater sampling tested with PA in order to define the groundwater not contaminated
- June, 2020: groundwater sampling tested with PA
- September, 2020: groundwater sampling tested with PA
- November, 2020: groundwater sampling tested with PA
- December, 2020: execution of soil testing surveys in order to define the soil as not contaminated soil for all the site.

## 3. Laboratory-scale application in field

## 3.1 Laboratory scale application

No laboratory scale application was done. The oxidant demand was calculated from site condition parameters such as lithology, contaminant concentrations, fraction of organic carbon, hydraulic conductivity, volumes of groundwater and soil to be treated. The calculation was made with a stoichiometric approach.

## 4. Pilot-scale application in field

### 4.1 Main treatment strategy

The RAP considered two injection campaigns: the first was carried out with EAB product, the second with ISCO product. No pilot test was conducted onsite considering the very small area of the contaminated site (1500 m<sup>2</sup>). The second injection was sized based on the result of the first injection activity.





## 5. Full-scale application

## 5.1 Main Reagent

- The first treatment application started in April 2018 and consisted of the injection of EAB product through 8 direct push points. The selected product is a specially formulated time-released grade of calcium peroxide designed to assist in the aerobic bioremediation in soil and groundwater. A volume of 600 liter of slurry, prepared with water in a concentration of 25%, was injected into the subsurface through each direct push point. Totally, 1200 kg of dry powder of product were used.
- The second treatment application started in July 2019 and consisted of the injection of ISCO product through 8 direct push points. The selected product is a single, formulated product consisting of high pH-activated persulfate and calcium peroxide. A volume of 600 liter of slurry, prepared with water in a concentration of 25% was injected in the subsurface through each direct push point. Totally, 1800 kg of dry powder of the selected product were used.
- The amount of applied reagent was calculated based on a stoichiometric approach

#### 5.2 Additives

The ISCO selected product is formulated to provide a self-activated persulfate oxidation system, therefore no additives were used beside the main reagent.

#### 5.3 Injection type

Eight direct-push injection points to treat from 1 to 6 m bgl. The injection was done from top to down for each 0.5 m interval. In some intervals, it was difficult to inject all the reagent as planned, so the string was shifted to just below the interval in order to complete the injection.



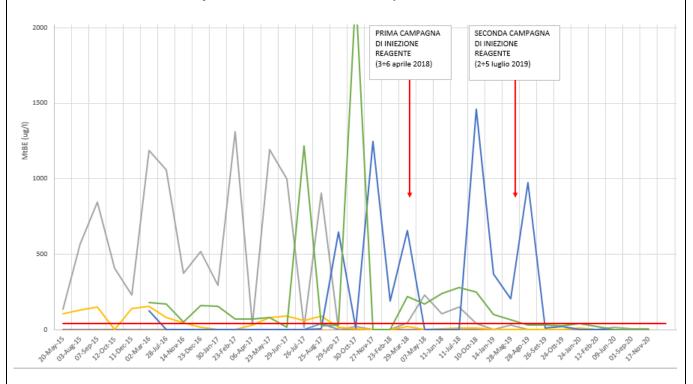


#### 5.4 Radius of influence

The radius of influence was estimated to be not less than 2 m, based on lithologies and injection method.

## 5.5 Process and performance monitoring

- Monthly monitoring for the first 3 months: MTBE (lab analysis) and chemicalphysical parameters (measured onsite);
- Quarterly monitoring with extended analytical set: Total hydrocarbons, Benzene, Ethylbenzene, Toluene, Xylenes, MTBE, ETBE (lab analysis) and chemical-physical parameters (measured onsite).
- In the graph below there are the evolution of MTBE concentration during time.
   The two red arrows indicate the first EAB injection carried out on April 2018 and the second ISCO injection carried out on July 2019







## 6. Post treatment and/or Long Term Monitoring

### 6.1 Post treatment and/or Long Term Monitoring

After 1 year from the ISCO injection job: quarterly monitoring with extended analytical set: Total hydrocarbons, Benzene, Ethylbenzene, Toluene, Xylenes, MTBE, ETBE (lab analysis) and chemical-physical parameters (measured onsite). After 4 monitoring campaigns without exceedances it will be possible to close the environmental case. (these conditions are case-specific and defined by the PA)

#### 7. Additional information

#### 7.1 Lesson learnt

The case study can be defined as a case of success since the goal of reducing the contamination below the threshold limits has been achieved and it will soon be possible to request closure of the environmental case. However, it is possible to make some considerations. The sending of the RAP to the authorities could have been done more quickly but, above all, the choice of an ISCO+EAB products since the first injection would have potentially allowed compliance to be achieved more quickly. This hypothesis could has been verified by laboratory or field tests.

### 7.2 Additional information

Given the modest size of the site and the concentrations of contaminants, the choice to implement the remediation by injection of reagents has been successful performed in a relatively short time and has been shown to be relatively sustainable.

### 7.3 Training need

- It would be useful to have an e-learning training on these aspects: proper design of the remediation; use of laboratory and field tests and use of indicators to verify the progress of the remediation (taking into account not only chemical analysis).
- In addition, it may be useful to analyze and discuss case studies through workshops.





• It would be useful if this training were not provided only by reagent producers, even though they have produced a great deal of research and studies in the field, but rather by a synergic team containing various interests: the need to improve remediation products, to remediate effectively and quickly, and to be able to propose remediation that is effectively and well accepted by the PA.

#### 7.4 Additional remarks

Really consider reagent injections remediation technology as a robust alternative to remediation plant technologies.

## **Glossary of Terms**

Term (alphabetical order)	Definition
EAB	Enhanced Aerobic Bioremediation
PA	Public Authority
RAP	Remedial Action Plan according to the Italian law
	"Progetto Operativo di Bonifica - POB"

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	<sup>3</sup> Austria
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	<sup>3</sup> Evonik
1.4 Position	<sup>1</sup> Technical lead – environmental fracturing
	<sup>2</sup> Authorized technical representative Italy and Spain
	<sup>3</sup> Business Development Manager EMEA
	Business Development Manager LIVILA
1.5 Duties	<sup>1</sup> Project engineer for the design and implementation
1.5 Duties	of innovative, <i>in situ</i> remediation techniques and
	·
	enhancement technologies
	Responsible for high-level collaboration with
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	owners, regulators and the academic community
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	±42 0041003000





## 2. Site background

## 2.1 History of the site: Challenges and Solution

The subject site is situated near Frankfurt am Main, Germany on the grounds of a former chemical manufacturing facility which produced solvents for metalwork, cleaning chemicals, and specialty oils. Other facilities of environmental concern on the property included a former oils and chemicals storage building, as well as an underground storage tank and pipeline for the storage of industrial solvents.

The plant was in operation from the mid- 1960s until a fire destroyed it, causing the plant to cease operations in 1974. It is suspected that the fire and resulting explosion was a major factor in the release of contaminants to the subsurface environment. The property was subsequently acquired in 1985 by new owners who used the site for manufacture of industrial presses until 2014. Since then, the property is used for general warehouse storage, parking lot, and auto mechanic shop.



Site of former chemical manufacturing facility in Hessen, Germany

Significant challenges to the implementation of remedial measures at the site were the massive impacts of co-mingled contaminants of concern to underlying soils and groundwater including

- Chlorinated aliphatic hydrocarbons, primarily cis-Dichloroethylene (cDCE)
- Aromatic petroleum hydrocarbons (BTEX), including trimethybenzene (TMB)
- Aliphatic total petroleum hydrocarbons (TPH)
- Trace amounts of polycyclic aromatic hydrocarbons (PAHs)
- Free- phase oil at one location

Other challenges at the site included:





- Deep contaminant impacts
- Site constraints: nearby plant buildings; underground tank and pipeline facilities; small stream downgradient of contamination (within 50 m)
- Unfavourable geology for conventional in situ remedial technologies
- Need for developing feasible site-specific remediation criteria
- Negotiated allocation of clean up costs among responsible parties
- Remedial costs

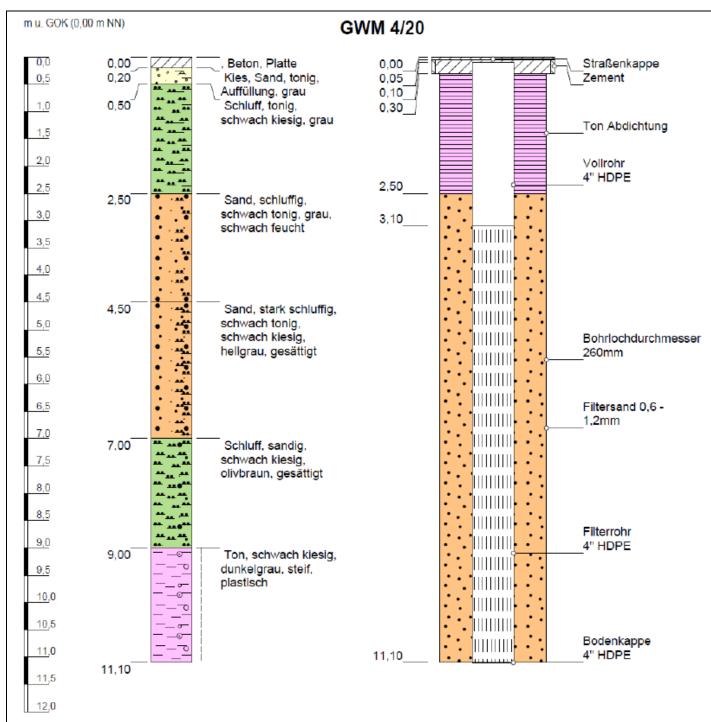
A technology was sought by the site owner and environmental consultant which could cost-effectively mitigate subsurface contamination within the site- specific constraints and limitations mentioned herein.

## 2.2 Geological and hydrogeological setting

The area of investigation consists of a surface layer of concrete which is underlain by gravel and sand fill to a depth of 1,3 m below the ground surface (bgs). Underlying the fill soils are quarternary deposits of gravel and sand colluvium of variable thickness, interbedded with sand and clay layers. Silty clays are encountered below the colluviums between depths of 3,6 to 8,3 m bgs which forms a hydraulic boundary between the overlying quarternary colluvial aquifer and an underlying tertiary (drinking water) aquifer comprising fine to medium sands. The depth to groundwater ranges from 2 to 3 m bgs.



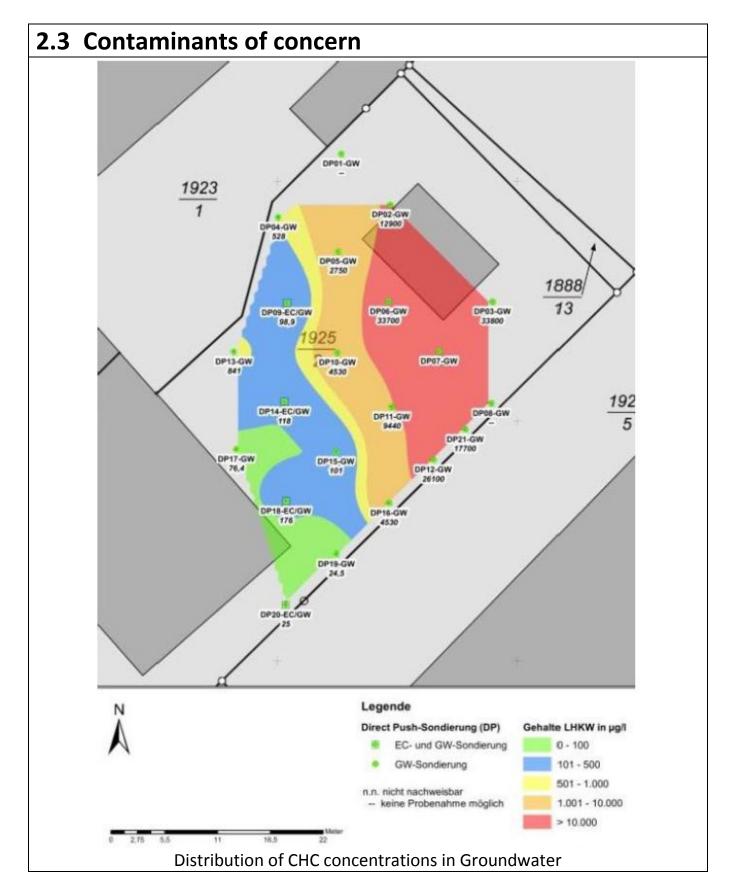




The results of pump testing conducted over 72 hours in the upper aquifer sediments determined an average hydraulic conductivity of  $1.3 \times 10^{-6}$  m/s indicating an aquifer of marginal yield (2L/min), due to the presence of significant silt and clay fines within the aquifer matrix. The direction of groundwater flow is to the south-southwest with a hydraulic gradient of approximately 5%.











A total of 6 subsurface investigations were conducted between 1999 and 2017 in an effort to delineate and quantify the distribution of contaminants underlying the site. The results of these investigations determined that petroleum hydrocarbon contamination (TPH and BTEX impacts) were largely confined within soils in the unsaturated zone with contaminant concentrations upwards to 5,000 mg/kg and 344 mg/kg respectively. Dissolved- phase contaminant impacts to groundwater within the quaternary aquifer consisted primarily of total chlorinated aliphatic hydrocarbons (CHCs) of upwards to 44,300  $\mu$ g/L, followed by TPH (2,000  $\mu$ g/L) and BTEX (1,800  $\mu$ g/L).

The major component of CHC contamination was cis-1,2 DCE (54%), followed by tetrachlorethylene ("PCE" 28%), and trichloroethylene ("TCE" 16%). The major component of BTEX contamination was trimethylbenzene (TMB >76%) followed by xylenes.

Free-phase oil product was detected at one monitoring well location with an apparent thickness of a few cm.

Calculations to estimate the mass of contaminants present within the quaternary aquifer indicated a total of approximately 3.7 kg of dissolved phase CHCs and 8.7 kg of sorbed phase CHCs respectively. The estimated total of BTEX and TPH contaminants (dissolved and sorbed) was approximately 2.5 kg. Applicable groundwater regulatory limits for contaminants of concern found in groundwater at the site are summarized below:

CHCs: 20 μg/L
 VC: 0.5 μg/L
 BTEX: 20 μg/L
 TMB: 1 μg/L
 TPHs: 100 μg/L

The delineation of the various contaminants of concern was achieved using a combination of soil probe borings, drilling and sampling of groundwater monitoring wells, and through the use of innovative Direct Push technologies using Geoprobe® drilling equipment and specialized sampling technology such as Membrane Interface Probe (MIP), Screen Point groundwater sampling, and Electrical Conductivity (EC) downhole tools.





# 2.4 Regulatory framework

Based upon the results of subsurface contamination quantified at the site, the regional environmental regulatory authority ordered that soil and groundwater remediation efforts be implemented at the site to mitigate contaminant impacts on potential environmental receptors. The specific goal of the regulatory clean up order was to "prevent the danger of contaminant exposure to receptors and prevent the long term spreading of contaminants". In order to achieve this goal, the regulation requires that "applicable remedial measures be applied to minimize or remove contaminants (i.e decontamination) and to prevent or minimize the spread of contaminants i.e. (containment)".

A Remediation Action Plan was subsequently requested by the authority to comply with the above mentioned regulatory requirements. The remedial plan submitted to the authority proposed remediation of the heavily impacted unsatured zone soils by excavation and disposal, resulting in the removal of approximately 300 m³ of contaminated soil to a depth of 2 m to 3m bgs. This remedial measure was implemented concurrently with the decommissioning and removal of the existing oil and chemical storage building on the property. There were no specific contaminant clean up criteria for soil quality required for the excavation of impacted surface soils. For the remediation of dissolved phase contaminants in the unsaturated zone, a feasibility study for the implementation of in situ chemical oxidation (ISCO) and in situ

feasibility study for the implementation of in situ chemical oxidation (ISCO) and in situ bioremediation (ISBR) was proposed as possible cost-effective and sustainable remediation alternatives to conventional excavation/disposal and large diameter soil replacement borings that were being considered. The results of the study determined that ISCO was a viable approach, although its effectiveness for practical purposes could be severely limited based upon the low hydraulic conductivity of the saturated zone sediments. To overcome this limitation, the authority approved the application of "environmental fracturing" using Targeted Solids Emplacement (TSE®) technology by Sensatec GmbH as the preferred means of distributing solid-phase oxidants as slurry into the impacted aquifer sediments.

Risk-based remediation criteria were developed for CHC contaminants at the site whereby a reduction of total CHC concentrations (i.e for PCE, TCE, DCE and VC) of 95% over 3 consecutive monitoring events in source area monitoring wells was required.





# 3. Laboratory-scale application in field

# 3.1 Laboratory scale application

A laboratory feasibility study was conducted by Sensatec GmbH at its facilities in Kiel, Germany, to compare the efficacy of ISCO and aerobic/anaerobic ISBR to degrade concentrations of total CHC and BTEX contaminants in soil and groundwater samples obtained from the site.

The scope of the laboratory work consisted of:

#### ISCO:

- Characterization of ISCO relevant parameters TIC, TOC, metals, pH and EC;
- Determination of Soil Oxidant Demand;
- Determination of reaction kinetics and oxidant demand;

#### ISBR:

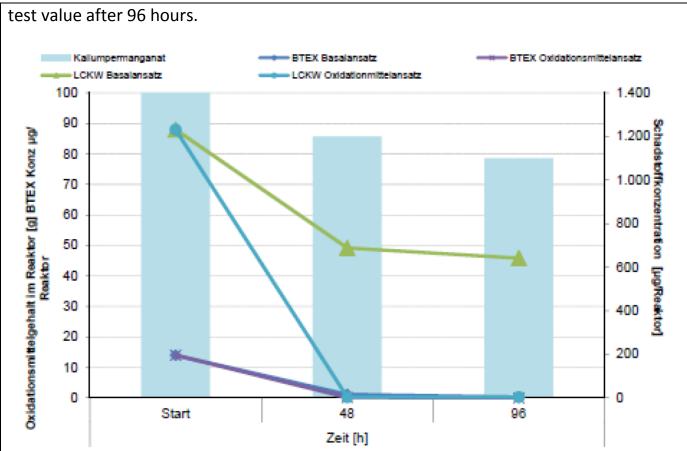
- Determination of site-specific micro-biological nutrient deficiencies and requirements (N,P,K,S,C);
- Conducting substrate induction tests to identify cosubstrate utilization profiles for various carbon based substrates at differing concentrations;
- Contaminant degradation testing using 5 varieties of substrates to enhance anaerobic and aerobic attenuation rates.

The results of laboratory analyses for ISCO determined that the impacted sediments contained very little natural organic matter (foc = < 0.001) compared to inorganic carbon (0,0038 g/g) due to high levels of oxidizable iron. The corresponding soil oxidant demand was determined by 96 hour batch testing to be 14 g oxidant/kg soil matrix which is classified as low oxidant demand (Oppermann, 2013), thereby indicating that ISCO was a viable remedial option for the site.

Of the three candidate oxidants considered in the laboratory feasibility analysis, potassium permanganate and activated persulfate oxidants showed the greatest destruction efficiency (contaminant mass removal/oxidant consumption) of CHC and BTEX contaminants compared to Fentons reagent, which exhibited the greatest oxidant consumption and shortest longevity (94% reduction within 48 hrs). The results demonstrated that Fentons Reagent was a less efficient oxidant compared to permanganate or persulfate (comparitive efficiency of 25%) due to its non-selective oxidation of metals and NOM, and fast kinetics, which result in the rapid depletion of oxidation potential and short remedial duration. Fentons reagent also exhibited the largest decrease in pH over the course of the test (from 7,1 to 2,7), whereas potassium permanganate exhibited the slightest decline (from 7,1 to 6,6) and returned to its pre-





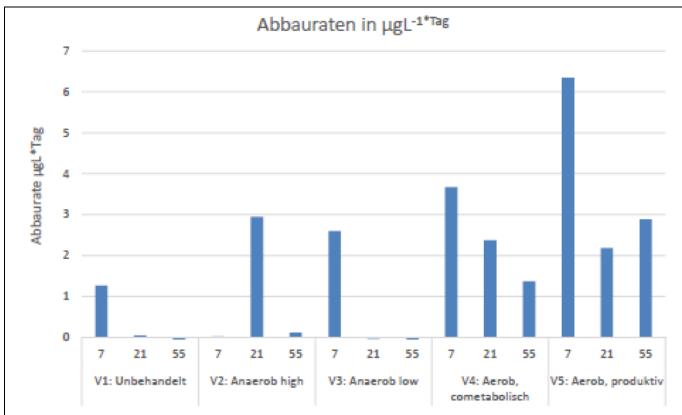


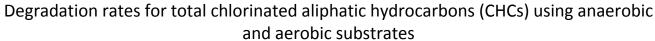
Oxidant consumption (KMnO<sub>4</sub>) and contaminant reduction (CHC and BTEX) during 96 hr reactor test

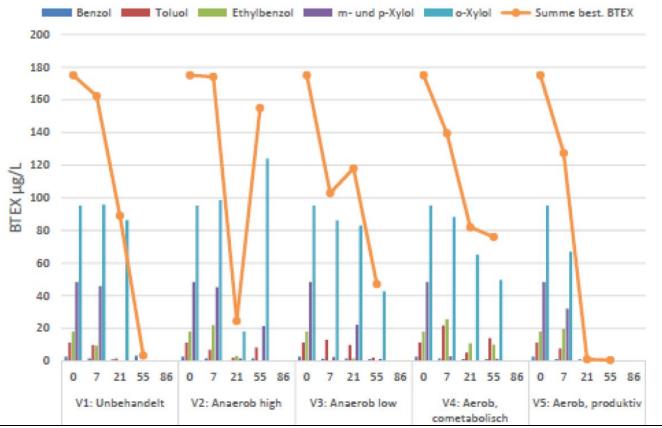
The results of laboratory analyses for ISBR determined that nutrient addition to site groundwater samples did not appreciably increase aerobic respiration rates, thereby indicating that there were no deficiencies of ambient NPKS nutrient concentrations existant at the site for aerobic biodegradation to take place. Investigations into the efficacy of aerobic and anaerobic cosubtrates ("cosubstrate screening") were conducted to determine oxygen consumption and redox potential, respectively. This was done to assess the performance of 4 candidate aerobic cosubstrate and 5 anaerobic cosubstrates being considered. The results of cosubstrate screening indicated that two anaerobic substrates (molasses, vegetable oil) two aerobic substrates (hydrogen peroxide, methanol), and an untreated reference standard, be selected for further testing on contaminants to determine bioremediation performance over a period of 55 days. In addition, qPCR gene testing was carried out on the anaerobic substrates to assess whether the gene copy count of dehalogenase bvcA and dehalogenase bvrA enzymes increased in the presence of the substrate or not. The testing was carried out to investigate the relative biodegradation potential of each of the substrates for mitigating both total CHC concentrations and BTEX concentrations, the results of which are indicated in the graphics below.















# Degradation rates for aromatic petroleum hydrocarbons (BTEX) using anaerobic and aerobic substrates

The conclusions derived from the laboratory feasibility study are summarized as follows:

- Elevated background respiration rate of 7 mg/L/day O<sub>2</sub> within aquifer samples are indicative of strongly aerobic conditions.
- The most effective cosubstrates for degrading CHCs and BTEX contaminants were molasses (anaerobic cosubstrate) and methanol (aerobic substrate) at a concentration of 1000 mg/L respectively.
- Low concentrations of dehalogenase enzymes (< 2x10<sup>3</sup>) as measured by qPCR analysis suggests that ambient populations of dehalococcoidis within the aquifer may be inadequate to stimulate anaerobic dechlorination without additional bioaugmentation.
- The greatest biodegradation observed for CHCs and BTEX was by the aerobic process through the addition of oxygen (for this study, hydrogen peroxide).
- ISCO was recommended for the full scale remediation of the site for CHCs and BTEX





# 4. Pilot-scale application in field

# 4.1 Main treatment strategy

Based on the results of a comprehensive site-specific laboratory feasibility study to assess the efficacy of various in situ approaches (see previous section 3), and their demonstrated, long term experience in advanced ISCO applications in the field, neither the regulatory authority, environmental consultant, nor site owner expressed a need or desire for conducting a field pilot study.

# 5. Full-scale application

### 5.1 Main Reagent

The primary remediation strategy for the site-specific conditions (i.e. geology, contaminant situation, and hydrogeology) was to first conduct an ISCO treatment comprising:

- Targeted emplacement of an activated, dual- phase oxidant solids with significant treatment longevity (potassium persulfate activated by calcium peroxide);
- · Construction of oxidant emplacement boreholes as injection wells;
- Monthly then quarterly groundwater sampling and analysis ("iterative feedback loop");

"Secondary Treatment", once indications that the primary oxidants were exhausted:

- Optimized reinjection of solution oxidants (sodium permanganate) into injection wells exhibiting contaminant rebound
- Continued groundwater monitoring ("iterative feedback loop")

"Tertiary Treatment" follows in those remaining areas where contaminants persist:

• Enhanced aerobic bioremediation through slow release oxygen and nutrients ISCO technology using solid phase oxidants emplaced by environmental fracturing (Targeted Solids Emplacement, "TSE®") was selected due to its cost-effectiveness for treating multiple contaminants (chlorinated and petroleum hydrocarbons) present in low- permeability soils (silty sands and clay), its relatively non-disruptive implementation (direct push drilling) compared to ex situ methods considered, and due to its environmental sustainability (contaminant destruction vs. transfer).

Potassium persulfate was chosen as the primary oxidant due to its ability to form sulfate





radicals by alkaline activation through the addition of calcium peroxide (aktivator and secondary oxidant). The potassium form of persulfate also provides greater oxidant longevity due to its relatively low solubility. Calcium peroxide, in addition to activating persulfate, ensures a steady supply of slow-release oxygen into groundwater even after the persulfate oxidant has been exhausted.

The ISCO approach implemented at the site was designed to oxidize primarily CHCs with secondary consideration to BTEX contaminants, as these had been largely removed in a limited excavation of surface soils at the site. Persulfate is effective in oxidizing these contaminants and is less sensitive to SOD than other oxidants considered, and less hazardous to handle on site.

Environmental fracturing using Targeted Solids Emplacement (TSE®) coupled with Direct Push drilling was used as the preferred emplacement technology to ensure the distribution of high-solids oxidant slurry at selected depth intervals within contaminated soil zones of at least 6 m radius from injection boreholes. A total of 15 injection boreholes were used to emplace over 10.000 kg of persulfate-peroxide oxidants (dry mass) to depths of 10,5 m below ground surface (bgs). The mixing and fracture-emplacement of oxidants took place over 2 weeks followed by 3 weeks of injection well drilling, construction, and well development.



High pressure injection/fracturing/ and mixing equipment used for fractureemplacement of persulfate-peroxide oxidant slurry into subsoils through direct push drill holes

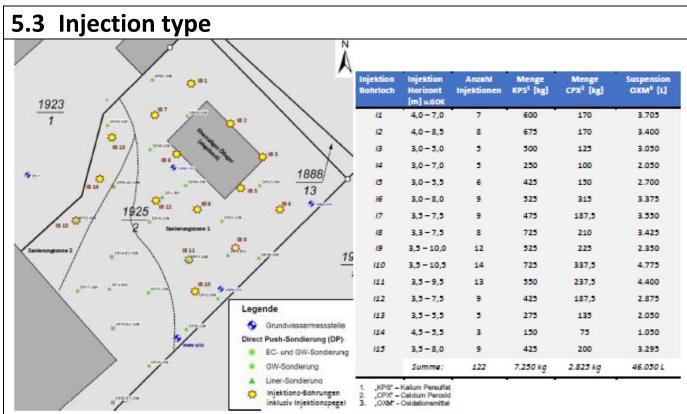




#### 5.2 Additives

The ISCO approach implemented at the site required that high- concentration oxidant slurry comprising low-solubility, solid based oxidants be emplaced into low permeability subsoils. This method of oxidant emplacement differs fundamentally from a simple injection of a solution based oxidant at low concentration which is normally the case (e.g. 4% potassium permanganate solution).

In order to ensure that solid based oxidants stay suspended in a water based slurry during pumping, and to avoid oxidants being "screened out" by fine grained aquifer sediments during emplacement into subsoils, a food-grade organic polymer gel was used to thicken the slurry to the required viscosity to ensure its placement at a concentration of upwards to 40% oxidant solids throughout its radius of distribution..



The fracture-emplacement of oxidant slurry using TSE® technology was achieved by advancing drill rods using Direct Push drilling equipment into subsoils to predetermined depths, followed by injection of slurry under hydraulic pressure using specialized high-pressure injection assemblies, and pumping and mixing equipment. The injections were conducted in a "top-down" approach, at 0,5 m depth intervals to the maximum depth of

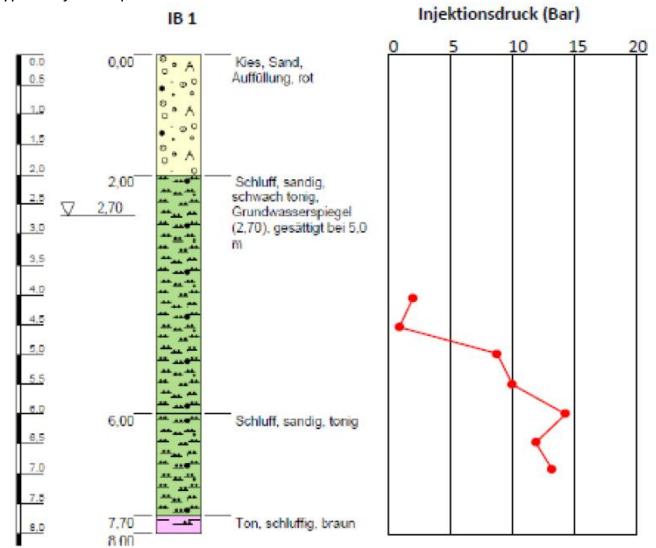




impacts at each location. A total of 122 injections at 15 pre-determined locations within the contaminant plume was thus achieved (see site plan and table above).

Two of the injection boreholes were initiated in previous MIP investigation borings in order to minimize coring of the concrete surface. The spacing between borings was approximately 7 to 8 m which ensured overlapping oxidant distribution between injection points. All of the 15 injection boreholes were subsequently completed as 2" diameter (50 mm) injection wells for follow-up solution oxidant/biosubstrate treatment (where required), and for monitoring and sampling purposes.

All operational parameters were recorded during fracture-emplacement of oxidants (fracture and propagation pressures, flow rates, volume) including operational losses due to short-circuiting to ground surface (approx. 1% of the total volume injected). A typical injection profile is shown below.



In this manner a total of 10,125 kg of oxidant (dry) mass was hydraulically emplaced throughout contaminated sediments as a slurry with total volume (including flush volumes) of approximately 46 m<sup>3</sup>.





#### 5.4 Radius of influence

The determination of a "radius of influence" for the introduction of fluids into subsurface soils is seldom more than a theoretical calculation, as the actual "radius" of distribution is highly variable, even within a single injection point, as it is governed by soil heterogeneities (variable porosity, permeability, fines content), hydrogeological pressure gradients, and geotechnical/geotechnical properties of the subsurface (soil density, cohesion, plasticity, structure and fabric, and in situ stress conditions) see discussion in Section 7.2

In the case of the subject site, a theoretical radius of oxidant distribution of 3,5 m was used for the injection work at the site, although field observations indicated that the extent of oxidant distribution was upwards to 6 m at some locations.

#### 5.5 Process and performance monitoring

Performance monitoring of control parameters to assess the effectiveness of ISCO remediation need to be tailored to the specific chemical characteristics of oxidants being applied, and the physical, geochemical, and microbiological parameters in groundwater. Important field parameters that were included in the post application monitoring comprised pH, electrical conductivity, redox potential, temperature and dissolved oxygen. Monitoring of pH is especially important in order to assess the buffering capacity of the soils and the potential of metals mobilization. Oxidant specific parameters included sulfate, dissolved and potassium (indicators of the primary oxidant, dissolved oxygen, potassium persulfate), as well as calcium, alkalinity, and dissolved oxygen (indicators of the secondary oxidant, calcium peroxide). Monitoring of the component cations in groundwater serve as an indicator as to the extent of distribution and relative concentration of oxidants present within the groundwater contaminant plume. Monitoring of contaminants included BTEX, CHCs (PCE, TCE, DCE, VC) was conducted, as were reaction products methane and carbon dioxide. Also included were analyses of metals.

Groundwater monitoring campaigns were carried out on a monthly basis for the first three months after oxidant emplacement and bi-monthly thereafter. Monitoring data and groundwater laboratory analytical have been collected over a span of 10 months so far.

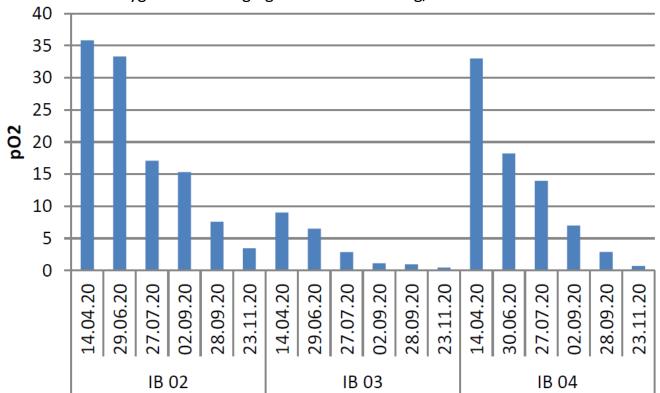




# 6. Post treatment and/or Long Term Monitoring

# 6.1 Post treatment and/or Long Term Monitoring

Initial field parameter measurements taken within a month of oxidant emplacement within the contaminant plume area showed strong evidence of oxidation taking place, as indicated by elevated redox (electron activity) conditions ranging from 250 to > 600 mv, and dissolved oxygen levels ranging from 10 to > 40 mg/l



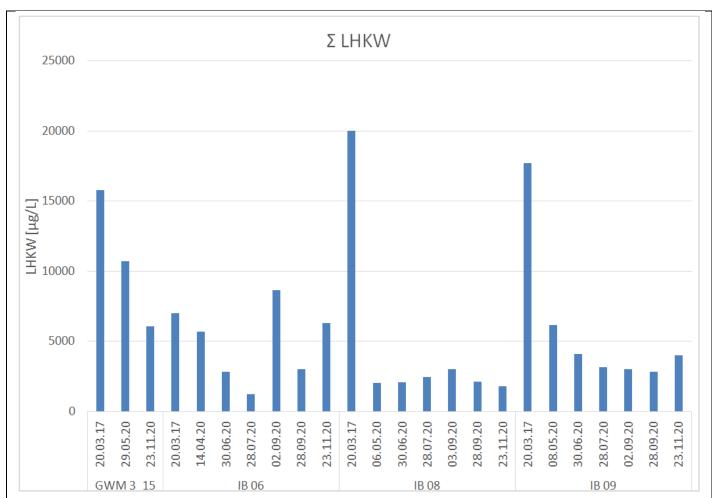
Typical oxygen depletion profile in plume area wells after injection

An assessment of oxidant longevity and contaminant persistence was made based on the decreasing concentrations of BTEX and CHCs in groundwater and the relative magnitude of contaminant rebound (where present) in the injection and monitoring wells over the monitoring period (10 months). The results indicated that of the 15 wells completed within the contaminant plume area:

- 1 well showed no contaminant rebound
- 6 wells showed moderate rebound effects (< 25 % of pre-treatment concentrations)</li>
- 8 wells showed strong rebound effects (> 50 % of pre-treatment concentrations)







Typical CHC concentration profiles for: perimeter groundwater monitoring well GWM3\_15; plume area wells IB05 (with rebound); plume area well IB07 (no rebound) after oxidant injection.

The collective assessment of field parameters such as redox and dissolved oxygen with geochemical parameters (e.g. sulfate) and contaminant concentration over time suggests that a longevity of approximately 4 to 5 months was achieved with the primary oxidant (persulfate), and a continuation of milder direct oxidation processes with oxygen (slow-release peroxide) from the secondary oxidant for a few months longer.

Based on this initial phase of performance monitoring in the "iterative feedback loop" approach, a second round of oxidant injection is planned in 2021 for those wells exhibiting rebound effects. This process is continued until concentrations have reached a low enough concentration whereby microbial amendments can be effectively applied to "polish" the remaining residual and trace concentrations of contaminants to reach remedial goals.





### 7. Additional information

#### 7.1 Lesson learnt

The subject site presented many challenges to an effective ISCO strategy, due to:

- Uncertainty as to origin of some contaminants (possibly off-site?)
- Extremely high concentrations of co-mingled and mixed chlorinated and petroleum hydrocarbon contaminants
- Low permeability of aquifer sediments
- Enforcement order to remediate
- Cost sensitivity
- Need for decommissioning of former building and shallow excavation of contaminants prior to in situ remedial work
- Presence of underground facilities (storage tank, pipeline)
- Nearby stream adjoining property
- Active business operations on property

Before an ISCO plan was even considered, the property was subject to high resolution characterization (Direct Push MIP and EC investigation) in order to better delineate the lateral and vertical extent of contaminants to allow a better estimation of contaminant mass in the subsurface. This was done in conjunction with pump testing and soil vapour extraction trials which determined that pump and treat and vacuum extraction were not feasible remedial methods for the site geology.

These data formed the basis of a laboratory feasibility study to assess applicable in situ oxidation and bioremediation options, which determined that ISCO was the preferable option in the initial stage of treatment.

The key to an effective ISCO treatment was to determine:

- Effective oxidation product(s) for treating both CHC and BTEX contaminants;
- Oxidant dosing rates which could be applied in the field for the various magnitude of contamination present across the site;
- Likely Mode of Distribution of oxidants (soil permeation or fractures) and best suitable drilling method for injection (auger, sonic, Diret Push, manchette wells with packer, open hole packers, etc)
- Optimization techniques in the field to minimize loss of oxidants to the ground surface through existing boreholes, underground structures, and backfilled soils
- Applicable monitoring parameters and frequency of monitoring/sampling
- Determination of plan and timing for follow-up injection treatment

Despite careful planning of the design based on the above criteria, problems arose in the field related to short circuiting (loss of oxidant slurry) to surface through backfilled soils





after recent excavation activities, and though old investigative boreholes not adequately sealed.

Attempts were made at an operational level to mitigate such losses by increasing fluid viscosity, and oxidant slurry density, while reducing total injection volumes to mitigate the surfacing of oxidants at certain injection locations. Oxidant slurry coming to surface was collected and stored in IBCs for subsequent injection at other borehole depths or locations.

Fracture-emplacement was the dominant mode of distribution of dual stage oxidants (slurrified potassium persulfate and calcium peroxide, supplied by *PeroxyChem*) which proved effective over a period of at least 5 to 6 months. "Iterative Feedback Loop"monitoring of post-injection groundwater quality was effective in determining those locations within the contaminant plume where, and to what extent, follow up injections (oxidants or bioamendments) are required. This example of a staged, treatment train approach to in situ remediation optimizes the resources (time and material costs) related to achieving site-specific remedial objectives at site without disruption to ongoing business operations.

#### 7.2 Additional information

The determination of a "radius of influence" for the introduction of fluids into subsurface soils is seldom more than a theoretical calculation, as the actual "radius" of distribution is highly variable, even within a single injection point, as it is governed by soil heterogeneities (variable porosity, permeability, fines content), hydrogeological pressure gradients, and geotechnical/geomechanical properties of the subsurface (soil density, cohesion, plasticity, structure and fabric, and in situ stress conditions). In fact the "radius" of distribution for liquid and solid treatment amendments is in most cases not a radius at all, rather a measure of the general extent of oxidant distribution from the point of injection. The distribution can be elliptical, off-center, or asymmetrical for example. This is due to the fact that distribution is a function of the inherent properties of injected amendments (viscosity, temperature, pH, polarity, particle size, ionic properties, precipitation, etc.) as it relates to soil properties. Therefore, a fundamental consideration in the estimation of the effective lateral Extent of Amendment Distribution (EAD) to a site is an assessment of the likely Mode of Distribution that is to be expected, based on the physical and chemical characteristics of the treatment amendment to be injected in relation to the soil characteristics (primarily porosity and permeability) being injected into. This is extremely important, as it is the mode of amendment distribution which will govern the actual extent of subsurface distribution for any given amendment





into soil or even bedrock, and can vary significantly. Therefore, the likey mode of distribution must be recognized in any remedial design involving the introduction of treatment amendments into subsoils.

Empirical evidence for the Mode of Distribution of liquid and solid phase chemical and biological treatment amendments in various geology over two decades of project work at sites across North America, Europe, and Asia was summarized by Bures (2009) as follows:

# Injection of Amendments: Mode of Distribution is important!

Infiltration if: Dp < √Kf/7 or ... Induced Pathways (FRAC) if: Dp > √Kf/7

Harris and Odem, 1982 (Dp: Particle Diameter in microns, Kf in millidarcys)

	MODE OF AMENDMENT EMPLACEMENT INTO SOILS AND BEDROCK (with respect to Hydraulic Conductivity)									
COMMON AMENDMENTS EMPLACED	>10 <sup>-3</sup> m/s	10-3	to	10-5	<10-5	<10-6	<10-7	<10-8	<10-8	<10-6 m/s
	Gravel	coarse	Sand medium	fine	silty Sand	Silt	silty Clay	Clay	Competent Bedrock	Fractured Bedrock
Silica Sand (Proppant)	INF	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC
Coarse Zero Valent Iron	INF	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC
Micro- Iron	INF	INF	INF/ FRAC	INF/ FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	INF/ FRAC
Oxidant Solids (as slurry)	INF	INF	INF	INF/ FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	INF/ FRAC
Oxidant Liquids (in solution)	INF	INF	INF	INF	INF	INF/ FRAC	FRAC	FRAC	FRAC	INF
Solution Blo-Amendments (Lactates, Vegetable Oils)	INF	INF	INF	INF	INF	INF/ FRAC	FRAC	FRAC	FRAC	INF
Viscous Bio-Amendments (Molasses, Whey, etc.)	INF	INF	INF	INF	INF/ FRAC	FRAC	FRAC	FRAC	FRAC	INF
Solid Bio-Amendments (Cellulose, Chitin)	INF	INF/ FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	FRAC	INF/ FRAC

INF = Infiltration and permeation through pore space is the primary mode of amendment emplacement.

FRAC = Targeted Solids Emplacement (creation of a network of permeable treatment Pathways) is the primary mode of amendment emplacement.

In general, the mode of distribution of a liquid or solid treatment amendment in subsoils and bedrock can be estimated by a comparison of the particle size of the material to be injected to the pore throat diameter of the receiving geology, which can be defined as the square root of: formation permeability, Kf, divided by seven (Haris and Odem, 1982). For treatment amendments where the particle size is smaller than the pore throat diameter, the mode of amendment distribution is by pore space permeation (blue area above). If





the amendment particle size is smaller than the availabe pore throat diameter, then the mode of distribution is through the formation of a fracture, defined by its thickness, width, length, orientation, and inclination to the ground surface (green area above).

Even for liquid amendments where the receiving geology has tiny pore space measured in angstroms, (e.g. clays), the mode of amendment distribution will trend towards a fracture, since even moderate injection rates cannot be accommodated by low effective porosity soils, resulting in a tensile failure of the soil and the creation of a fracture.

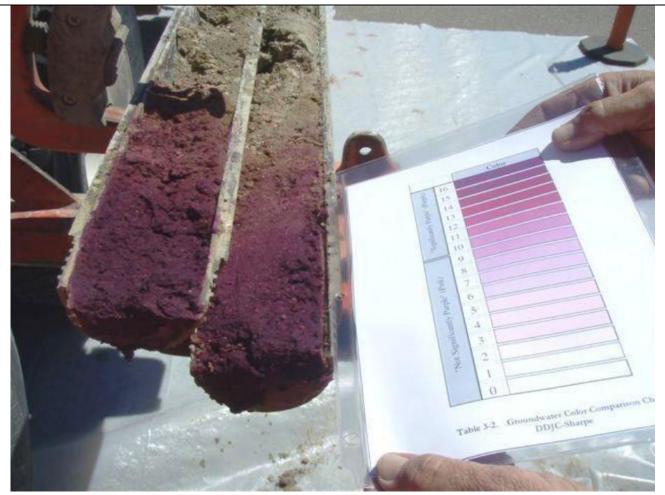
There can also be instances where the amendment being emplaced into the subsurface exhibits characteristics of both infiltration through pore space by permeation and the formation of a fracture. These are so called "hybrid" fractures, that is, fractures with significant "leak off" into surrounding soil pores.

Why is this important? Because the mode of emplacement has a significant bearing on the radius of influence and the transport processes for contaminant distribution via injection techniques in subsoils. For example, even the injection of solution based, liquid amendments (oxidants or bioamendments) will result in the formation of fractures or hybrid fractures in soils with hydraulic conductivities of  $< 1 \times 10$ -6 m/s. For any given volume of solution amendment injected, therefore, the observed "radius of influence" will appear to be much greater than what would normally be expected if this calculation were based on the assumption of permeation. A volume of 1000 L of liquid amendment injected into a soil with an effective porosity of 10 % will correspond to a theoretical radius of influence of roughly 1,8 m per m of well screen if permeation through pore space were assumed, but the same volume would result in an theoretical fracture radius of 8 m (!!) if soil permeabilty is insufficient to allow radial porous flow. Therefore it would be prudent to know what the predominant mode of distribution to be expected at a site is, before implementing a full scale remedial design using "radius of influence" calculations, and hence injection well spacing, that are possibly based on faulty premises. An equally important consideration in the importance of understanding the mode of distribution is the contact mechanism of injected amendments with respect to

distribution is the contact mechanism of injected amendments with respect to contaminants. Injection by radial permeation through the pore space in soils with relatively high permeability results in advective and dispersive mixing with dissolved phase contaminants. In contrast, the mechanism of contaminant treatment via emplacement of treatment amendments by fracturing, which by implication means in fine grained soils, is primarily through pressure induced penetration into soils at the fracture face, chemical gradient, and diffusion of oxidants from fractures into soil mass between individual fractures (see below).







Example of oxidant diffusion profiling in silt soil cores 90 days after fracture-emplacement of potassium permanganate oxidant slurry (bottom of core) Photo courtesy of URS, 2006 – Bures archive

# 7.3 Training need

Effective in situ remediation using oxidants requires a multi-disciplinary approach across a wide spectrum of scientific and engineering know-how. The end effect means that remedial design, and particularly the practical application of ISCO can be complicated, as it requires specialized knowledge in:

- Geology
- Groundwater hydraulics
- Organic and inorganic chemistry
- Biochemistry
- Polymer chemistry
- Fluid mechanics





- Soil / Rock mechanics
- Drilling technology
- Injection technology
- Mixing and pumping technology
- Tracer and geophysical mapping technology
- Risk assessment
- Knowledge of regulatory requirements

It become obvious that based on the comprehensive suite of expertise required above, that the effective application of ISCO is very much a team effort. Although much of the expertise listed above are standard fields of study at universities or technical colleges, there is simply no substitute for experience gathered on actual project applications. Therefore, academic and industry workshops, conferences, technology specific webinars, and shared practical experience are of significant importance for anyone wishing to be a competent practitioner in this field.

#### 7.4 Additional remarks

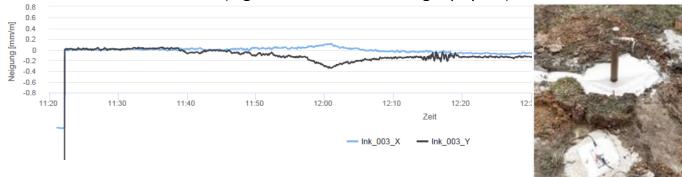
Any meaningful discussion of in situ chemical oxidation (ISCO) is incomplete without an understanding of the Mode of Distribution of oxidants being introduced into the subsurface environment and Contact Optimization with contaminants residing there. These considerations are essential elements for achieving remedial success using an ISCO approach, yet tend to be poorly understood by many working in this field. Fortunately, there exist a variety of innovative remedial enhancement and remedial performance monitoring technologies to rectify these shortcomings, among others:

- Dual— or multiple component oxidant formulations with a variety of activation mechanisms to achieve the highest oxidation potentials for oxidizing even mixed or co-mingled subsurface contaminants (e.g. CHCs and TPH) over long periods
- Incorporation of environmentally benign surfactant technology into the ISCO process to improve the performance oxidants by improving contaminant availability and oxidant penetration into pore spaces
- Specialized mixing, pumping, and rapid delivery technologies that enable precise and targeted emplacement of high concentration oxidant solids (as slurry) into permeable as well as impermeable sediments, including bedrock (e.g. TSE® technology with Direct Push drilling), or the emplacement of permeable pathways (e.g. sand fractures) in low permeability soils which can then be repeatedly injected with oxidant solutions.
- Employing the use of non-intrusive and robust geophysical techniques to map





subsurface distribution of liquid or solid oxidants from their point of delivery either as radial permeation, fracture proppagation, or hybrid distribution in subsurface sediments (e.g. SensaTrax® tiltmeter geophysics):



Furthermore, ISCO as a remedial application should not necessarily be viewed as the sole approach to site remediation, as by itself it rarely achieves every remedial target goals for every contaminant at every site. Rather it should be seen as part of a Treatment Train approach whereby oxidation can, at an appropriate point in the remedial process, be transitioned into a more passive bioremediation approach (aerobic or anaerobic Engineered Natural Attenuation, "ENA") to mitigate remaining contaminants to their remedial endpoints.

# **Glossary of Terms**

Term (alphabetical order)	Definition
BTEX	Benzene, Toluene, Ethylbenzene, Xylenes
CHC	Chlorinated aliphatic hydrocarbons
TPH	Total petroleum hydrocarbons
DCE	Dichlorethylene
EAD	Lateral, effective Extent of Amendment Distribution
ENA	Engineered natural attenuation
IFL	Iterative Feedback Loop
ISCO	In situ chemical oxidation
ISBR	In situ biological remediation
MIP	Membrane interface probe (high resolution
	characterization of contaminants)
PAH	Polycyclic aromatic hydrocarbons
PCE	Tetrachlorethylene
SOD	Soil Oxidant Demand
TCE	Trichloroethylene
TMB	Trimethylbenzene
TSE∘	Targeted Solids Emplacement (by Sensatec GmbH)
VC	Vinyl chloride

# 1. Contact details - CASE STUDY: ISCO n.7

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# 2. Site background

### 2.1 History of the site: Challenges and Solution

- Background- In an industrial area in Israel contamination from solvents was found in groundwater from the site. Apparently due to the industrial activity of some factories from the 1950s.
- The remediation was performed as part of a change in the site designation from industrial activity to commercial activity.
- Characteristics of the contamination In investigations performed on the site over the years, high concentrations of chlorinated solvents were found, the main one was trichloroethylene (TCE).
- The goal- reduction of the concentrations of chlorinated carbon in the groundwater, in a total area of about 300 square meters. The reduction was examined by comparing the target values as agreed with the Water Authority.
- The selected rehabilitation technology- An alternative survey was prepared for the remediation of the site. Following its findings, it was decided to treat the groundwater by injecting a chemical oxygen (potassium permanganate KMnO<sub>4</sub>).
- The main challenge in performing the remediation during the remediation period, construction work was performed to establish a new tower and an underground parking in the site, therefore safety measures had to be taken so that the combination of installing the foundations of the tower during the remediation period would be possible.

# 2.2 Geological and hydrogeological setting

- The soil at the site is sandy.
- The depth of the groundwater at the site, approximately 20 m below the ground.

#### 2.3 Contaminants of concern

The results of the groundwater sampling show that the contaminants whose concentration exceeded the threshold values are trichloroethylene, manganese and chromium.





### 2.4 Regulatory framework

- In Israel, water remediation is in the responsibility of a government ministry the Water Authority.
- The remediation plan and remediation reports are reviewed and approved by this authority.
- The following is a list of the target values of the contaminants, as approved by the Water Authority:
- 1. Tetrachlorethylene 187 μg/L
- 2. Trichloroethylene 374 μg/L
- 3. 1,1-dichloroethylene 187 μg/L
- 4. cis-1,2-dichloroethylene 935 μg/L
- 5. trans1,2-dichloroethylene 935 μg/L
- 6. Vinyl Chloride 9 μg/L

# 3. Laboratory-scale application in field

#### 3.1 Laboratory scale application

Performing preliminary actions included:

- TOD test as part of the installation of the injection wells, soil samples were taken
  to perform tests for the "natural oxygen demand" of the soil. Based on the results,
  precise calculations of the amount of oxygen and solution volumes required for
  the treatment of the contaminant on the site were performed.
- Pilot test This test included injecting water in small volumes in order to examine injection rates, pressures and flow rates in the various wells before performing the oxidizing injections.





# 4. Pilot-scale application in field

#### 4.1 Main treatment strategy

- The work includes three main stages:
  - Stage A performing a preliminary pilot checking flow rates and pressures
  - performing tests in the field and in the laboratory to determine the injection parameters.
  - o **Stage B** Perform the complete remediation by performing the injection.
  - Stage C Concluding monitoring of groundwater to examine compliance with remediation, in accordance with target values of the contaminants.
- This remediation technology was chosen after examining all the remediation options.
- Considering the characteristics of the site and the fact that during the remediation period construction on the site was being carried out at the same time, it was decided to apply this technology.
- The challenge in this project was to enable the construction work and the construction of the underground parking at the same time as the groundwater treatment at the site.
- The oxidation injection was performed through 8 double injection wells to a depth:
  - Shallow: 0–3 m below groundwater level. Deep: 3.5–8 m below groundwater level.
  - The injections were performed for 3 days during which approximately
     93,300 liters of permanganate solution were injected at concentrations of
     0.5% to 2%, which included 1,025 kg of potassium permanganate.
  - At the end of the injections, air was injected for about three weeks to disperse the oxidants in the horizontal dimension so as to increase the distribution of oxygen in the aquifer.
- In order to monitor the remediation process, every six months groundwater monitoring and sampling was carried out for laboratory analysis of contaminants and geochemical parameters.





### 4.3 Injection type

- The layout of the wells at the center was designed according to the treatment area, the depth of contaminant concentration, the oxidizing properties and the soil properties.
- The permanganate solution, similar to the TCE substance, has a higher density than water and therefore, by its nature, "sinks" downwards. Therefore, the layout of the wells in the vertical axis was designed so that the effect of the treatment by the injected solution would cover the entire incision, up to a depth of 8 meters below groundwater level.
- The injections were performed for 3 days during which, approximately 93,300 liters of permanganate solution were injected at concentrations of 0.5% to 2%, which included 1,025 kg of potassium permanganate.

# 4.4 Radius of influence

 The radius of impact was defined as 4m in the horizontal dimension in accordance with experience from other sites with similar characteristics and in accordance with preliminary tests that included injecting water in limited volumes to test injection rates, pressures and flow rates in the various wells before performing the oxidation injections.





#### 4.5 Control parameters

Field monitoring and sampling program that will adequately monitor both the dispersion of the oxidant and the effectiveness of the treatment in three dimensions are required. Usually measurements concerning oxidant dispersion are conducted more frequently than COC analysis and are completely different if the oxidant is in liquid or gas form.

- Below is the sampling frequency of the monitoring wells:
  - Before the injection
  - A month and a half after the injection
  - o Three months after injection
  - Nine months after the date of injection
  - One year after the date of injection
- The following are the parameters tested in the groundwater sampling:
  - o VOC
  - o TDS
  - Metals
  - Alkalinity
  - o Bicarbonate
  - Nitrite
  - Main ions
- The following are the field findings examined in the groundwater sampling:
  - o ORP
  - o EC
  - $\circ$  pH
  - o OD





# 5. Full-scale application

### 5.1 Main Reagent

- Potassium Magnet (KMnO<sub>4</sub>) Permanganate in aqueous solution exists in the form of anion (MnO<sub>4</sub> $^{-}$ ), as an oxidizer with high oxidizing power to organic hydrocarbons in general and chlorinated hydrocarbons in particular.
- The solution was applied in a concentration of 0.5% to 2%.
- There was no change compared to the pilot test.

# 5.3 Injection type

- The injection was performed through eight new double injection wells, which were installed as part of the Remediation project:
- Shallow strainer from 0 (water surface) to 3 meters deep.
- Deep strainer from 3.5 meters to 8 meters deep.
- At the end of the injections, air was injected for about three weeks to disperse the oxidants in the horizontal dimension.
- The injection wells were placed as a rounded mesh cluster, 4 m apart.

### 5.4 Radius of influence

With no change from the pilot, as described in section 4.4





# 5.5 Process and performance monitoring

The injections included:

- Mixing the chemicals and preparing the injection solution in an outdoor facility.
- Transferring the injection solution to the site with the help of a dedicated tanker.
- Positioning the tanker on an elevated ramp at the site (20 m above the wellheads) and flowing the solution to the heads to the control and manifold.
- The control and monitoring manifold included a main faucet and a pressure gauge which allowed control of the injection flow to the wells and a system of faucets for controlling the flow of the solution to each faucet separately.
- A safety surface, made of flexible and thick HDPE (high density polyethylene)
  plastic, is spread out under the working point and the pipe branch, to prevent
  leakage outside the activity area in case of emergency.





# 6. Post treatment and/or Long Term Monitoring

# 6.1 Post treatment and/or Long Term Monitoring

A VOC analysis should be performed to detect chlorinated carbon concentrations as a result of the "rebound" effect.

### 7. Additional information

#### 7.1 Lesson learnt

Injecting the oxygen at high pressure may cause it to leak from the surface of the soil. It should be injected at an adjusted pressure that will not cause leakage.

#### 7.3 Training need

Training through workshops, preferably by the Ministry of Environmental Protection in order for the remediation processes to comply with the regulator's guidelines.

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# 2. Site background

# 2.1 History of the site: Challenges and Solution

The site is located in the Salberdin industrial area within the town of Zarautz. Outside its boundaries are urban residential areas formed by collective housing. The Zarautz Railway Station is located in the northeast. The sea is present 500 m north of the site.

# 2.2 Geological and hydrogeological setting

Geological description (below topsoil, asphalt or concrete)

- 0 0.5 / 2 m: filling
- 0.5 / 2 2 / 3.8 m: clay
- 2/3.8 m 2.2/4 m: clay silt
- 2.2 / 4 m -?: Sand

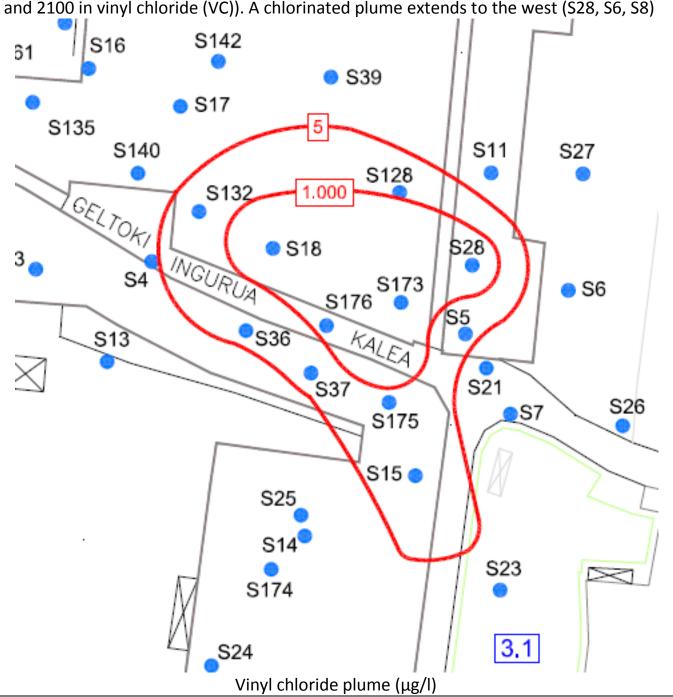
Presence of groundwater in the sand, direction of flow oriented towards the north, with an old channel.





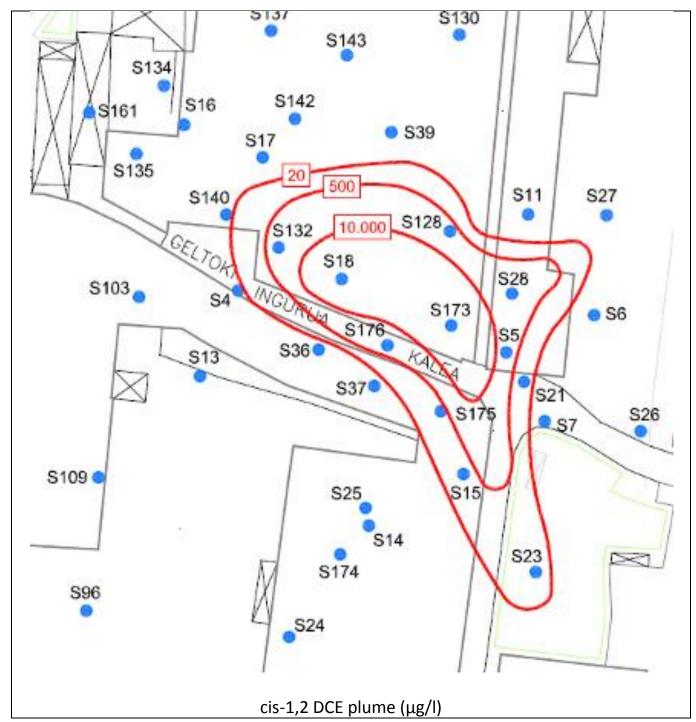
### 2.3 Contaminants of concern

The site presents a contamination of the groundwater for chlorinated compounds in concentrations that exceed the normative reference values (max 23,000  $\mu$ g/l in cis-DCE and 2100 in vinyl chloride (VC)). A chlorinated plume extends to the west (S28, S6, S8)













### 2.4 Regulatory framework

The aim of the treatment is to reduce the contaminating mass present in the groundwater and thus reduce or eliminate the potential health risks for the people living around it.

The treatment area corresponds to the right of way of the 88 injectors over a thickness of 10 m of aquifer.

The proposed target values for the impacted groundwater are presented in the table below:

Target values for groundwater (μg/l)

Interest compound	Target value (μg/l)
Vinyl Chloride	45
cis-1,2 Dichlorethylene	800
TPH AlifaticsC12-C16	30

Target values for groundwater (µg/l)

In addition, to evaluate the effectiveness of the treatment, SUEZ Remediation proposes the following reception criteria:

- 80% reduction in the average chlorinated solvent content,
- Minimum reduction of 50% on each individual piezometer,
- No abatement calculation for low concentrations <100 μg/l.</li>

# 3. Laboratory-scale application in field

We did not carry out a pilot sizing test prior to the implementation of the ISCO treatment.





## 5. Full scale application

#### 5.1 Main treatment strategy

The network of injection points consists of 88 points (I1 to I88), by means of a zoning in 3 areas with the following characteristics:

- Concentrated area, with a narrow network of structures of 31 injection points.
- Intermediate area, with a narrow network of structures with 27 injection points.
- Diffuse area, with a narrow network of structures with 30 injection points.

In each of the points, the injection of the reagent is carried out directly on the area where the aquifer develops, that is, on the basal stretch of sand located below the clayey silt and up to 10/12 m deep.

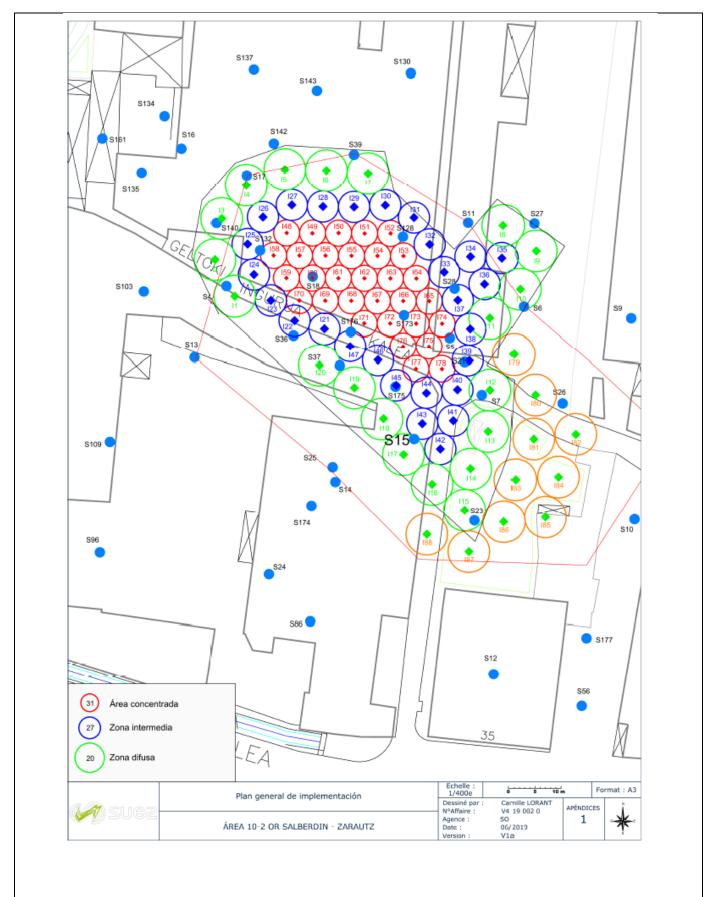
The injections were made with a system of 7 injection plates composed of

A system of non-return valves, a 3.5 bar pressure limiter, and a filter at the water inlet, A sodium permanganate IBC, connected to a dosing pump that allows to dilute the oxidant in line at 1 or 2%

3 lines with  $1m^3/h$  float flow meters, shut-off valves, to perform injections at each point. NB: it is possible to connect the injection point directly to the system output to increase the flow rate to  $3m^3/h$ .

























#### 5.4 Radius of influence

The theoretical amount of reagent to be injected is based on

- the amount of pollutant present in the aquifer
- the void volume in the aquifer (taking into account the hydraulic parameters of the aquifer in this sector and the important porosity)

The sodium permanganate reagent with a solution dosed at 1 or 2%, is implemented according to several successive campaigns, separated in periods of 3 months.

Oxidant dosage in each of the injection campaigns

ISCO Campaign	Injection point number	Injected volume (m³)	Quantity of 40% Permanganate (Tons)
1st campaign	88	1 385,5	19,5
2 <sup>nd</sup> campaign	24, recalcitrant points and/or with rebound effect		15: (9 + 5) In order to increase treatment performance according to SUEZ IBERICA and its client, 5 T of additional permanganate were injected during the second campaign

During the first injection campaign, SUEZ Remediation followed the theoretical dimensioning, namely

- Injection in each injector (88)
- Amount of permanganate solution at 40% (I): 16 m<sup>3</sup>
- Average dilution: 1.24%.
- Total volume of injected solution: 1385.46 m<sup>3</sup>
- Injection flow rate: 0.86 m<sup>3</sup>/h

For the second injection campaign, SUEZ Remediation has injected into the injectors where the VOCs were higher than the reference values.

In addition, Suez Remediation increased the dilution % and the injection flows (validated by a field test) to increase the diffusion of permanganate in the groundwater. The parameters are:

- Injection in each injector (24)
- Amount of permanganate solution at 40% (I): 12 m<sup>3</sup>
- average dilution: 1.65%
- Total volume of injected solution: 751.3 m<sup>3</sup>

Injection flow rate: 5.66 m<sup>3</sup>/h





### 5.5 Process and performance monitoring

In order to verify whether the water recovery targets are met and to define the evolution of the targets, the baseline situation will be determined and regular monitoring will be carried out to assess the progress of recovery.

During remediation, all wells and piezometers that present severe affection and values above the quality objectives, will be connected to the remediation system or for monitoring.

Likewise, periodic controls of the unaffected points were carried out to guarantee that the area of dispersion of the affection is not in expansion.

At the end of the treatment, after two injection campaigns, the results are:

For Cis1,2-Dichloroethylene,

- 87 out of 88 injectors have concentrations below the reference values, i.e. 99%
- the average reduction in concentrations between the initial state and the final state is 91%.

For vinyl chloride,

- 81 out of 88 injectors have concentrations below the reference values, i.e. 92%
- the average reduction in concentrations between the initial state and the final state is 91%.

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## 2. Site background

#### 2.1 History of the site: Challenges and Solution

The Site is a petroleum service station, with fuel storage in underground tanks, located in central Italy.

In 2005, the station was refurbished which included the replacement of the old underground tanks with new ones, which were installed in a different area of the Site. During the excavation for the removal of the old tanks, evidence of contamination was detected in the soil located below the tanks, therefore different environmental investigations were carried out over the year (in 2005, 2013, 2015 and 2017) on various environmental matrices (soil, groundwater and soil gas).

The results of the investigations showed the presence of two potential secondary sources of contamination, with exceedances of the Italian threshold limits (CSC D.Lgs. 152/06 and limits DM31/15):

 unsaturated deep soil (depth > 1m below ground surface (bgs)), with benzene, ethylbenzene, toluene, xylenes, light C≤12 and heavy C> 12 hydrocarbons and MtBE.
 The maximum detected concentrations were:

benzene	163	mg/kg SS
ethylbenzene	502	mg/kg SS
toluene	648	mg/kg SS
xylenes	1,472	mg/kg SS
light hydrocarbons C≤12	19,509	mg/kg SS
heavy hydrocarbons C>12	5,742	mg/kg SS
MtBE	736	mg/kg SS

 groundwater, with benzene, toluene, xylenes, total petroleum hydrocarbons and MtBE. The maximum concentrations were

benzene	163	μg/l
toluene	648	μg/l
p-Xylene	1,472	μg/l
Total hydrocarbons (as n-hexane)	19,509	μg/l
MtBE	736	μg/l

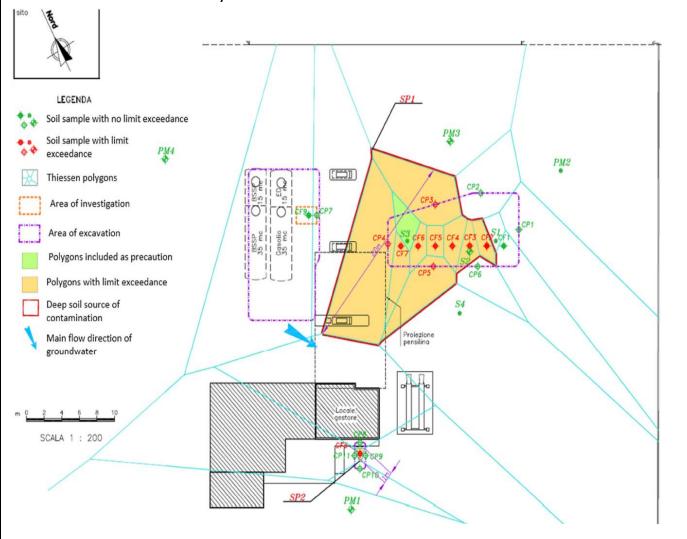




A human health risk assessment was developed for the Site, as required by Law. The assessment showed that the human health risk was acceptable but the Italian Law requires that groundwater contaminant concentration, at the wells located at the Site downgradient boundary, must meet with Italian threshold limits (CSC D.Lgs. 152/06). Some exceedance were detected in those wells and therefore groundwater remediation was deemed necessary for the site.

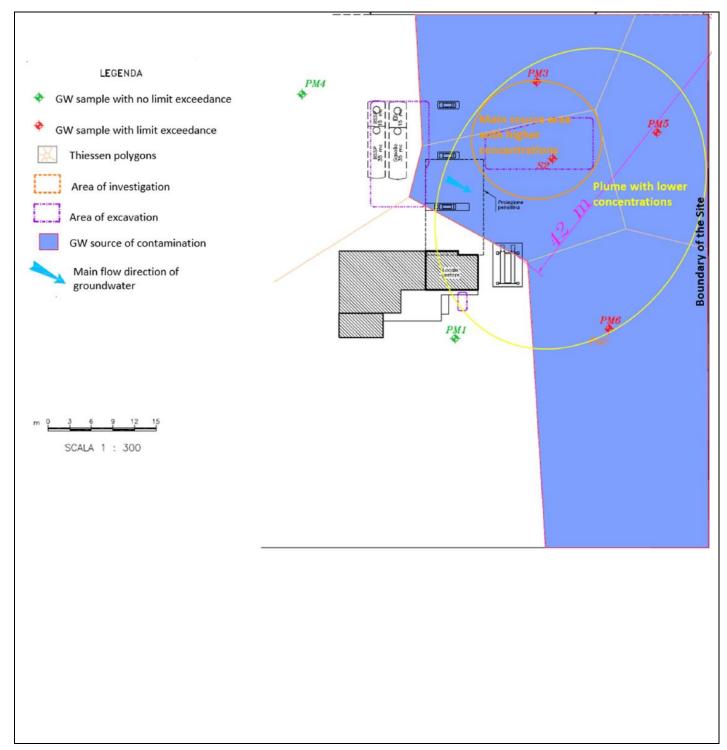
A screening of applicable remedial technologies was undertaken, using the screening matrix provided by ISPRA, showing that the best remediation technology for the Site is a combination of ISCO and bioremediation.

ISCO resulted more suitable for the area in which the old tanks were located, because of higher contaminant concentration. Bioremediation, i.e. the delivery of oxygen release compounds in the subsoil to stimulate hydrocarbons aerobic degradation, resulted more suitable at the Site boundary where the concentration of contaminants was lower.













## 2.2 Geological and hydrogeological setting

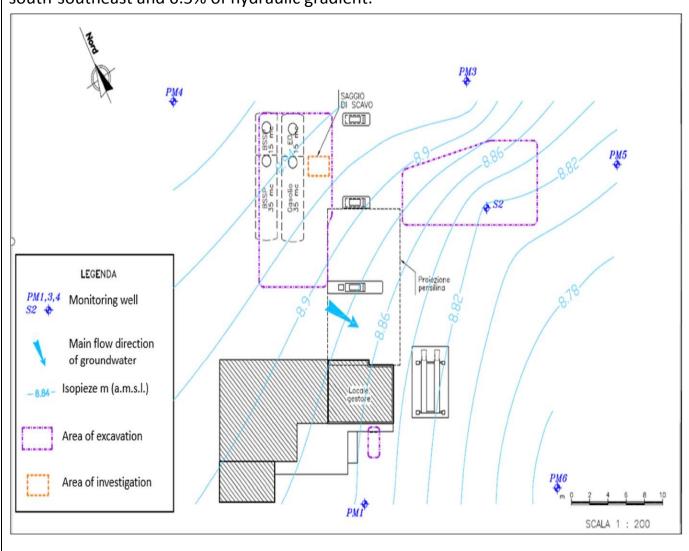
Site soil consists of fill material up to a depth of 0.8 m bgs, followed by sandy silt or silty clayey sand up to 4.2 m bgs (sandy loam). Below the latter fine to medium-fine sand is found up to the maximum investigated depth equal to 10 m from bgs (loamy sand).







The depth to groundwater is approximately 11 m bgs, with flow direction mainly south-southeast and 0.5% of hydraulic gradient.







#### 2.3 Contaminants of concern

The results of the environmental investigations showed the presence of two secondary potential sources of contamination, with exceedances of Italian threshold values (CSC D.Lgs. 152/06 and limits DM31/15):

- the deep soil (exceedance for benzene, ethylbenzene, toluene, xylenes, light hydrocarbons C≤12, heavy hydrocarbons C>12, MtBE);
- groundwater (exceedance for benzene, toluene, p-xylene, total petroleum hydrocarbons, MtBE);
- No LNAPL was detected on Site.

A human health risk assessment was developed for the Site, as required by Law. The assessment showed that the human health risk was acceptable but the Italian Law requires that groundwater contaminants concentration, at the wells located at the Site downgradient boundary, must meet with Italian threshold limits (CSC D.Lgs. 152/06). Some exceedance were detected in those wells, due to the plume generated from the main source area where the former tanks were located, and therefore a groundwater remediation was deemed for the Site The tables below shows the results of the comparison of the maximum detected concentration and the risk-based site-specific threshold limits (CSR), which are the remediation targets:

unsaturated deep soil (depth > 1m bgs), no remediation is needed.

Contaminant	Max conc.	Remediation	Unit
	on Site	targets	
benzene	163	163	mg/kg SS
ethylbenzene	502	502	mg/kg SS
toluene	648	648	mg/kg SS
xylenes	1,472	1,472	mg/kg SS
light hydrocarbons C≤12	19,509	19,509	mg/kg SS
heavy hydrocarbons C>12	5,742	5,742	mg/kg SS
MtBE	736	736	mg/kg SS

No remediation is needed





• groundwater within the site, no remediation is needed.

Contaminant	Max conc.	Remediation	Unit
	on Site	targets	
benzene	46	46	μg/l
toluene	3800	3800	μg/l
p-xylene	2619	2619	μg/l
total hydrocarbons	13000	13000	μg/l
(as n-hexane)			
MtBE	230	230	μg/l

• groundwater at the boundary of the site remediation is needed

Contaminant	Max conc.	Remediation	Unit
	on Site	targets	
benzene	3.2	1	μg/l
toluene	<0,13	15	μg/l
p-xylene	<0,16	10	μg/l
total hydrocarbons	220	350	μg/l
(as n-hexane)			
MtBE	230	40	μg/l





#### 2.4 Regulatory framework

The main environmental law in Italy is the Legislative Decree no. 152/2006 (D.Lgs. 152/06) that in Part four, fifth title sets specific rules for remediation of contaminated sites.

Moreover, a specific decree exists for petroleum service stations, Ministerial Decree no. 31/15 (DM31/15), which sets specific simplifications and procedures for those sites. There is no specific legislation for the application of ISCO technology.

The reference legislation establishes some threshold values (CSC D.Lgs. 152/06 and limits DM31/15) for the main contaminants both in soil and groundwater, if during the characterization there are one or more exceedance of threshold values, the site is defined as "potentially contaminated", and a human health risk assessment can be developed to estimate the risks deriving from the potential sources of contamination detected on site (defined by the samples with exceedance) and to calculate risk-based site-specific threshold limits (CSR). The legislature also states which values are of acceptable risk for the assessment.

If the estimated risks are lower than acceptable values, the site is defined as "not contaminated", and no remediation is needed. If the estimated risks are higher than acceptable values, the site is defined "contaminated", and remediation is needed. The risk based site-specific threshold limits (CSR) are the remediation targets.

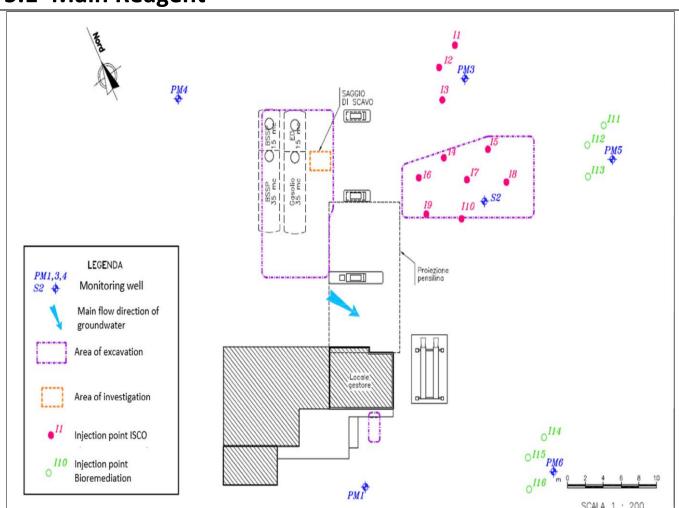
For the wells located down-gradient site boundary, the Italian Law sets the CSC as targets, and if exceedances are detected, a remediation is required.





## 5. Full-scale application

### 5.1 Main Reagent



The selected reagent was sodium persulfate ( $Na_2S_2O_8$ ), that needs to be activated to release the persulfate anion and radicals ( $S_2O_8^{2-}$ ) in water, which are strong oxidizing agents, successfully applied in similar contexts.

The activation can be performed by several means, in this case study the choice was alkaline activation by adding sodium hydroxide (NaOH).

To accelerate the reduction of the contamination in the boundary wells, it was also chosen to inject an oxygen releasing compound, specifically the calcium peroxide (CaO<sub>2</sub>), to enhance bioremediation.

The groundwater remediation was thus conducted by a combination of two technologies:

ISCO in the main source area (below the old replaced tanks, identified as the





primary source), characterized by higher concentration of contaminants and which generated the plume that extended to the Site boundary. The chosen oxidant was sodium persulfate ( $Na_2S_2O_8$ ), activated by creating an alkaline environment in the groundwater with the addition of sodium hydroxide (NaOH). The oxidant was applied in no. 10 injection points located in the main source area.

 Bioremediation in the area near the Site boundary, invested by the plume generated by the main source and characterized by lower concentration of contaminants. The chosen compound was calcium peroxide (CaO<sub>2</sub>). The compound was applied in no. 6 injection points located near the boundary wells that showed exceedance.

The treatment comprised of one single injection event and eventually, after 12 months of groundwater monitoring, a second injection event, to be assessed based on the results of the monitoring campaigns.

The injection points were drilled between March 13 and April 6, 2018, by installation of "manchette tubes" (see paragraph 5.3 for detailed information), while the injection activities took place between May 7 and May 11, 2018, applying the following dosages:

 Main source area, ISCO treatment (injection points I1÷I10): a solution of sodium persulfate, water and sodium hydroxide (as activator) was injected with the following dosages:

Thickness of injection	5	m
Dose of sodium persulfate per point	150	Kg
Slurry of sodium persulfate per point (diluted 15%)	1000	L
Sodium hydroxide diluted 25% per point (as activator)	288	L

• Wells near the boundary, Bioremediation treatment (injection points I11÷I16): a solution of calcium peroxide and water was injected with the following dosages:

Thickness of injection	5	m
Dose of sodium persulfate per point	53	Kg
Slurry of sodium persulfate per point (diluted 20%)	265	L





#### 5.2 Additives

Sodium persulfate ( $Na_2S_2O_8$ ), the selected oxidant for ISCO remediation in the main source area, needs an activator that allows its decomposition in persulfate anions and radicals ( $S_2O_8^{2-}$ ), which are strong oxidizing agents. The chosen activator was sodium hydroxide (NaOH) at 25% concentration, able to create an alkaline environment in groundwater.

Sodium persulfate activated with alkaline environment was applied successfully in similar contexts.

#### 5.3 Injection type

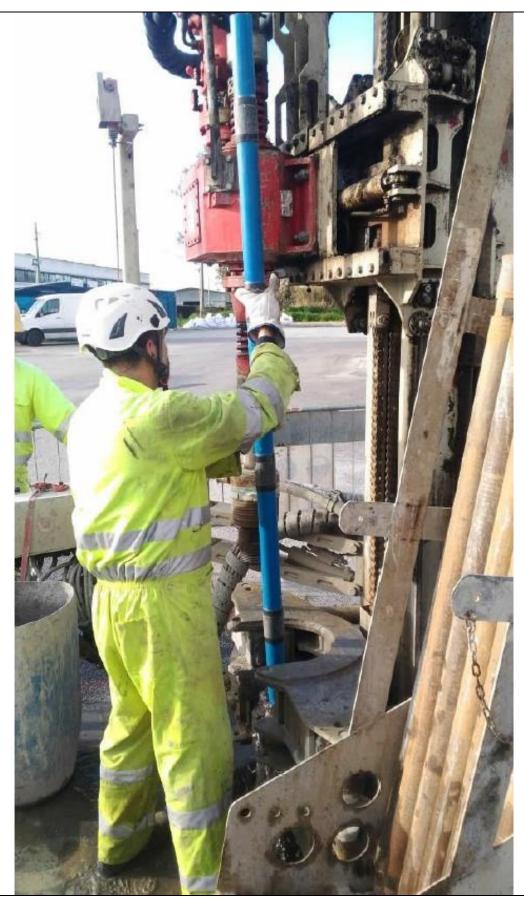
To perform the injection no. 16 new injection points were introduced on site, by drilling and installing no. 16 manchette tubes, with valves located every 50 cm between 10 and 15 m bgs, which is the saturated zone to be treated.

In the main source area no. 10 injection points were installed following an orthogonal grid to the main direction of flow separated by a distance between 3 and 5 m. While in the area near the Site boundary no. 3 injection points with interdistance between of the injection of approximately 3 m were installed upstream each of the two impacted well (n. 6 points in total). As mentioned before the treatment comprised one single injection event and eventually, after 12 months of groundwater monitoring, a second injection event, to be assessed based on the results of the monitoring campaigns.

The injection points were drilled between March 13 and April 6, 2018, by the installation of "manchette tubes", while the injection activities took place between May 7 and May 11, 2018.











#### 5.4 Radius of influence

Radius of influence estimated for the geology found on Site (medium-fine sands) is about 3 m.

#### 5.5 Process and performance monitoring

The monitoring of the remediation lasted one year following the schedule below:

- Before the injection (December, 2017):
  - first monitoring campaign, with groundwater sampling and measurement of physic-chemical parameters in all monitoring wells, to be used as an initial value (t0) to verify the progress of the treatment;
- Injection as illustrated in paragraphs 5.1 and 5.3 (May, 2018);
- During the first three months after the injection (June, July and August, 2018):
  - monthly monitoring of all monitoring wells, with sampling of groundwater and measurement of chemical-physical parameters;
- From the fourth to the twelfth month after the injection (November, 2018, February and May, 2019):
  - quarterly monitoring of all monitoring wells, with sampling of groundwater and measurement of physic-chemical parameters;

The physic-chemical parameters measured using a multiparameter portable probe were the following:

- temperature;
- redox potential;
- pH;
- electrical conductivity;
- dissolved oxygen.

The samples collected from the wells were chemically analyzed to determine the concentration of the following parameters:

Parameter	Method
BTEX+S	EPA 5030C 2003 + EPA 8260D 2018
total hydrocarbons (as n-hexane)	ISPRA Man 123 2015 - Metodo A+B
MtBE	EPA 5030C 2003 + EPA 8260D 2018
Sulphates	UNI EN ISO 10304-1:2009
Nitrates	UNI EN ISO 10304-1:2009



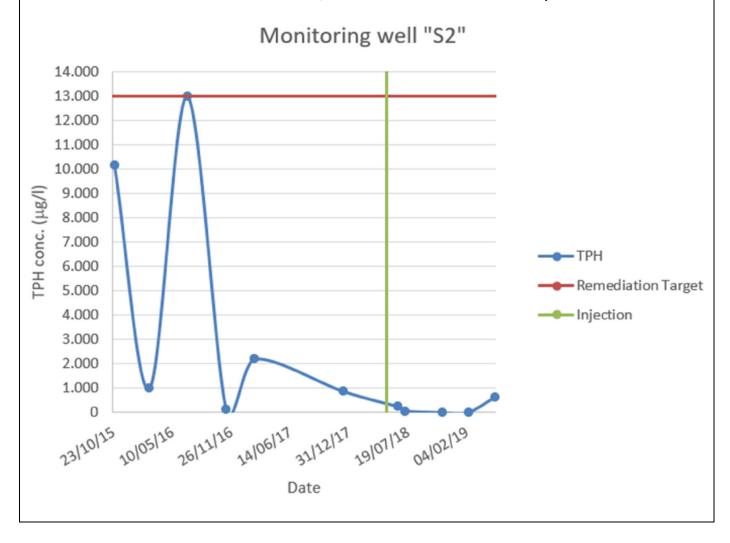


The last monitoring campaign was supervised by the local authorities who sampled the wells located at the boundary, validating the results obtained.

The remediation was completed successfully in the estimated time and one single injection event.

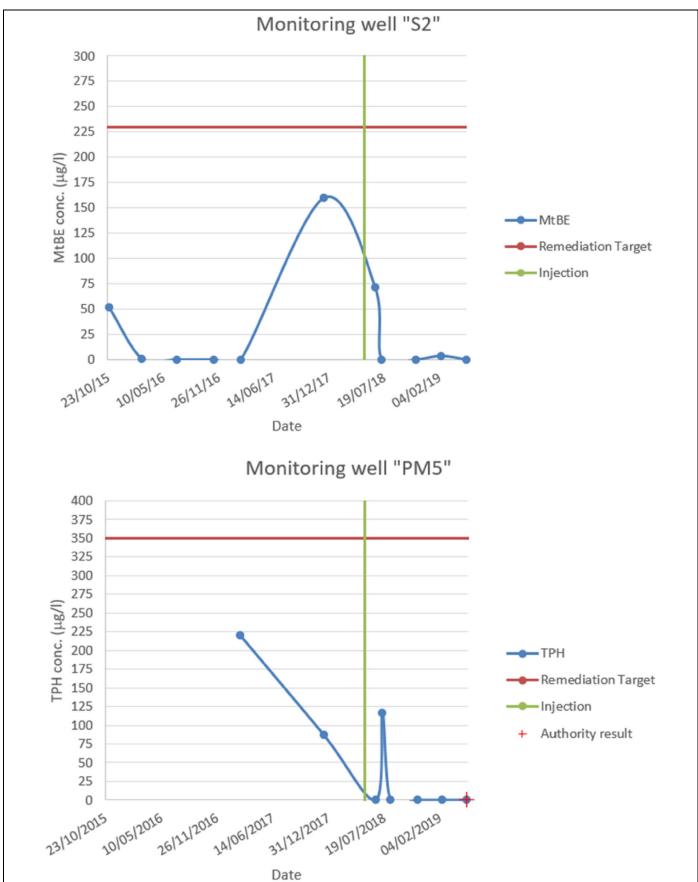
Based on the results the second injection events was not undertaken. Since both the Authority's and project manager results met remediation targets, the remediation process was certified as being concluded.

In the charts below is shown the contamination reduction obtained in S2, located in the main source area and in PM5 and PM6, both located at the boundary:



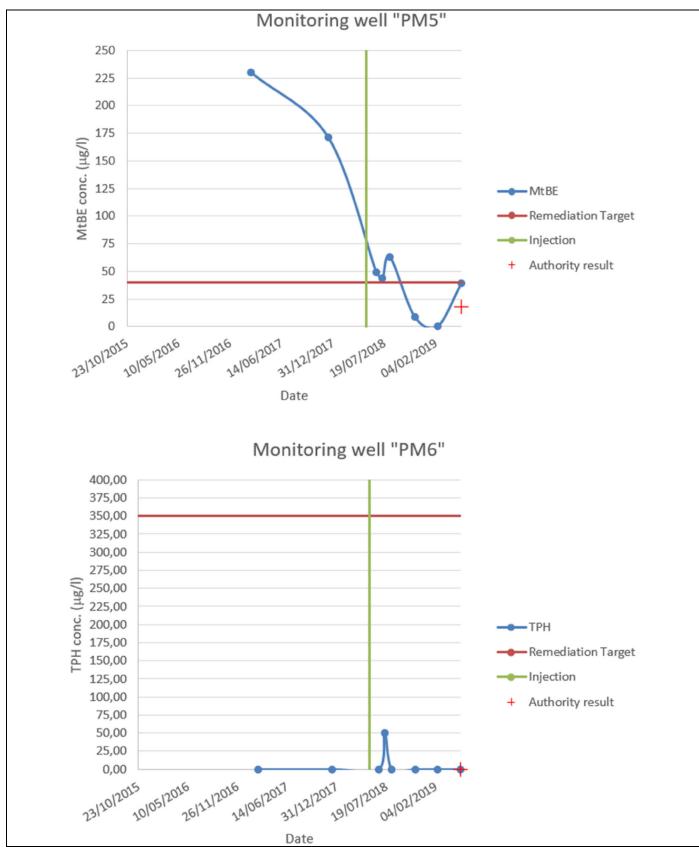






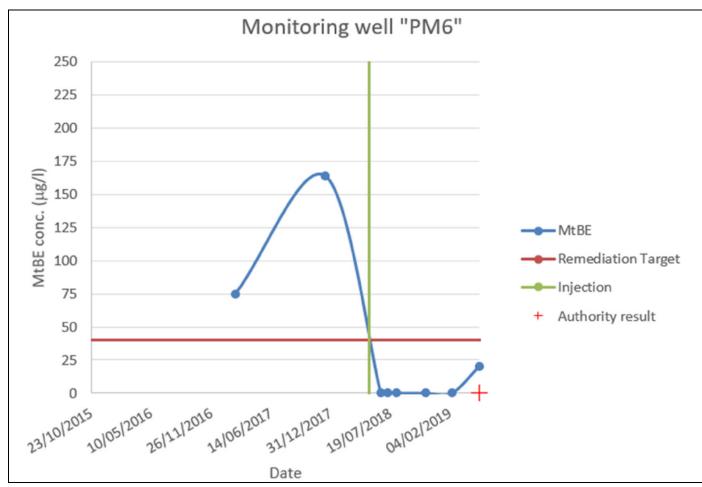












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## 2. Site background

## 2.1 History of the site: Challenges and Solution

- Former petrol station which had been dismantled and is in the process of ceasing activities
- Impact of soil and groundwater due to an incident release of hydrocarbons
- Preliminary remediation works on the unsaturated zone using digging techniques and on-site treatment with biopile and landfarming
- Identification of residual impacts around the groundwater table which were not accessible using excavation techniques (close to site boundaries / Soil stability)
- Implementation of laboratory testing to identify the best solution and design of the solution based on ISCO technique.
- Implementation of the ISCO technique (one campaign) at the site and reception of the treatment based on soil results after 1 year and monitoring of groundwater and soil gas quality over 2 years after the injection.

#### 2.2 Geological and hydrogeological setting

Geology: 0 to 0.2m – pavement or topsoil; 0.2 to 0.8m – made ground; 0.8 to 5m – clayey silt; 5 to 6m – marl limestone

Groundwater encountered at around 4.15 to 4.50m below ground level

Very low permeability of the clayey silt





#### 2.3 Contaminants of concern

Contaminants of concerns:

Total petroleum hydrocarbons and BTEX

Concentrations in the soil:

- TPH C5-C10: 250 up to 1,500 mg/kg
- BTEX: 80 up to 820 mg/kg
- And to a lesser extend: TPH C10-C40: 120 up to 3 100 mg/kg (mainly C12 to C21)

Maximum concentrations in the groundwater:

- TPH C5-C10: 52 000 up to 48500 μg/l
- BTEX: 43,000 up to 96980 μg/l
- And to a lesser extend: TPH C10-C40: 780 up to 7920 μg/l

No free-phase products.

No clean up goals –the aim was to improve the quality of the soil regarding residual concentrations of hydrocarbons

Treatment to be focussed on the soil around the groundwater table as the residual impacts are located in this area.

#### 2.4 Regulatory framework

Site into the process of ceasing activities (ICPE)

Guideline for contaminated site of 2017 – remediation of source area: The April 19th 2017 ministerial Note.

Remediation targets for former motorway petrol station were used but there were no regulatory remediation targets as such – these values are defined during a study conducted by a group of petrochemical companies, motorway operators and consultants in order to harmonise practises: "Approche méthodologique harmonisée pour la gestion de stations-services autoroutières – Guide de mise en oeuvre – Décembre 2005 – A37808/C"





## 3. Laboratory-scale application in field

### 3.1 Laboratory scale application

Phase 1 – test with different oxidant during 48h

Tests on soil mixed with groundwater samples collected at the site:

- Potassium permanganate
- Sodium persulfate:
  - Activated in alkaline conditions
  - Activated with hydrogen peroxide
- Fenton (hydrogen peroxide catalysed with iron under 3 different forms)
- Concentrations of oxidizing agent selected based on a stoichiometric approach and a SOD test

Total of 6 tests + 1 test as a reference

Following the phase 1, results indicated that the potassium permanganate was the most efficient and therefore selected for the phase 2.

Phase 2 – assessment of the concentrations and the dosage of the oxidizing agent :

Total of 4 tests: 2 doses x 2 concentrations during 48h

Results indicated that a high dose and a high concentration were optimal, especially on BTEX and C5-C10.

In phase 1 and 2, monitoring was conducted before and after the test – each jar was analysed for TPH C5-C10, TPH C10-C40 and BTEX.

In phase 2 colorimetric tests were also conducted at the end of the test.





## 4. Pilot-scale application in field

No pilot scale application in the field due the small size of the area to be treated

## 5. Full-scale application

#### 5.1 Main Reagent

The oxidizing agent was injected into the ground between 3.5 and 8m bgl using direct push technique (Geoprobe).

The injection was conducted in 2 successive phase:

- Phase 1: injection across all the impacted area: each injection point was around 1 apart (the radius of influence was estimated around 1m due to the low permeability of the soil this hypothesis was checked and confirmed at the beginning of the injection)
- Phase 2: injection in-between the injection points of the Phase 1 in the most contaminated area

The works for conducted over a period of 3 weeks.

Total of 83 injection points (over around 200m<sup>2</sup>) and of approximately 25m<sup>3</sup> of sodium permanganate at 20%.

Injection points were placed using a grid on a plan and in a staggered arrangement. The injection pressure was at the maximum of around 2 to 4 bars.

### 5.5 Process and performance monitoring

- Soil boring was conducted regularly to confirm the radius of influence of the
  injection points and that the oxidizing agent was diffusing homogeneously over
  the length of the injection (between around 4 and 8m bgl) controls were done
  visually as the permanganate has a violet colour.
- Groundwater in the piezometer at and around the treated area were also controlled – visual control as the permanganate has a violet colour and in the laboratory to measure the percentage of remaining oxidizing agent and to analyse TPH and BTEX.





## 6. Post treatment and/or Long Term Monitoring

#### 6.1 Post treatment and/or Long Term Monitoring

Monitoring of groundwater and soil gas over 2 years after the injection.

As soon as the colouration had disappeared, groundwater samples were tested for TPH C5-C10, TPH C10-C40

After 12 months, soil samples were taken from the treated area – results indicated a real improvement in soil quality. Soil results were used as an indicator for defining the success of the treatment.

Results indicated a reduction of 65% of the mass of the contaminants of concern.

#### 7. Additional information

#### 7.1 Lesson learnt

- In soil of low permeability, the colour may be retained for a longer period of time in the ground, however, often at very low percentage colorimetric tests are a very simple and good approach.
- In low permeability soil, the time for the ground/groundwater to find a new equilibrium after the injection can be very long (up to 24 months)

#### 7.2 Additional information

- Chemical processes / molecules are relatively well-known.
- A key success factor is the understanding of geological and hydrogeological conditions at the site and to some extend the geochemical conditions. This is the first things to consider when thinking about techniques to use and coming up with the best strategy for treating the impacted area.
- You have to control the volume/quantity of oxidizing agent that you are storing on the site during the treatment storing too much oxidizing agent may demand that you obtain a permit for doing so.

### 7.3 Training need





- Workshops are a good approach to exchange experiences and get the basic knowledge and tools to be able to face real situation
- On the job training to allow people to be confronted to real situation as there is a gap between the theory (what we can read in books and hear from others) and what is really happening in the field.

# **Glossary of Terms**

Term (alphabetical order)	Definition
BTEX	Benzene, toluene, ethylbenzene and xylenes
TPH	Total petroleum hydrocarbons
M bgl	Meter below ground level

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## 2. Site background

### 2.1 History of the site: Challenges and Solution

For several years, a manufacturing facility was in operation at a location near the city center of Uden, Netherlands. As a result of business activities at the site, soil and groundwater have been impacted with chlorinated hydrocarbons. Following demolition of the buildings in 2005, site investigations revealed high levels of contamination. In the groundwater aquifer, concentrations of more than 16,000  $\mu$ g/l of trichloroethylene (TRI) were measured, indicating the presence of a source zone (SZ). The impacted SZ is 270  $m^2$  and contaminated in the saturated zone from 3.0 to 7.0 meters below ground level. For the planned redevelopment of the site into a residential area, the local regulatory authorities mandated remediation of the contaminations to stringent clean-up target levels.



Site overview an location source zone

Following detailed Site Investigations (SI), as per standard operating procedures the first step was excavation of contaminated soils to the top of the groundwater level, and then backfilling the area with certificated clean soils. In view that the envisioned end-use by a real estate developer following the land transaction was construction of residential housing, rapid remedial results were required. As elements of the Remedial Options Appraisal (ROA) process, selection of a technological solution required high reliability, cost-effective implementation and quick results as key objectives. The engineering consultants conducted the SI and were involved with results verification, whereas the lead contractor was responsible for overall project management including technology selection, remediation design and implementation.





### 2.2 Geological and hydrogeological setting

The site is located in between two geological shear zones (Peelrandbreuk en Raambreuk) which mark the transition from the higher Peelhorst area (+20 meter above sea level) tot the lower Roerdalslenk area (+10 meter above sea level). The surface level at the location is about +16.5 meter above sea level (masl). The groundwater table is located at 3.0 meters below surface (mbs). The groundwater flow is in south-west direction with a hydraulic gradient of 0,002 m/m.

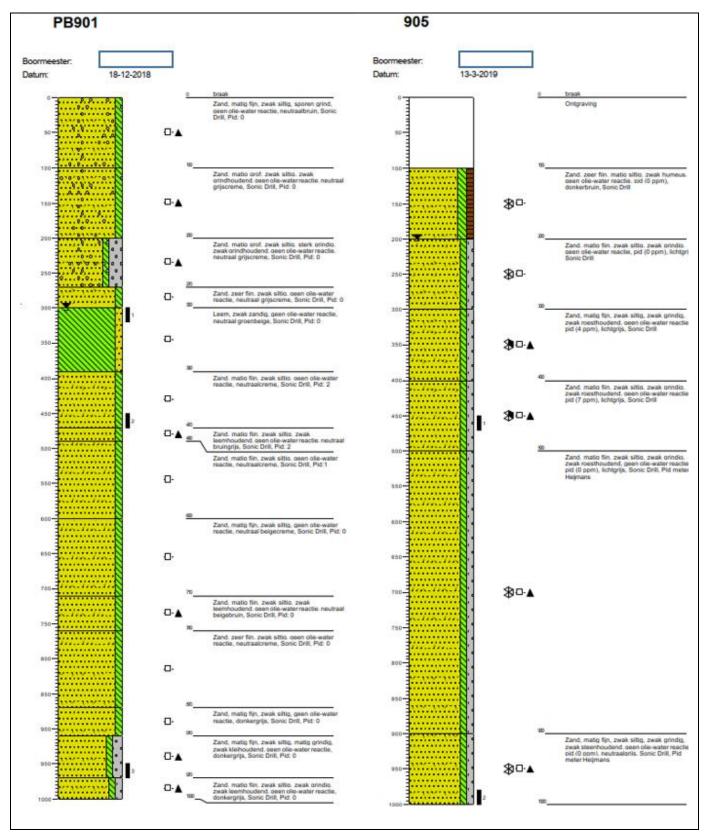


Drill core borehole (sonic drill)

Below the groundwater table until a depth of 16 mbs the site consists of medium fine to coarse sand. Locally some gravel layers and silty clay are present. The fraction organic carbon (Foc) is less than 0.5 %. The hydraulic conductivity varies between 2 and 20 m/day and the effective porosity is about 27.5%.











### 2.3 Contaminants of concern

The contaminants of concern are chlorinated solvents, especially trichloroethylene (TRI). Concentrations of more than 16,000  $\mu$ g/l of trichloroethylene (TRI) were measured in the saturated zone.

In the unsaturated zone more than 16,000 mg/kg of TRI was present.

A NAPL is not demonstrated and the soil contamination in the saturated zone is negligible.







### 2.4 Regulatory framework

The quality of the top 1 meter has to meet the standards for Maximale waarde Wonen (MWW). In the subsurface there should be a substantial removal of contamination in order to create a stable groundwater plume. Therefore the remediation goal for the saturated source zone is as follows:

Contaminant	Target value [mg/kg]	Target value [ μg/l]
TRI	0,5	2500
CIS	-	1000
VC	-	450

### 3. Laboratory-scale application in field

### 3.1 Laboratory scale application

The ISCO-remediation design was based on expert judgement. There was no time available to perform a batch test.

The Foc was assumed to be less than 0.5%.

The soil oxidant demand was assumed to be 3.0 g Persulfate/kg soil. Together with the amount of oxidant needed for the pollution, this resulted in an amount of 9,225 kg Persulfate (Klozur One) for the total source zone (safety factor 1.5).





### 4. Pilot-scale application in field

### 4.1 Main treatment strategy

The Klozur® One ISCO technology was selected primarily because it met all ROA objectives. The blend of sodium persulfate with built-in activation chemistry provided powerful oxidation capacity as a "ready to use" product suitable for this highly contaminated treatment area. A total of 9,225 Kg was required, delivered in 25 kg bags from a nearby warehouse, helping to keep the logistics carbon footprint low. As persulfate requires careful handling, the contractor took all necessary safety measures for storage and handling. Factors such as fire safety and unpredictable summer weather also played a role. From the storage facility the product was transport to an onsite mixing facility. There the bags were opened under controlled conditions, ensuring little physical contact between field technicians and the sodium persulfate. Special attention was focused on reducing the production of any dust particles.



The injections were made per batch, and in the injection plan there are several different concentration batches provided. A typical batch contained 4 m<sup>3</sup> of clean water into which a specified amount of Klozur One was added. From the mixing unit, the proper solution of Klozur One is transferred into the injection tank.





Volgorde		Batch Volume	Batch Aantal	Klozure-One	Aantal zakken	Batch conc.	Volume per Filter	Filters
aanmaak	-1	[Liter] -	[n] -	[kg/batch] -	[25 kg] 🔻	[g/l] -	[Liter] -	-
1		3700	3,00	350	14	94,6	2775	D1, D2, D3, D4
2		4500	4,80	200	8	44,4	3600	M1, M3, M7, M10, M11, M15
3		4500	4,00	200	8	44,4	3600	M2, M4, M8, M13, M14
4		4500	6,00	200	8	44,4	4500	01, 03, 07, 010, 014, 0116
5		4500	6,00	200	8	44,4	4500	02, 04, 06, 011, 013, 015
6		3600	4,00	350	14	97,2	3600	M5, M6, M9, M12
7		4500	4,00	425	17	94,4	4500	O5, O8, O9, O12
8		4000	4,50	200	8	50,0	3600	M16, M17, M18, M19, M20

Klozur One batching scheme

As each batch of injectable solution is mixed together, it is then applied to the subsurface through existing injection wells. In total, the contractor used 40 injection points at three different subsurface levels, in a grid pattern with a center-to-center distance of 5 meters (ROI of 2.5 meter)

With this grid, it was possible to engineer contact across the entire source area. At spots with higher concentrations of contaminant, more solution was applied with a higher concentration of persulfate. At each injection points between 2,775 and 4,500 liters of solution were applied. Through use of a manifold system, 4 to 6 wells were worked simultaneously, using a little overpressure to prevent blow-out at the surface. The sequence of the injections was performed from outside to inside located filters. In total, the field works lasted nine days to inject 155 m<sup>3</sup> injection fluid of sodium persulfate.



Overview injection filters in source zone







Detail injection well manifold





#### Results

Before the initiation of the injections there was an investigation of the TRI concentrations onsite. Monitoring activities during and after the injections including measurements of pH, oxygen, redox and electrical conductivity. Following the injections with sodium persulfate, there was a notable decrease in pH and increase in electrical conductivity visible. The contractor used Klozur Field Test Kits to determine an indication of the amount of active sodium persulfate still available.



Klozur Field test kits

The parameters were monitored weekly. After four weeks most of the active sodium persulfate was consumed, allowing the monitoring wells to be used for groundwater quality. In total, monitoring was conducted through 10 wells and in all of them the TRI concentration was decreased to below remediation targets. Four weeks later, an independent verification by the engineering consultants confirmed the positive results. They also concluded that there was no active sodium persulfate left and that the trichloroethylene was sufficiently removed.

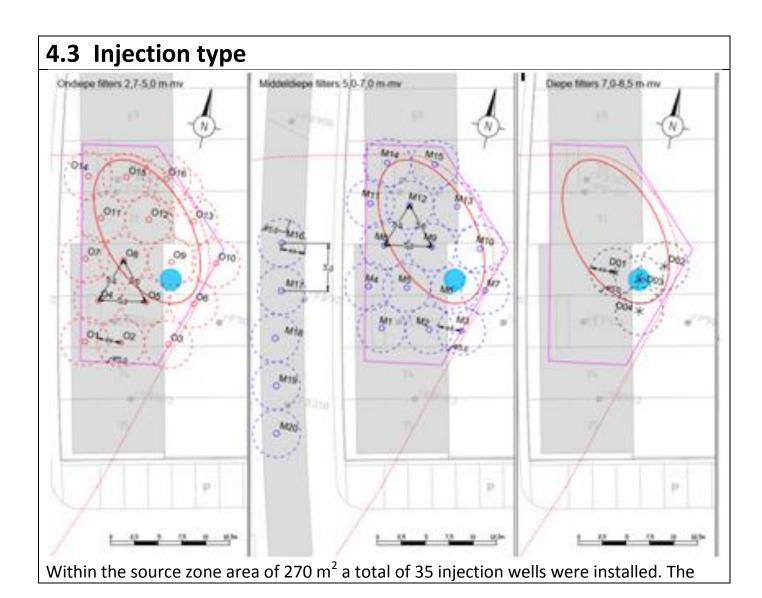




Monitoring well 2	Parame	eter	06-09-2019	24-09-2019	21-10-2019	21-11-2019
depth: 4,5 m below	PER	μg/l	<50	<1	<1	<1
ground level	TRI	μg/l	14.000	2,4	<1	<1
	CIS	μg/l	<50	1,4	1,4	1,4
	VC	μg/l	<100	<2	<2	<2
	Sodium	mg/l	13	n.a.	3.900	2.600
	Sulfate	mg/l	30	n.a.	1.300	1.400

Analysis overview of MW2 as representative data

In total the chemical oxidation removed 99.6% of the TRI pollution. With this good result we were able to close the active remediation phase.







injection wells in the source zone were spaced in grid formation with a distance of 5.0 meter. Downstream the source zone another 5 injection wells were installed in barrier formation. For the injection we installed new fixed injection wells with a diameter of  $\emptyset$ 50 and a screened length of 2 meter.

The injection wells were installed with a screened interval 2.7-5.0 mbs, 5.0-7.0 mbs and 7.0-8.5 mbs. In order to prevent preferential flow or blow-outs every injection filter had a fixed clay-stop was grouted with cement/benthonite up to the surface.



With this grid, it was possible to engineer contact across the entire source area. At spots with higher concentrations of contaminant, more solution was applied with a higher concentration of persulfate. At each injection points between 2,775 and 4,500 liters of solution were applied. Through use of a manifold system, 4 to 6 wells were worked simultaneously, using a little overpressure to prevent blow-out at the surface. The sequence of the injections was performed from outside to inside located filters. In total, the field works lasted nine days to inject 155 m<sup>3</sup> injection fluid of sodium persulfate.





### 4.4 Radius of influence

The radius of influence (ROI) was based on expert judgement. The actual injection radius of influence is monitored during the first injections. In this way, the ROI and the amount of injection volume for each injection filter was validated in the field.

Injection des	ign		
	Lengte	22	m
	Breedte	12,5	m
	Permeability	5	m/day
	Vw		m/month
	Interval	1	month
		0,91	m
	h.o.h.	5	m
	Design ROI	2,5	
	Injection ROI	2,05	m
	Design AOI	19,63	m2
	Injection ROI	13,14	m2
	overlap	15%	





#### 4.5 Control parameters

Before injection we monitored the natural field conditions in control monitoring wells:

• pH, temperature, dissolved O<sub>2</sub>, redox potential, electrical conductivity, Sodium, Sulfate, Chemicals of concern

During injection we monitored the dispersion in the field in monitoring filters:

- pH, temperature, dissolved O<sub>2</sub>, redox potential and electrical conductivity,
- injection pressure was monitored on each injection well

After injection we monitored the dispersion and contaminant in monitoring filters:

- pH, temperature, dissolved O<sub>2</sub>, redox potential and electrical conductivity,
- Klozur Field Test Kits were used to determine an indication of the amount of active sodium persulfate still available.

The parameters were monitored weekly. After four weeks most of the active sodium persulfate was consumed, allowing the monitoring wells to be used for groundwater quality. In total, monitoring was conducted through 10 wells and in all of them the TRI concentration was decreased to below remediation targets. Four weeks later, an independent verification by the engineering consultants confirmed the positive results.

### 5. Full-scale application

### 5.1 Main Reagent

The first injection round of injecting 9,225 kg of activated persulfate Klozur One (155 m<sup>3</sup> of solution) proved to be enough to reach the target values. No rebound occurred.

### 7. Additional information

#### 7.1 Lesson learnt

We had limited time to reach our target values (2 months). As we had a low % of organic matter, we chose to perform a full-scale pilot instead of a laboratory batch test. This way we determined the amount of oxidant needed in the field (first injection round).





Eventually we would have had time to inject a second time, but this wasn't necessary anymore as we reached the target values after the first injection.

This way we have saved time and money.

### 7.3 Training need

The human safety regulations and creating a safe working process for the operating personnel have to be taken into account when applying this technique.

This includes the whole cycle of storage of the oxidant, handling, dust control, mixing and finally controlled injection.

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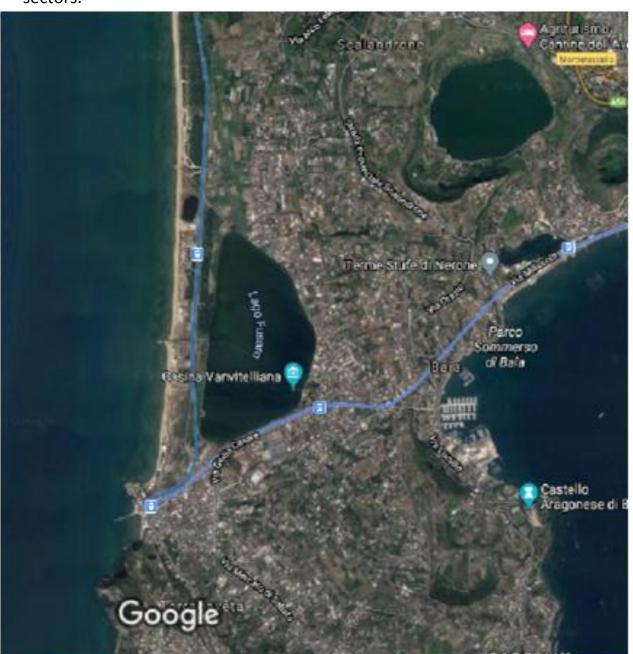




# 2. Site background

### 2.1 History of the site: Challenges and Solution

- The site is located in an area of medium population density and of important naturalistic / archaeological value.
- It is part of a former SIN
- The company operates and produces in the defence, aerospace and security sectors.







### 2.2 Geological and hydrogeological setting

In particular in the north-eastern portion of the site, where the ISCO technology was applied, the surface geological structure can be described by the following scheme:

- 0 0.60 m: ground floor including the underlying substrate composed of mixed inert material;
- 0.60 1.50 m: fill material composed of inert material mixed with a high permeability silty sand matrix;
- 1.50 4 m: fine sands, slightly clayey, of fluvio-lacustrine origin, high permeability;
- 4 9 m: coarse pumice and gray sands.

The water table has a depth ranging from about 1.5m to about 2 m from the ground level and is located in the alluvial lake deposits.

#### 2.3 Contaminants of concern

• Soils:

Hydrocarbons: 3500 mg / Kg

Groundwater

O Benzo(a)anthracene: 7.6 μg / L

Pyrene: 29 μg / L

Benzo(b)fluoranthene: 4.2 μg / L

Benzo(g,h,i)perylene: 2.2 μg / L

Polycyclic Aromatic Hydrocarbon (PAH sum): 10 μg / L

Tetrachlorethylene: 50 μg / L

Trichloroethylene: 5.4 μg / L

Vinyl chloride: 4.1 μg / L

Benzene: 27 μg / LXylene: 133 μg / L

Toluene: 22 μg / L

### 2.4 Regulatory framework

National Regulations (D.Lgs. 152/2006)





## 3. Laboratory-scale application in field

### 3.1 Laboratory scale application

Scope of lab test:

- determine the amount of oxidizing reagent (SOD), for the two different oxidizing compounds tested (sodium permanganate or percarbonate), necessary for the oxidation organic and inorganic pollutants present in the solid, liquid phase and in the saturated biphasic mixture.
- verify the reduction of pollutant concentrations using different stoichiometric ratios with respect to the SOD determined for each of the two oxidizer analyzed

#### 2. Lab scale test description:

The SOD determination tests were performed by preparing, for each reagent tested, 5 test tubes each containing an aliquot of 10 g of soil, kept stirred at 120 rpm at room temperature. The reagent solutions were added to the test tubes at three different concentrations. In order to verify reproducibility, the tests were performed in duplicate for each sample.

Reagent quantities for each test tube were calculated on the basis of samples TOC content and of similar experiences reported in the literature.

In total 6 tests were performed in duplicate at different stoichiometric ratios. The liquid / solid ratio used, based on literature reference data, was 3: 5.

During each test, lasting 8 days, the residual oxidant content was determined on a daily basis: for permanganate by means of a spectrophotometric absorption method at 520 nm, whereas for percarbonate by titration with permanganate. To evaluate the influence of the contamination on SOD, the determination of SOD was also carried out on a clean soil sample with the same procedure. Subsequent to the determination of the SOD, ISCO tests were carried out on soil saturated with groundwater using three different concentrations of oxidant in stoichiometric relation with respect to the SOD (ratios of 1: 1, 1: 3 and 1: 5) for two different times (24h and 72h). Consequently, 4 series of tests were carried out, one of which without the addition of oxidant, to check the quantities of pollutant volatilized in different test condition.

At the end of these tests, the oxidant residual quantity was determined, and in particular metals, C> 12 and PAHs were determined in the solid phase, whereas metals and chlorinated solvents were determined in the liquid phase.

The results of the tests conducted showed:

• In regards to the solid fraction: a marked reduction in total hydrocarbons C> 12, in





the case of using permanganate, even with a low stoichiometric ratio (1:1). The same efficiency was not achieved by percarbonate. A significant reduction in PAHs in the case of using permanganate with stoichiometric dosages greater than 1:3; the use of permanganate in a stoichiometric ratio 1:1 and percarbonate had instead shown unsatisfactory results.

With respect to the liquid fraction, the analytical results show: CrVI below the
instrumental detection limit after 72h of testing or at the end of the reaction
control period, either in the slurry where permanganate was used, and in those
where percarbonate has been used; complete oxidation of TCE and PCE when
using sodium permanganate.

On the basis of the tests carried out in the laboratory, it has been highlighted the necessity to provide a dosage of reactive, to reduce the pollutants present, much higher than the pure stoichiometric ratio between the moles of oxidant and those of pollutant.

### 4. Pilot-scale application in field

#### 4.1 Main treatment strategy

The laboratory tests showed that, due to the type of pollutants present, the most performing oxidant is permanganate, with percentages of pollutant reduction ranging from 40-50% up to about 90%.

The test consisted in the controlled injection of a solution consisting of:

- 2000 litres of industrial water
- 207.2 kg of sodium permanganate solution with a 40% concentration, corresponding to approximately 85 kg of permanganate.

The injection of the solution was carried on at a flow rate of about 15 l/min (0.9 m³/h) in order to minimize disturbance to the aquifer and avoid displacement of the contaminated water.





### 4.3 Injection type

To improve monitoring of the possible reagent downstream by migration, an additional control piezometer and a well (PE) were installed to recover any residual permanganate. Before the pilot scale application, in order to evaluate the migration routes of the injected solution, a test with fluorescein was performed. The test involved the controlled injection of a known volume and concentration solution (4000 liters of groundwater and about 0.4g of fluorescein), followed by monitoring of its propagation on a regular daily basis.

This test showed that despite the significant flow rates, the quantities of fluorescein recovered were equal just to approximately 30% of those injected, thus indicating minimal "migration" of the tracer.

The thickness of the saturated soil involved in the test was approximately 6.5m, from 1.5m up to approximately 8m below ground level. The pilot field consisted of: 1 injection well, 5 wells placed radially around the injection well, at a distance of 3, 5, 7 and 15m (internal control piezometers), and 6 external control piezometers/wells.

### 4.4 Radius of influence

The observation of the water colour in the piezometers adjacent to the injection point made it feasible to verify the solution distribution in the soil. The distinctive purple colour of the injected oxidant was found in the injection well alone, indicating that the reagent reacted completely before it could migrate downstream. Thus the reaction rate is higher than the rate of oxidant dispersion.

Field tests conducted by injecting an amount of sodium permanganate equal to 86 kg showed a radius of influence of less than 2 m, with a total consumption of the injected reagent over a few days.





#### 4.5 Control parameters

During field monitoring the following measurements were carried out:

- A check of the groundwater colour in all points of the cell for 3 days (72 hours);
- groundwater sampling in all the cell points for analysis of metals (Fe, Mn, Cr (III) and Cr (IV), As), chlorinated solvents, IPA, total hydrocarbons, BTEX and COD, CO<sub>2</sub>.
   During the sampling operations, the chemical-physical parameters were also measured after 1 day (T1), 10 days (T2), and 30 days (T3) from the injection.

The physic-chemical data collected during the sampling phases showed significant variations in the redox potential.

The chemical results showed an average percentage concentration reduction after 24 hours equal to 83%. In the following surveys (carried out after 8 days and after 1 month) the concentrations increased, but did not reach the values measured before the pilot test.

### 5. Full-scale application

### 5.1 Main Reagent

The laboratory tests and the pilot test highlighted the requirement for a higher dosage of permanganate than the dosage corresponding to the simple stoichiometric ratio, calculated with reference to SOD.

With permanganate, both the laboratory tests and the pilot field test showed a percentages of pollutant reduction of ranging from 40-50% up to about 90%.





### 5.3 Injection type

Considering the strong anisotropies, the most suitable and least impacting approach for the activities of the site involved injecting the oxidant mixture through a system of micro-perforations at different depths. In order to prevent reagent migration to the hydraulic barrier, wells had been created a few meters upstream of barrier itself, activated in the case of detection of unreacted permanganate (change of groundwater colour).

Considering the high consumption of oxidizer and the low radius of influence (less than 2m), in order to minimize the injection volumes per single point, 48 perforations were carried out (diameter of 127mm and a maximum depth of 7m) for the injection of the oxidizing compound. The drilling took place with continuous dry core drilling. In every perforation, two 1" PVC pipes were installed at different depths. The perforations were arranged along a regular 4 m side mesh with a thickening in the most impacted area, the distance between two injection points is about 2.5 m. A total of 48 injection clusters were created.

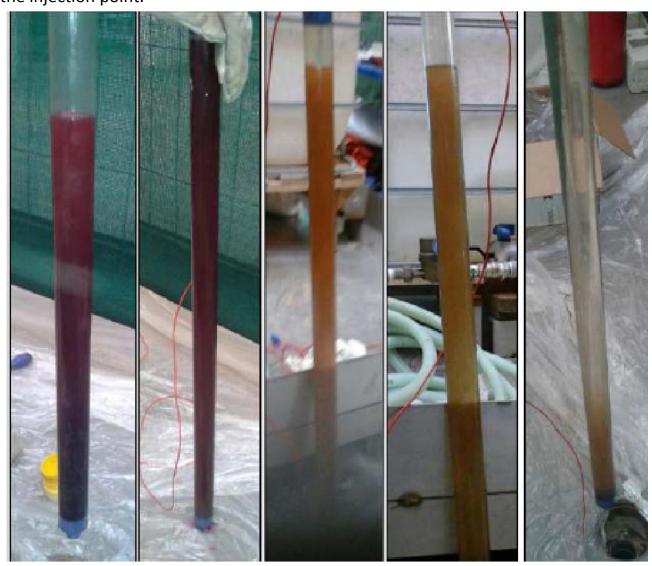






## 5.4 Radius of influence

Field tests enabled the estimation of a radius of influence of 2m or less with a total consumption of the injected reagent in few days as shown by water colour variation in the injection point.







#### 5.5 Process and performance monitoring

In addition to the 5 monitoring points of the pilot project phase, further 4 control points were installed.

- Before injections, in all the existing piezometers, the following parameters were measured at different depth using a multiparametric probe: temperature, dissolved oxygen, pH, conductivity, redox potential and salinity;
- During the injection phase all the piezometers were monitored in order to assess the propagation of the oxidation conditions;
- After the injections, all piezometers were monitored on a daily basis for the first 3 days in order to assess the propagation of the oxidation conditions following the injections.

## 6. Post treatment and/or Long Term Monitoring

### 6.1 Post treatment and/or Long Term Monitoring

After ISCO application all piezometers were monitored on a daily basis for the first 5 days, verifying oxidant traces and pollutants concentrations.

Then sampling surveys were carried out once a week for 1 month to check the content of: manganese, chlorinated solvents and polycyclic aromatic hydrocarbons

A long term monitoring was carried out to verify the fulfilment of remediation goals: Biannual monitoring of piezometers at quarterly frequency for the first year and then every six months.

The parameters analyzed during the biannual monitoring are: PAH, chlorinated solvents, BTEX, total hydrocarbons, Metals (Mn, Cr (VI), Cr (total)).

### 7. Additional information

#### 7.1 Lesson learnt

Presence of buildings or underground services was a limiting factor for the application of this remediation technique.





# 7.3 Training need

Training course on transport models in groundwater.

# **Glossary of Terms**

Term (alphabetical order)	Definition
PRB	Permeable Reactive Barrier
SOD	Soil Oxidant Demand
SIN	Sito di Interesse Nazionale (National Interest
	Megasite)

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## 2. Site background

### 2.1 History of the site: Challenges and Solution

The industrial site in guestion is located in Northern Italy within an area subject to archaeological and hydrogeological constraints, in the vicinity of an important surface water body (which passes 250 m downstream of the site). The site occupies an area of approximately 63,000 m<sup>2</sup>, of which approximately 50% is occupied by buildings (currently for non-productive use, but intended for the provision of services) and the remaining part used for parking and green areas.

The characterization highlighted a significant contamination by organohalogen solvents for the groundwater in the southern area of the site. This site corresponds to the area used for the storage of waste containing chlorinated solvents - used in degreasing and









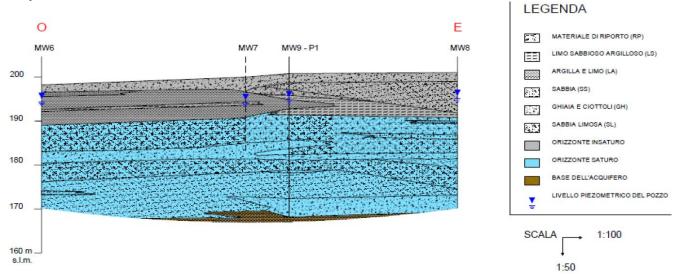
There are no specific protocols for the management of this site, but the control and technical evaluation activities in support of the Municipality (proceeding administration appointed by the Region for the management of contaminated sites) are carried out by ARPA. ARPA is the Environmental Protection Agency established in 1999 that deals with the prevention and protection of the environment, supporting regional and local institutions in multiple activities including:

- atmospheric pollution;
- noise pollution;
- water protection for surface water and ground water
- monitoring electromagnetic fields
- investigations on soil contamination and remediation processes.

The remediation activities through injection were carried out in the period between the 4th till 21st September 2012.

### 2.2 Geological and hydrogeological setting

The investigations carried out at the site revealed the presence of a semi-confined aquifer with a silty clay substrate (with low permeability at about 30-32 m from ground level and groundwater level at -6m. In the source of contamination area, which was subjected to ISCO, the saturated matrix had a thickness from -12m to -37m.



The local outflow of the aquifer is E-W in the portion of the site involved in the intervention. The average hydraulic gradient is equal to 0.5% and the permeability varies from  $4.4 \times 10^{-5}$  m/sec to  $5.5 \times 10^{-5}$  m/sec, the flow rate of the groundwater has been calculated equal to about 40 m/year.





In order to acquire more detailed information for the preparation of the remediation project, regarding the extent of contamination, an investigation was carried out using MIP (Membrane Interface Probe) consisting of 10 drilling points pushed to a depth of 35 m below ground surface thus being able to evaluate an area of about 175 m<sup>2</sup> around the MW8 piezometer.

#### 2.3 Contaminants of concern

The environmental characterisation study was performed in 2001, from which, chlorinated solvents with concentrations in groundwater equal to approximately 3,000  $\mu g/L$  were identified. The main contaminants detected were, in the order of concentrations found, PCE and, alternatively, TCE, 1.1 dichlorethylene. There was no evidence of the presence of the free product (DNAPL for chlorinated products) which is denser than the water to be sought at the base of monitoring piezometers. Unsaturated soils in the same area did not show contaminant values higher than the CMA (Maximum Permissible Concentrations) established by the then current Ministerial Decree 471/99, also because it had been the subject of an EVS intervention.





### 2.4 Regulatory framework

The proceeding was conducted according to Ministerial Decree 471/99 as the proceeding was initiated in 2001, before the entry into force of Legislative Decree 152/2006.

Following the finding of values higher than the CMA, the preliminary remediation Project was presented to define all the suitable and economic sustainable remediation methodologies useful for the site. The EVS intervention was selected for the unsaturated and a direct oxidation technology in situ (ISCO) with KMnO<sub>4</sub> for the saturated. This also involved the execution of appropriate laboratory tests and a pilot test in situ.

An emergency safety intervention was also carried out on the aquifer, through the construction of a hydraulic barrier to avoid the migration of contaminants downstream.

The ISCO treatment was performed in accordance with the technical indications provided in Protocol No. 28220 of 20/07/2005 prepared by APAT (now ISPRA) for the application of chemical oxidation in situ.





### 3. Laboratory-scale application in field

### 3.1 Laboratory scale application

The purpose of the tests was to evaluate the PNOD (Permanganate Natural Oxygen Demand) parameter which represents the natural oxidant requirement for permanganate; i.e. the amount of permanganate necessary for the oxidation of organic and inorganic compounds naturally present in the soil.

Four soil samples and two groundwater samples were taken from the source area, for the batch tests carried out independently by the operator. PNOD was found to vary between 1 and 7 kg  $KMnO_4/m^3$  soil as a function of the depth of the soil.

For the design, an average concentration of organohalogen compounds equal to 2 mg/l was considered; the stoichiometric KMnO<sub>4</sub>/contaminated ratio of 3.





# 4. Pilot-scale application in field

### 4.1 Main treatment strategy

In choosing the remediation technology, due consideration was placed for the presence of low permeability horizons that made more traditional techniques, such as Air Sparging, unfeasible.

Pharmaceutical grade Potassium permanganate (KMnO<sub>4</sub>) was used in a 3% solution, with a maximum content of impurities such as to allow the injection of a solution that complies with the quality requirements of the Ministerial Decree 471/99 with the obvious exception of the manganese parameter.

For permanganate (sodium or potassium), the half-reaction of reduction in the typical conditions of the subsoil with pH between 3.5 and 12 is as follows:

 $MnO_4^- + 2H_2O + 3e^- \rightarrow MnO_2 (s) + 4OH^-$ 

The manganese dioxide MnO<sub>2</sub> that is formed is an insoluble solid that is even used as a filter medium for the reduction of manganese concentrations from groundwater, therefore non-toxic and already known in remediation procedures. Manganese dioxide precipitates as a particle or as a colloid. As a result, the application of permanganate, at the end of the oxidation reactions, does not result in an increase in the concentrations of dissolved manganese.

Below are the oxidation reactions of the two main contaminants found in the groundwater:

Percloroetilene (PCE):

•  $4KMnO_4 + 3C_2CI_4 + 4H_2O \rightarrow 6CO_2 + 4MnO_2 + 4K^+ + 8H^+ + 12CI^-$ 

Tricloroetilene (TCE):

 $\bullet \hspace{0.5cm} 2\mathsf{KMnO_4} + \mathsf{C_2HCl_3} \rightarrow 2\mathsf{CO_2} + 2\mathsf{MnO_2} + 2\mathsf{K}^+ + \mathsf{H}^+ + 3\mathsf{Cl}^-$ 

Before the injection, a zero-time monitoring campaign was carried out, at T0, to be considered as a reference before carrying out the ISCO injections.

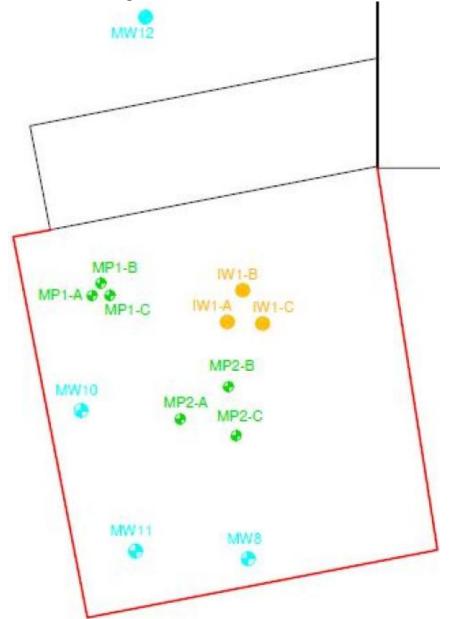




### 4.3 Injection type

The 2004 field pilot test was conducted with injection wells to allow greater flexibility during injections and sampling.

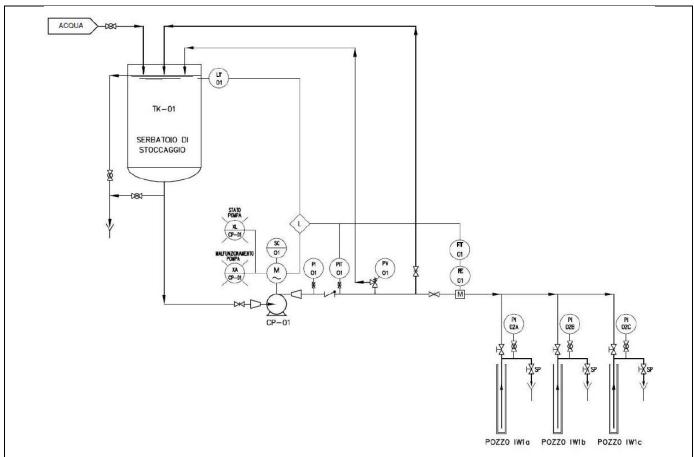
Three PVC injection wells with a diameter of 3"(IW1-A, IW1-B and IW1-C) made *ad hoc*, located at the vertices of a triangle, slotted respectively (slot 1 mm) in the following intervals: 25 -30m, 18-25m, 12-18m. In addition, three monitoring wells (MP2-A, MP2-B and MP2-C) were made with slits in the same intervals as those of injection. A plan of the pilot plant is shown in the figure below.



350 kg of permanganate were injected on the deep and intermediate intervals and 375 kg on the surface, in a single campaign, using the structures outlined below.







Finally, a photograph of the pilot plant is shown in order to demonstrate the scarcity of impact, compatible with an activity in operation.







### 4.5 Control parameters

The control parameters concerned the monitoring of the compounds of interest of any oxidation by-products and the recording of physical parameters with a multiparametric probe, with particular attention to the redox potential and conductivity. In general, the concentrations of organohalogen compounds rapidly decreased, even below the detection limit, and then sometimes increased again, usually to much lower values than the initial starting concentration, in the latest monitoring campaigns. This phenomenon can be explained by the spatial and temporal limitation of the intervention which had evidently not completely eliminated the secondary source of contamination in the soil (as confirmed by the preliminary MIP investigations). The most relevant PCE concentrations remained confined to downstream-flow control piezometers. Concentrations of TCE generally decreased, albeit to a lesser extent than PCE. In the triplet of injection piezometers, the redox potential remained stabilized around 500 mV. The conductivity values initially increased at all points, with values of the order of 10^3  $\mu$ S/cm at the injection points.

During the pilot test, no accumulations of organohalogen compounds with a low number of chlorine atoms (dichlorethylene and vinyl chloride monomer) or of other secondary organohalogen compounds were observed. This indicated that the oxidation of the organohalogen compounds was complete and that there was no risk of accumulation of compounds with a lower number of chlorine atoms.





### 5. Full-scale application

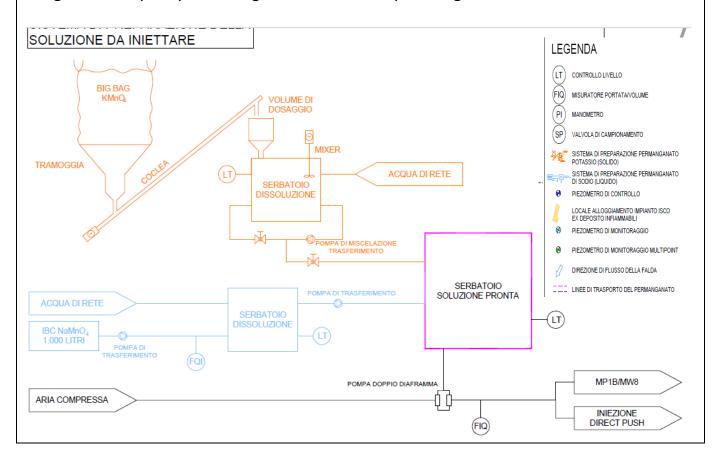
### 5.1 Main Reagent

NaMnO<sub>4</sub> was used as an alternative to the KMnO<sub>4</sub> used during the pilot scale test for operational needs, as it is more cost-effective, more soluble and with the advantage of using smaller injection volumes.

The reactivity of the two species is identical as the active ion is always the permanganate ion, with the only change being the dosage for the different molar weights.

The criteria for selection between the two salts was based on the greater ease of using a liquid instead of a solid and on the difference in cost. It should also be noted that the use of a solution presents fewer health and safety problems as the handling takes place entirely in the liquid phase without the emission of dust.

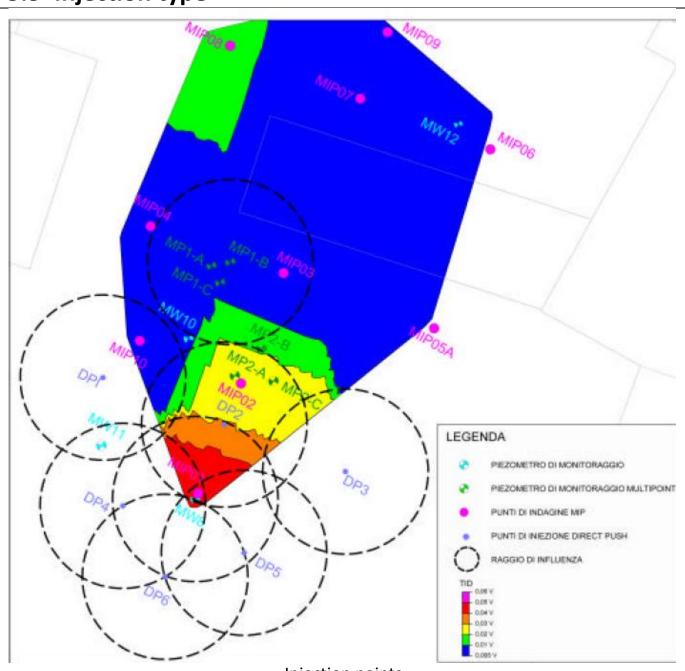
Below is a diagram of the injection systems of the oxidizing reagents which highlights the greater simplicity of management of sodium permanganate.







## 5.3 Injection type



Injection points

The treatment, from the Reclamation Plan (PdB) included:

 The construction of 6 injection points in the area with the greatest contamination (around MW8), identified by the initials DP1 ÷ DP6, with permanganate injection with direct-push technology, at depths between 12 m and 37 m below ground surface;





- direct injection of permanganate into existing MW8 and MP1b piezometers;
- construction of an ad hoc monitoring piezometer (MW13) located downstream of the area subjected to reclamation, equipped with a barrier well, in compliance with the indications of the APAT 2005 Protocol, to be activated in the event of the presence of unreacted permanganate, with re-entry of the same in the MW12 piezometer located upstream of the treated area, creating a closed circuit that also acted as a barrier.

A continuous monitoring system was installed on this piezometer, consisting of a parametric probe aimed at determining the redox potential, associated with an alarm system that would allow, in the event of an anomaly or a potential leakage of the oxidizing agent, the immediate activation of the pumping activity.

Considering the need to inject at different depths, on considerable thicknesses with volumes of complex geometry, the "direct push" methodology was used, which allowed better dosing of the reagents using closer injection points with lower costs than those of injection wells.

A Geoprobe type probe was used, through injections in the 3 intervals -12-16 m; -16-25 m; -25-37m from p.c., also monitoring of the volumes injected was carried out. The volume of land to be treated, at the design level, was estimated to be 4,648 m<sup>3</sup>, equivalent to approximately 7900 t, for which a quantity of KMnO<sub>4</sub> equal to 17973 kg was used, considering all the organic substances present in the soil on the basis of laboratory tests.

In addition to the piezometers from the PdB, an additional injection point (DP7) was also created for the injection of permanganate and the MW10 piezometer was also used, due to the poor filtering capacity of the piezometers which tended to disperse the reagent very slowly, slowing down injection operations.

We proceeded with a first dose of 20207 kg of NaMnO<sub>4</sub> equal to 9000 kg of permanganate (50% of the requirement), reserving the right to integrate this requirement later; being in solution at 40% by weight, this mass corresponded to an overall volume to be injected equal to 128  $\text{m}^3$  of solution.

Compared to the design data, the volumes of injected permanganate have been modified, mainly due to the lithological nature, represented by very compact silt in the deeper horizons.

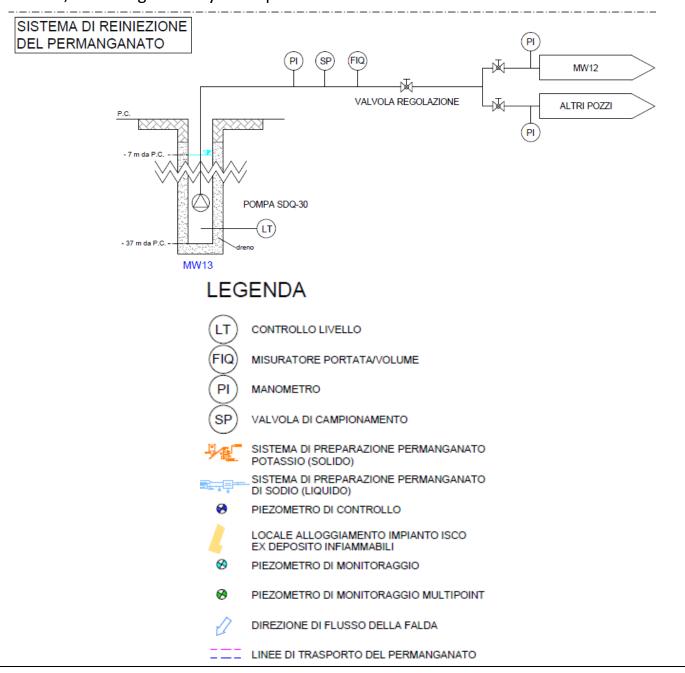
The injection pressure was always lower than 6 bar, thus avoiding macro-fracturing of the aquifer.

In order to control the possible migration of groundwater containing unreacted permanganate downstream from the hydrogeological area of the intervention area, a





control piezometer (MW13) was created which was pumped when the water present in the piezometer itself showed a violet colour (index of the presence of unreacted permanganate); the extraction was interrupted when the presence of permanganate was no longer visually detected and in any case before the test was carried out. The water extracted from MW13 was re-injected into the monitoring wells present inside the intervention area (in MW12 upstream of MW13) in order to fully exploit the extracted reagent and to create a dynamic treatment cell improving the distribution of oxidant, according to the system specified below.







#### 5.4 Radius of influence

A range of influence of 3.5 m was defined on the basis of the pilot test.

#### 5.5 Process and performance monitoring

The monitoring consisted during the ISCO injection in the measurement of physical parameters with a multiparametric probe, namely: redox potential, dissolved oxygen, conductivity, pH and temperature twice a day. In particular, in the MW13 control piezometer, the continuous measurement of conductivity was provided to evaluate any permanganate leakage.

The first check in terms of chemical analysis of the compounds of interest was performed one week after the end of the injections.

Below is a summary of the monitoring carried out:

Deadline	MW13	pH, redox,	Chemical
	(colour and	conductivity,	analyses
	redox)	DO, T°, colour	
Before the	NO	YES (a)	YES (a)
injection			
End of injection	YES	YES (b)	NO
T0			
1 week from T0	YES + analysis	YES (b)	NO
2 weeks from T0	YES	YES (b)	NO
3 weeks from T0	YES	YES (b)	NO
1 month from T0	YES + analysis	YES (b)	YES (b) + MW1
6 weeks from T0	YES	YES (b)	NO
2 months from T0	YES + analysis	YES (b)	YES (b) + MW1
3 months from T0	YES + analysis	YES (b)	YES (b) + MW1
4 months from T0-	YES	YES (a)	YES (a)
TESTING			

<sup>(</sup>a) Complete piezometric network: MW1, MW2, MW3, MW4, MW5, MW6, MW7, MW8, MW9, MW10, MW11, MP1-B, MP1-C, P1, EW1

The groundwater samples taken during the monitoring were subjected to the determination of organochlorine solvents, manganese and the following metals: Cd, Cr

<sup>(</sup>b) Reduced piezometric network: MW7, MW8, MW10, MW11, MW12, MP1-B, MP1-C and P1





VI, Fe, Cu, Pb, Zn. For the MW13 piezometer alone, the permanganate ion concentration was also determined.

The monitoring plan provided that, if the project objectives were achieved four months after the remediation intervention, post-operam monitoring would be activated; alternatively a second injection session would have been performed maintaining the same monitoring protocol as above, which was not necessary.

Due to an "anomalous" PCE value found on the expected date of testing on the MW4 piezometer upstream of the intervention area as well as the persistence of the purple colour inside the source area, testing was postponed to the next sampling, but also in this circumstance it was ascertained the persistence of the violet colour on the MW8 and MW11 piezometers inside the source area.

In the subsequent monitoring campaigns this criticality no longer emerged and the achievement of the remediation objectives for the organohalogen solvents for all the monitored points was verified.





# 6. Post treatment and/or Long Term Monitoring

## 6.1 Post treatment and/or Long Term Monitoring

From the end of the testing, post-construction monitoring was carried out for six years, starting from May 2013, according to the specifications shown in the table.

	Sampling	Frequency	Follow up actions
1st year	Complete piezometric network	quarterly	If compliant with the Italian threshold limits (CSC-CSR): shutdown of P1 (barrier well)
2nd year	Complete piezometric network	quarterly	
3rd year	Complete piezometric network	half-yearly	
4th year	Complete piezometric network	half-yearly	
5th year	Complete piezometric network	half-yearly	
6th year	Complete piezometric network	half-yearly	

The groundwater samples taken during post-construction monitoring involved the determination of organochlorine solvents and metals (Mg, Cd, Cr VI, Fe, Cu, Pb and Zn) only.

The last campaign carried out showed significant reductions in Mn, indicating that the permanganate had completely reacted in all the monitoring piezometers. Barrier well P1 was shut down in July 2016.





### 7. Additional information

#### 7.1 Lesson learnt

In general, ISCO offers the following advantages:

- ability to quickly treat a wide range of organic contaminants;
- allows you to set up temporary construction sites of limited size;
- It is particularly suitable for aliphatic compounds that chlorinate in not excessively fine horizons, to avoid the risk of rebound.

As a case-specific criticality, the presence of unreacted permanganate in the injection area was highlighted and therefore the barrier well was kept in operation until the injected permanganate was used up, as well as the maintenance of CSCs at the point of compliance. On the basis of the pilot test performed on site, however, the consumption of the injected product had occurred completely.

The lithological nature of the area subjected to injection, represented by very compact silt in the deeper horizons, has presumably influenced the distribution of permanganate, greatly slowing down its degradation. On the other hand, the failure to detect permanganate in the MW13 spy piezometer, located immediately downstream of the injection area, confirmed the poor mobility of the product due to the low permeability of the soil.

It is also noted that, from the analysis of the results of the post-operam monitoring, it is observed that with the exception of the MW8 piezometer, in which it is possible to appreciate the effectiveness of the intervention with total abatement of organohalogen solvents, for the other monitored points located in the area source (MW10, MP1B, MP1C) the concentrations of some halogenated solvents after total abatement in the first 4 months from injection gradually increased, settling on values around 15-20 ppm. This result is difficult to explain, especially if associated with the presence of unreacted product in the same points.

In relation to the monitoring of metals, the analysis of the analytical data showed a significant increase in the concentrations of Mn and lower increase of Fe in the piezometers located in the source area. No significant variations in the concentrations of metals before and after the intervention (therefore correlable to ISCO) for the other monitored piezometers were noted.





### 7.2 Additional information

The remediation objectives consisted of achieving concentration values below the Italian risk threshold (CSRs) for all the piezometers inside the site, defined by applying the site-specific risk analysis and coinciding with the contamination threshold concentrations (CSCs), i.e. the table limits conformity verification) for the MW7 piezometer placed at the site boundary, in the hydrogeological valley position, as specified in the table:

Parameter	CSR (µg/L) for all piezometers inside the site, from Risk Assessment evaluation	CSR = CSC (μg/L)  For MW7 or site compliance point	
	with reference to the "inhalation" path		
PCE tetrachlorethylene	97	1.1	
Trichloroethylene TCE	440	1.5	
1,1 dichlorethylene	6.8	0.05	
Cis-1,2 dichlorethylene	16000	60	
1,2 dichloropropane	87	0.15	
1,1,2 trichloroethane	220	0.2	
Vinyl chloride	38	0.5	

### 7.4 Additional remarks

With regard to the main limitations of this technology, it should be noted that:

- T here is a need to ensure a physical or hydraulic barrier / margin protection system downstream of the treatment, in order to evaluate any leakage of the oxidizing agent outside the site or to avoid any migration phenomena of the reaction by-products towards sensitive targets (also in compliance with the 2005 APAT protocol mentioned above);
- Very strong oxidants can be corrosive and potentially explosive therefore particular attention must be paid to health & safety consideration;
- The effectiveness of the process is influenced by the presence of heterogeneity of





the subsoil or by the poor mixing of the reagent in the groundwater;

- In certain cases, in areas difficult to access to the reagent, such as fine materials, the occurrence of rebound phenomena is noted. Consequently it is necessary to proceed with further injection cycles;
- Some reactants can be consumed by other oxidizable substrates present in the subsoil, thus limiting the effectiveness of the treatment;
- The use of permanganate could cause temporary increases in manganese concentrations and the precipitation of manganese oxides.

## **Glossary of Terms**

Term (alphabetical order)	Definition
CMA	Maximum Permissible Concentrations
CSC	contamination threshold concentrations
CSR	Risk Threshold Concentrations
D.Lgs.	Legislative decree
D.M.	Ministerial decree
MIP	Membrane Interface Probe
PdB	Remediation Plan
PNOD	Permanganate Natural Oxygen Demand
SVE	Soil vapour extraction

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# 2. Site background

## 2.1 History of the site: Challenges and Solution

From 1945 through 1983 "processing" of used chlorinated hydrocarbons took place on the subject area (former garage shop). Initial site investigations started in 1984. A six-year-long pump & treat remediation ceased in 2006.

Due to continuously high groundwater concentrations of PCE/TCE of up to 200,000  $\mu$ g/L remediation was necessary.



Because parts of the subsurface contamination are located below a main road (see picture below)with services including a sewer, gas pipeline, telecommunications only in-situ remediation technologies were deemed feasible for remediation of this specific sub area. A variety of methods were evaluated including thermal but from a cost-benefit viewpoint was ISCO using hydraulic fracturing as the preferred method.







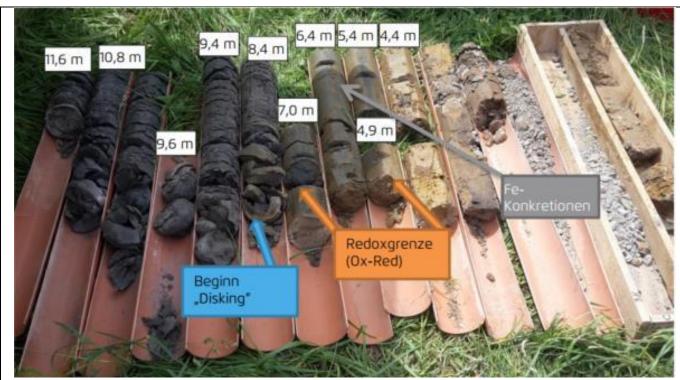
## 2.2 Geological and hydrogeological setting

The site is underlain by fill and Quaternary loess to a depth of 4 m below ground surface (bgs), which is overlaying a clayey silt layer followed by some silty clay layer of weathered marlstone to a depth of 7.5 m containing some perched water which is followed by the more competent, naturally fissured marlstone (Lias  $\beta$ ). The marlstone reaches down to at least 12 m bgs and serves as a confined low permeability dual porosity "aquifer" with a groundwater flow of only 5 m / month mainly occurring in the fissures. The permeability of the weathered marlstone clay is 7 x 10<sup>-9</sup> m/sec and marlstone exhibits a permeability of about 5 x 10<sup>-7</sup> m/sec.

A redox boundary formed at a depth of around 6 m bgs (see picture below).







During the investigation phase, it was noted that the soil exhibited spots of high concentrations with neighbouring spots of low concentrations. The only interpretation at the time concluded that the subsurface is heterogeneous.

As a consequence, it was conducted research into the depositional environment followed by statistical analysis of the soil data including the type of clays. It could be shown that the natural heterogeneity based on TOC; Fe, Mn, Al, and NOD analyses had only a variation of  $\pm 20\%$ . This was much lower than the contaminant data variation which exceeded  $\pm 140\%$ . It was found that secondary disking structures were formed post-depositional and as a result of Tertiary and Quaternary overburden weathering. Post deposition and thereafter a vertical fracture network developed (as shown today in the marlstone) which subsequently partially healed as shown in the overlying tight weathered silty clays. This narrow spaced natural fracture network (fracture distance 0.2 m to about 1 m) was the pathway for contaminates to enter the subsurface to a depth of at least 10 m bgs. Vertical analytical transport modelling using a spreadsheet software proved this hypothesis.

From the CPTU data it was concluded that the soil contained some perched water to a depth of 4 m bgs. Below that depth the soil exhibited a pore water suction potential between -0.04 to -0.09 MPa to a depth of 12 m bgs. This fact had the potential of limiting the ISCO application significantly. Further research showed that for the reported suction potential enough water is present for sufficient diffusion of the oxidative front emanating from the permanganate and persulfate agents.





#### 2.3 Contaminants of concern

Results of investigations in 2006 identified CVOC soil concentrations of the weathered clay of up to 75 mg/kg. Subsequent MIP-and CPTU investigations (pre RiskCom's involvement) including liner sampling provided a more detailed picture of the contamination and provided relevant geotechnical data in order to reliably plan the injection using hydraulic fracturing. Significantly higher CVOC soil concentrations of > 6,000 mg/kg were analysed during this campaign.

Groundwater samples indicated extreme concentrations of up to 447,000  $\mu$ g/L total CVOC (on average about 150,000  $\mu$ g/L) and up to 6,200  $\mu$ g/L BTEX.

### 2.4 Regulatory framework

Due to continuously high groundwater concentrations of PCE/TCE of up to 200,000  $\mu$ g/L, a remedial order was instigated.

The remediation plan focused on achieving a reasonable groundwater quality. Hence, a maximum CVOC discharge rate (i.e. mass flux) was prescribed. The prescribed goal is to reduce the contaminant mass with proportional means to such an extent that the long term total CVOC emission via the groundwater path is below 1 kg/a. An initial remediation target value for soil was 100 mg/kg total CVOC.

The competent Authorities were well satisfied with the method of hydraulic stimulation and the injection of 6 tons of permanganate and persulfate as solids was approved for the pilot test.





## 3. Laboratory-scale application in field

## 3.1 Laboratory scale application

Several lab tests for determination of a stoichiometric oxidant demand were conducted.

- SOD1 test on four samples with permanganate, and persulfate
- SOD2 batch tests on four soil samples before the injection with ground and intact soil samples from the clayey silt layer and the weathered marlstone for a period of 28 days
- SOD2 batch tests on ten soil samples from liner bores of the clayey silt layer and the weathered marlstone after the injection.

TIC and TOC as well as Fe and Mn were determined. With that data we were able to determine the oxidation state of the organic matter (OC) to +2.1 on average. With this evaluation the stoichiometric demand could be determined much better than using the standard methodology.

From the test results before the injection and after the injection the effectiveness of the oxidation was determined for the fraction of organic carbon and the CVOCs. It was determined that the clayey silt layer had a permanganate oxidant demand of 59 g/kg and the weathered marlstone had a permanganate oxidant demand of 78 g/kg. The SOD fast portion consisting of mainly Fe and OC required about 54% of the total SOD.

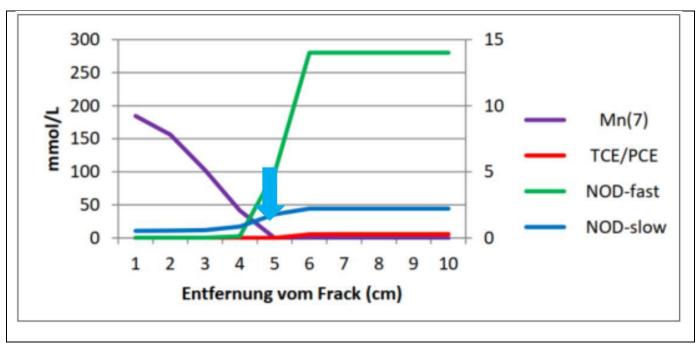
It could be shown from a detailed evaluation of the CPTU data that the suction potential is still in a range where saturated diffusion occurs. Henceforth, further kinetic parameters from the SOD2 tests were derived and initially an analytical kinetic diffusion model was developed and run. Later a numerical diffusion model ("quasi 2D") was run using CVOC input concentrations between 100 mg/kg to 2,000 mg/kg. It could be shown that PCE and TCE are faster oxidised than the SOD fast. SOD slow was much slower than the SOD fast which resulted in a 5 cm diffusion front for permanganate even for the 2,000 mg/kg CVOC concentration.

Consequently, the vertical distance between the hydraulically emplaced and permanganate laden fractures had to be in a 10 cm distance for a complete oxidation of the soil profile.

This distance of 10 cm was smaller than the distance originally chosen in the pilot test. Nevertheless, the result indicated that a complete remediation can be achieved if the full-scale application is conducted.







# 4. Pilot-scale application in field

### 4.1 Main treatment strategy

Due to the low permeability soil at the site hydraulic fracturing as injection technique was selected for the subsurface contamination below the main road, which was the preselected location for the pilot test. For the pilot test only one injection borehole was drilled. Remediation reagents were injected in the main contaminated area between 6.2 m and 10.5 m depth in a vertical distance of mainly 0.15 m.

Due to the low permeability clay and marlstone and the limited advective groundwater flow in the saturated zone as well as the dry conditions in the unsaturated zone, the injection of a combination of potassium permanganate ( $KMnO_4$ ) and sodium persulfate ( $Na_2S_2O_8$ ) was planned for the pilot test.

 $KMnO_4$  was selected due to its fast reaction kinetics for the CHCs, its high diffusion coefficient, and its lack of interference with hydrogen carbonate ions.  $Na_2S_2O_8$  was selected due to its low solubility and the long persistence as well as it's reported (e.g. Siegrist et al., 2011) lower tendency to oxidize the NOM (natural organic matter). It was intended to inject a 50%-mix of the reagents.

After the injection of each 1.6 t  $KMnO_4$  and  $Na_2S_2O_8$  between 6.2 m and 7.55 m a slight uplift of the road and a slight widening of an existing crack in the road were observed. It was concluded that the heave was attributable to a spontaneous gas formation. The gas





formation was generated through the addition of  $Na_2S_2O_8$  and the addition of 10% NaOH as activator. A post evaluation of the reaction kinetics showed that the amount of bivalent iron available in the soil would have been sufficient for the activation of  $Na_2S_2O_8$ . In this case the alkaline activation was excessive and unnecessary. As a consequence of the crack widening, the injection of  $Na_2S_2O_8$  was immediately ceased. The subsequent injection was carried out with  $KMnO_4$  only. At a depth of 8.8 m  $Na_2S_2O_8$  was again injected, however without the addition of NaOH. The uplift of the road and the widening of the crack stopped and declined immediately after the amended reagent formula was applied.

Originally it was planned to inject 5.3 t of oxidising agents including the gelling agent and activators. Due to the amended reagent formula as a result of the gas formation only 3.4 t of reagents were injected.

The schedule for the fieldwork was extremely tight since the injection borehole was located in the middle of a main road which was blocked for bus and public traffic for only two weeks.

#### 4.2 Additives

Guar Gum was selected as a gelling agent and viscosifier.

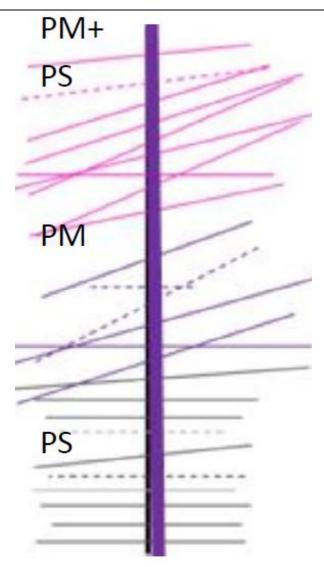
10% NaOH was added as activator.

### 4.3 Injection type

Remediation reagents were injected under pressure (hydraulic fracturing) in the main contaminated area between 6.2 m and 10.5 m depth via one injection borehole using the direct push system in one campaign. A total of 2.53 t of solid reagents ( $Na_2S_2O_8$ , "PS" and  $KMnO_4$ , "PM") without additives were injected.







From 6 m bgs to 7.5 m bgs a mixture of solid permanganate and persulfate was injected. This was probably the first time when both agents were injected simultaneously. Thereafter, only solid permanganate was injected. The loading ranged from 150 to 250 kg per frac. From 9 m depth onwards only persulfate solution (50% concentration) was injected.

Our evaluation concluded that the reaction kinetics of persulfate and permanganate reached similar oxidation effects. However, the necessity of persulfate activation and its presumably lower diffusivity added additional complexity.

Mainly horizontal injection layers were generated in a vertical distance of mainly 0.15 m (see picture). Spatial monitoring of the artificially generated fractures was done using tiltmeters which were placed on the road's surface. A live evaluation of generated tiltmeter data allowed the on-site determination of each fracture with respect to its dip and strike. Later evaluation allowed the determination of the fracture thickness and lateral extent.





Measurements of the groundwater potential at three groundwater monitoring wells located around the injection borehole, indicated that existing fissures were (re)activated and thereby filled with reagents. This was also proven from real-time monitoring of the injection pressure data.

It could also be shown, that the generation of the 25 fractures increased the permeability at the area affected by the pilot test, which means that the groundwater flow locally (horizontally and vertically) became faster, which in turn positively influenced the distribution of the emplaced reagent afterwards.

It could be analytically proven that 25% of the injected KMnO<sub>4</sub> was still available in the subsoil two months after the injection. After nearly two years of monitoring it could be shown that a one-time injection of the remedial agents was enough to reach the remediation goals in the pilot test area.

#### 4.4 Radius of influence

The following calculations are based on the evaluation of the tiltmeter data (also see next section):

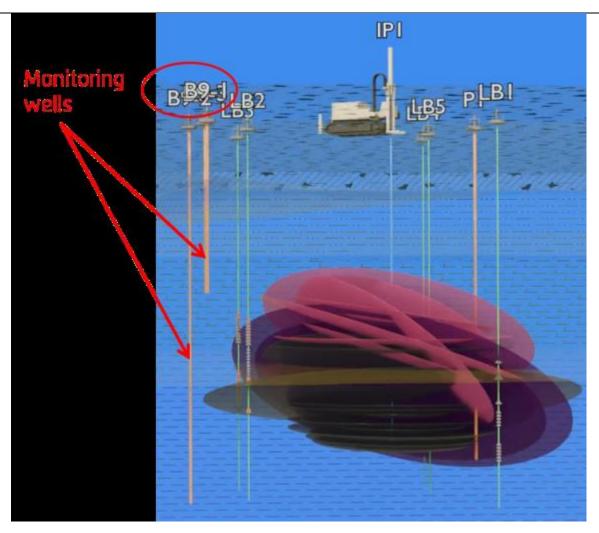
Over the entire depth of injection approximately 70% of the fractures show dip angles of less than 20° and only 30% of the fractures dip < 45°.

The average thickness of the fractures was estimated to about 12 mm. The aspect ratio is  $\frac{3}{4}$  indicating that in line with the acting geotechnical stresses, ellipses instead of circles were formed. Compared to the radius of influence projected (5 m radius meaning 10 m in diameter) the calculated radii of influence for the  $Na_2S_2O_8$  at depth is below the radii of influence of the permanganate application due to its soluble state and larger leak-off into the surrounding soils during injection. The permanganate only injection and the mixture of permanganate and persulfate reached about a 4.5 m extent in one direction. The calculated radius of influence was proven by the results of the sampling of the liner bores (see picture below), which were spatially placed at the tip of the planned injection coverage.









Furthermore, we were able to observe reagents in two groundwater monitoring wells screened in either the clayey silt layer and the weathered marlstone, which are located 7 m away from the injection bore aligned with the smaller axis of the ellipsoids (see picture left).





### 4.5 Control parameters

We recommend analysing for:

Soils (also from leachates as needed):

- CVOCs; here:
- Tetrachloroethane, Trichloroethane, cis-1,2-Dichloroethene, trans-1,2-Dichloroethene, Vinyl chloride, 1,1 Dichloroethane, 1,2 Dichloroethane, 1,1,1 Trichloroethane,
- dissolved iron, dissolved manganate, aluminium,
- Sodium, potassium, sulfate
- pH
- TC, TIC, TOC
- SOD.

#### Groundwater:

- CVOCs; here:
- Tetrachloroethene, Trichloroethene, cis-1,2-Dichloroethene, trans-1,2-Dichloroethene, Vinyl chloride, 1,1 Dichloroethane, 1,2 Dichloroethane, 1,1,1 Trichloroethane,
- CVOC Isotopes (C12/C13)
- anions, cations dissolved iron, dissolved manganate
- heavy metals
- chloride, and carbonate hardness.

## 5. Full-scale application

### 5.1 Main Reagent

Currently only a pilot scale application was performed. However, the results showed that there is no need for a full scale application for the achieved radius of influence in the pilot test area.





# 6. Post treatment and/or Long Term Monitoring

## 6.1 Post treatment and/or Long Term Monitoring

Two months after the injection of the reagents verification liner borings at five locations within the radius of influence were drilled. The liner borings (LB) showed the locations of most of the fractures via visual prove of pink (permanganate) and white (persulfate) discolouration.





Comparisons of CVOC concentrations in soil samples before the reagents were injected with CVOC concentrations in soil samples after the injection showed concentration reductions and concentration increases in various depths. We attribute these





discrepancies to the heterogeneity of the CVOC concentrations and the distance to the borehole where the baseline samples were taken from.

Interestingly, the analytical results of Na, SO<sub>4</sub>, and K show a clear correlation of the increased concentrations at the depth ranges where persulfate and permanganate were injected.

For monitoring and evaluation purposes of the remediation success groundwater samples were also taken from three groundwater monitoring wells around the injection bore immediately before the injection starts and within 18 months after injection on a bimonthly basis. The following parameters were analysed: Tetrachloroethene, Trichloroethene, cis-1,2-Dichloroethene, trans-1,2-Dichloroethene, Vinyl chloride, 1,1 Dichloroethane, 1,2 Dichloroethane, 1,1,1 Trichloroethane, Isotopes (C12/C13), dissolved iron, dissolved manganese, chloride, heavy metals, anions, cations and carbonate hardness.

Analytical results more than 18 months after injection showed an average 92% decrease of CVOC concentrations in the groundwater at all three monitoring wells with individual reductions between 80 % and 98% compared to the concentrations before the injection of the reagents.

The ratio between groundwater flux and mass reduction showed that the groundwater mass flux reduction is at least twice as high compared to the mass reduction in the soil after the injection of 3.4 t of oxidising reagents nearly two years ago.

### 7. Additional information

### 7.1 Lesson learnt

The presumed heterogeneity of the soils beneath the site is limited to about 20% variance. The presumed heterogeneity was caused by the presence of a natural fissure system, which could be proven by contaminant transport modelling. As a consequence, the previously existing conceptual site model was significantly enhanced, paving the way for a successful pilot test.

SOD analyses revealed that a high natural soil oxygen demand prevails at the site, which would in most cases have meant that ISCO would not be applicable as a remedy for the site. However, intensively evaluated data analyses also for kinetic parameters have led to a 2D numerical diffusion model (see Section 3.1) which showed that ISCO is a feasible remedy that can achieve remediation targets. A 10 cm fracture distance can achieve complete oxidation of the contaminants between the fractures. Consequently, verification bores should be placed not earlier than 4 months after the injection was completed.





Stringent injection data analyses were able to demonstrate not only the 3D-position of the fractures in the subsurface, but also the filling of the natural fissure system with the oxidative agents.

Mass-flux-reduction/mass-removal behaviour is a key indicator for sites with high groundwater concentrations and the existence of the natural fissure system as the mass-transfer process is rate limited. We were able to demonstrate that at the site there is no 1:1 ratio between mass-flux-reduction and mass-removal; instead we found a 2:1 ratio. This means that a significant mass flux reduction can be achieved by partial removal of contaminant mass from presumed DNAPL sources.

The ISCO pilot test using hydraulic fracturing as an emplacement method showed that a 50% reduction in contaminant mass achieved a 92% groundwater mass flux reduction.

A further outcome of the pilot test was that at the site persulfate activation is barely controllable for both the combination of permanganate and persulfate, and persulfate only. Uncontrolled persulfate activation in low permeability site coupled with the use of viscosifiers can lead to rapid gas development. The escape route of the produced gas can be limited by the low permeability of the soils.

Specific activation guides for persulfate are absent which would enable a safer handling of persulfate in high concentrations in low permeability environments coupled with a variety of metal oxides in the subsurface.

For a successful emplacement of oxidisers by means of hydraulic fracturing the diffusion coefficient plays a crucial role. The diffusion coefficient permanganate appears to be three orders of magnitude larger than the one for persulfate. Therefore, for the full scale application it was recommended to inject permanganate only.

### 7.2 Additional information

The utmost importance for the successful completion of this pilot test was the
fact that the client was convinced that a sound investigation phase and a proper
and detailed evaluation period is a key factor. Without the applied scientific
approach, both from the client and its consultant, this project would have been
buried two months after injection, when the results of the liner bores first came
to light and the CVOC reductions were well below the expectations.

#### Other factors were:

 There is a large difference between stoichiometry and kinetics especially for sites where very high concentrations (30-80%) of oxidisers are emplaced. This process should not be overlooked. Using kinetic information can lead to a remediation of site with very high SOD.





Ambiguity exists for diffusion coefficients of persulfate especially when the
activation energy is taken into account. Excess activation and oxidizable matter
can lead to rapid gas development, which in low permeability environments must
be controlled.

## 7.4 Additional remarks

Apply science and do not rely on gut-based comments from practitioners. Perform indepth analyses for every process – even the ones you haven't specifically targeted for or were not on the radar screen.

Most tools are already available, for specific questions one might have to go the extra mile. It pays off in the end.

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# 2. Site background

## 2.1 History of the site: Challenges and Solution



Drilling work for the remediation wells in front of building 25/1

The project site, which has been in industrial use for some 90 years, exhibited massive contamination of the groundwater in the Keuper gypsum. The CVOC concentrations, whose origin could not identified despite extensive investigation, peaked at 50 mg/l. Remediation was required in order to avoid further spreading of the contamination and minimize the hazard to the lower groundwater horizons. A number of different site-specific factors – e.g. the complex hydrogeological conditions and the continued use of the contaminated area as a customer centre – had to be taken into account. Within the framework of a feasibility study on groundwater remediation various methods

were examined in detail. They had to satisfy the following site-specific factors:

Deep-lying fissured aquifer





- Extensive spread of the contamination plume through built-over area
- Consideration for the use of the affected works area as a customer centre with some
- 600 customers every working day
- Risk of so far undetected old explosive devices in the subsurface resulting from several bombardments of the works site and the former airfield during World War II.
- Multiple branched network of supply lines and sewer conduits

The feasibility study examined a number of innovative remediation. As a result of the study, in situ chemical oxidation was recommended as the method that can be most effectively implemented in compliance with the given site-specific factors. Apart from the site-specific reasons, it was decisive for the selection of this method that the high contaminant concentrations in the groundwater could potentially be effectively reduced within a relatively short period of time, thus decreasing the existing contaminant and hazard potential.

## 2.2 Geological and hydrogeological setting

#### **Geological subsurface conditions**

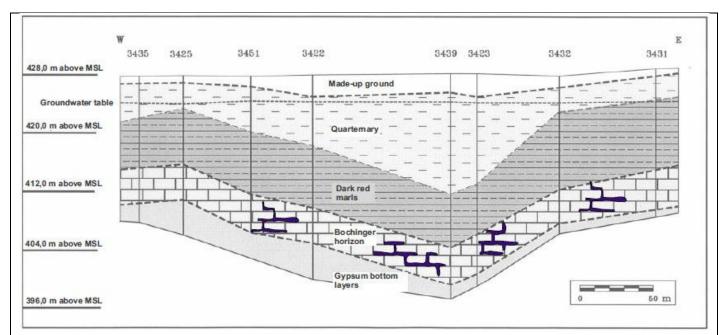
The area of investigation lies in the northeast to southwest valley plain of two watercourses.

These two brooks were rerouted in the 1970s and partly buried in underground conduits. The natural subsurface is composed of Quaternary valley deposits, interlocking in some areas with mud. The Quaternary deposits consist of a very variably structured sequence of clay, silt, sand, and fine gravel, interspersed with 0.5-3.0 m thick layers of peat, as well as of mixed-grain mud fractions. The thickness of the Quaternary deposits ranges from approx. 3.5 m to 15.0 m.

The figure below is a schematic cross-section in west-east direction through the investigation area, also showing the depression. Further down lies the stratigraphic sequence of the Keuper gypsum, encompassing the units dark red marl, Bochinger horizon, and gypsum bottom layers. The dark red marls are mostly reddish-brown, clayey silt soils with individual leached gypsum residues and friable, layered silty claystones.







Geological cross-section in west-east direction, schematic

The layers of the underlying Bochinger horizon are composed primarily of claystones and silty claystones with leached gypsum residues and dolomitic beds. In the boreholes the thickness of the Bochinger horizon is between 4.6 m and 5.8 m. Further down the Bochinger horizon is succeeded by extensively leached gypsum bottom layers consisting of silty claystones incorporating numerous leached gypsum residues as well as residual silts and marly beds.

Towards the east and southeast there are also thicker gypsum layers. Gypsum leaching can produce cavities which are reproduced by the overlying layers. In the area of the contamination centre a doline-type structure with its lowest point near monitoring points GWM 3423 and GWM 3439 was encountered.

#### **Hydrogeological conditions**

The investigation area shows two groundwater storey formations across the subsurface range explored by drilling. The upper groundwater horizon lies in the Quaternary valley deposits of the two brooks. Because of the interstratified subsurface structure with cohesive, peaty and sandy gravelly soils, the permeability conditions vary greatly, as ascertained by short pumping tests.

The Keuper gypsum layers generally show a layered and fissured aquifer system where the groundwater circulates in individual zones of increased permeability. In the investigation area, the groundwater within the Keuper gypsum sequence is carried mainly in the Bochinger horizon which has been accessed through the groundwater monitoring points installed.

The Bochinger horizon is characterized by a relatively high permeability (kf value 10<sup>-4</sup> m/s)





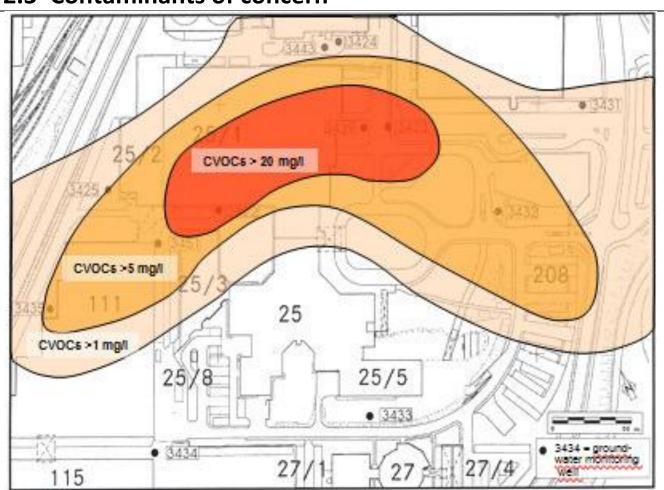
and a high yield. The groundwater circulating there is hydraulically confined. The top edge of the Bochinger horizon was encountered at depths between 12.6 m and 23.6 m below ground level; the piezometric groundwater surface lies between approx. 3.5 m and 4.5 m below ground level.

In general, the direction of groundwater flow in the eastern section of the works site runs from east to west and then turns southwest to the south of the investigation area. However, locally one must expect differing flow directions and a highly variable flow pattern.





### 2.3 Contaminants of concern



Contaminant distribution based on groundwater modelling before the start of remediation work in 2005

#### **Contaminant distribution in the groundwater**

On the basis of the drillings and investigations performed, it was possible to largely delimit the lateral contaminant distribution, which extends in the contamination centre over an area of approx. 5,000 m². The groundwater showed a clear CVOC maximum with concentrations from 30 to 50 mg/l in the region of monitoring points 3423 and 3439 on the eastern side of building 25. The contaminant spectrum was dominated by PCE which makes up approx. 80-90% of the CVOC total. Ranking secondary were TCE and cDCE as well as 1,1-dichloroethylene (1,1-DCE) and VC.

The drilling results from the actual investigation area, the works premises and the surroundings supplied the data for developing a groundwater model of the Keuper





gypsum aquifer at the Sindelfingen site. This model served to simulate the contaminant distribution in the investigation area using analytical findings from fixed-schedule sampling. The modelling result showed a contamination plume extending from east to west and turning southwest underneath building complex. In the downgradient flow further southwest, the contaminant concentrations were found to be reduced to 2-5 mg/l CVOCs.

With the isotope analyses it was established that the TCE and cDCE components found in the investigation area were direct by-products of the reductive dechlorination of tetrachlorethylene and not separately introduced contaminant components.





### 2.4 Regulatory framework

Remediation permit and other legal aspects

With innovative remediation schemes in particular – here the full-scale application of the in situ chemical oxidation (ISCO) method for the first time in Germany – it is advisable to negotiate a public law contract, making it possible to regulate complex matters within the framework of a cooperation agreement. The preamble expressed the will of the contractual parties to undertake the required remediation, and thus defined a starting point for potential contract interpretations or changes at a later date. The contract also laid down the procedure, the cleanup implementation, the monitoring measures, and the contract adaptation or termination in specific cases. In addition, the contract covered steps for a possible change concerning the method, special control mechanisms, and the establishment of a project group. Because of the novelty associated with the chosen remediation method, it was contractually agreed to publish the procedure and provide a special documentation. The water resources permit was granted taking into account all aspects for the withdrawal of groundwater and the introduction of the oxidant. It was further contractually stipulated that the fundamental effectiveness of the method should be checked on site by corresponding laboratory and field tests, and that the applicable criteria for the dosing of the oxidant should be determined with a view to the soil properties in the aguifer. The mode of action of the ISCO method was to be checked by means of a remediation test, and the required peripheral conditions with respect to occupational health & safety, well location density and oxidant injection modalities were to be optimized. Due to the positive results obtained, a permit was granted for the remediation of the Keuper gypsum aquifer.

Year	Measure
May 2003	Laboratory test
Sept. 2003	Injection test
Oct. 2003	Pilot-scale test
May 2004	Remediation test
Sept. 2005 – May 2008	Source remediation
Since June 2008	Monitoring programme

Before being able to commence with the groundwater cleanup activities at the site, using the in-situ chemical oxidation method, it was necessary to carry out a step-by-step check of the suitability of the ISCO method under site-specific conditions. Ever since the conclusion of the source remediation in May 2008, a monitoring programme has been in place.

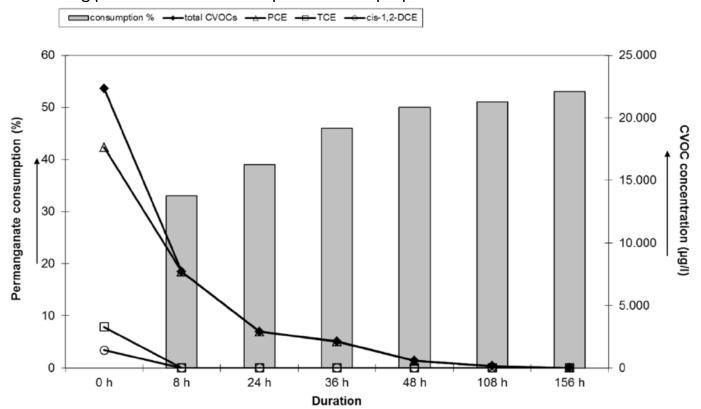




## 3. Laboratory-scale application in field

## 3.1 Laboratory scale application

Chemical laboratory analyses of groundwater samples from the highly contaminated monitoring point GWM 3423 were performed in preparation for a field test.



Laboratory test, oxidation of CVOC contaminated groundwater from monitoring point GWM 3423 using permanganate

The figure represents an example of a concentration curve of a measurement series with a permanganate concentration of 80 mg/l.

After only 8 hours the original CVOC content was degraded by almost 70%. The last measurement after 156 hours merely showed a concentration of 6.5  $\mu$ g/l CVOC. Accordingly, the chlorinated ethylenes (PCE, TCE, cDCE, VC) were almost completely oxidized; this was however not the case with the chlorinated ethanes (1,1-dichloroethane and 1,1,1-trichloroethane) which were only present in low concentrations. In agreement with the literature (ITRC, 2006), the results show that low-chlorinated ethylenes are oxidized faster than higher chlorinated ones. The fast degradation was also promoted by a very low organic content of 2.0 to 3.5 mg/l TOC in the groundwater sample.





## 4. Pilot-scale application in field

## 4.1 Main treatment strategy

#### Pilot scale test

After the positive laboratory findings, the next step consisted of a pilot-scale test in the investigation area as a preparation for the ISCO cleanup of the groundwater using permanganate. It involved an injection test at monitoring point GWM 3424 and the actual pilot scale test with permanganate injections at monitoring points GWM 3423 and GWM 3424 as well as a four-week pumping measure at groundwater monitoring points 3422 and 3425 simultaneously.

The injection test showed that the groundwater level in the well rose in the case of permanganate injection compared to injection with ordinary water. The cause may possibly be that oxidation reactions in the filtration area already occurred during injection, so that reaction products hampered the outflow of the infiltrate. Moreover, the greater viscosity of the injection solution, compared with water, may also have been a reason for the rise in the water level.

The objective of the pilot-scale test was initially to test the method's basic mode of action at the site. Furthermore, it was to be checked to what extent the permanganate injected at monitoring points 3423 and 3424 could be distributed underneath building 25 by means of pumping measures in the downgradient monitoring points 3422 and 3425, in order to remediate the building area not directly accessible through drilling.

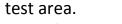
During the five-week pilot-scale test a total of approx. 1,390 kg  $MnO_4$  – in the form of sodium permanganate – was introduced into the contamination centre via the two monitoring points 3424 and 3423. To prevent uncontrolled drifting of the permanganate, approx. 7.5 m³/h of groundwater was abstracted in the downgradient flow at the two wells GWM 3425 and GWM 3422. The water was then cleaned in a mobile stripping plant down to CVOCs < 10  $\mu$ g/l and discharged into the sewer system.

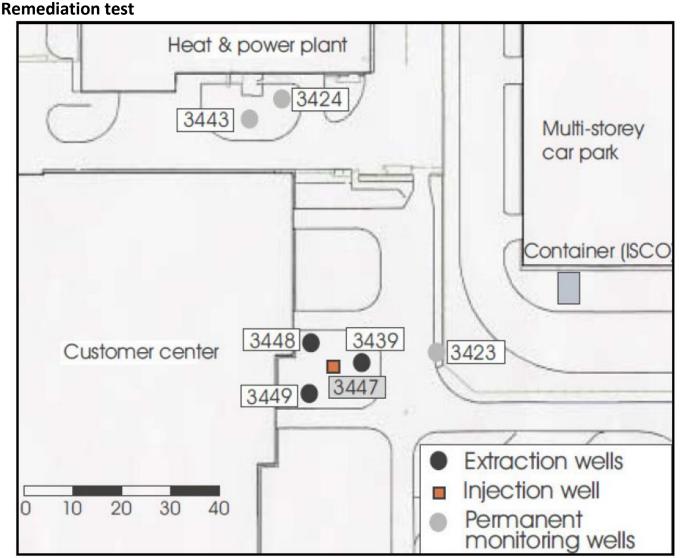
Accompanying the tests was an extensive monitoring programme carried out at a total of eight wells. The parameters CVOC, Mn<sup>2+</sup>, Na<sup>+</sup>, Cl<sup>-</sup>, pH value, temperature, conductivity, and redox potential were measured at regular intervals at monitoring point 3439, which lies closest to the injection monitoring points, and at the two extraction wells. The other monitoring points were sampled before the start and after the end of the test phase, in order to assess the impact on the further surroundings.

It was found that the CVOC concentrations fluctuated. After injection of permanganate the contaminant levels were at first clearly reduced, only to increase later. The increase is most likely due to the subsequent inflow of contaminants into this area. In general, the results were inconclusive and could not be fully interpreted because of the rather large









Layout plan of the test site with location of groundwater monitoring points

While the pilot-scale test in autumn 2003 covered a large area, the remediation test was carried out in a relatively small green area directly adjacent to building 25/1. The objective of the remediation test was to gain further insights with a view to the technical application of the ISCO method at this particular site. Firstly, significant oxidative destruction of CVOCs was to be proved; secondly, it was to be investigated how an optimum distribution and dosing of the oxidant in the subsurface could be implemented. Furthermore, it was to be tested to what extent the technical measures (drilling, pipe laying, etc.) would be acceptable in the area of the customer centre without undue disturbance. The test concept envisaged the provision of a closed circulation system where the

groundwater would be pumped off, enriched with permanganate and then reinfiltrated. The remediation test was also studied within the framework of a diploma thesis.





The test site comprised a small area of some 400 m² with the central groundwater monitoring point GWM 3447 used as an injection well and the three groundwater monitoring points GWM 3439, 3448 and 3449, which form a star pattern, serving as extraction wells. This arrangement and mode of operation were intended to ensure that the permanganate is distributed across the area without uncontrolled migration. The average circulation rate of the groundwater amounted to approx. 2.8 m³/h. In total, about 1.4 t of permanganate with a concentration of approx. 500-1,000 mg/l was injected and about 3,800 m³ of groundwater was recirculated in the aquifer of the test site. The permanganate was injected over a period of six weeks. After termination of the permanganate injection the test facility was run for another three weeks. Before, during and after the test, groundwater samples were analysed for CVOCs, and on-site measurements were performed with regard to the pH value, temperature, conductivity, and redox potential.

Manganese, sodium and chloride were analysed as additional parameters. Furthermore, samples were taken regularly and examined photometrically in the laboratory for their permanganate content.

The table presents a comparison of the CVOC total concentrations at the start of the test compared to two and fourteen weeks after the end of the test.

	CVOC total concentrations [mg/l] percentage (%)			
Time period	GWM	GWM	GWM	GWM
t[d]	3439	3448	3449	3447
	extraction	extraction	extraction	injection
t = 0 d	37.3	34.2	34.3	35.1
	(100%)	(100%)	(100%)	(100%)
t = 77 d	22.5	22.8	17.6	15.5
	(60.3%)	(66.6%)	(51.3%)	(44.2%)
t = 162 d	20.7	16.1	19.9	2.71(%)
	(55.5%)	(47.0%)	(58.0%)	

Contaminant concentrations at the groundwater monitoring points of the remediation test during the period t = 0 d to t = 162 d

The contaminant reductions achieved within only five months are quite remarkable. However, without further remedial measures the contaminant concentrations would presumably have risen again due to the inflow of groundwater with higher pollution levels.

The parameters pH value and redox potential (Eh), which were measured on site, showed a clear reaction directly after the injection of permanganate and proved to be suitable indicators for the CVOC oxidation processes occurring in the subsurface. By contrast, the





measurements of conductivity and temperature did not find any significant changes. In tandem, a sample from GWM 3439 was examined for possible by-products by means of LC-MS screening (U.S. DOE., 2000). Initially, the findings showed glyoxylic acid with 0.12 mg/l, hydroxyacetic acid with 0.04 mg/l and oxalic acid with 0.46 mg/l. This screening was repeated 4 months later using another sample from the same monitoring point, in order to be able to assess the potential long-term accumulation of these acids. Here, the concentrations were below the respective determination limit of 0.05 mg/l for glyoxylic acid as well as 0.1 mg/l for hydroxyacetic acid and oxalic acid. With the exception of the existing CVOCs, chlorinated organic compounds, such as trichloroacetic acid, were not detected.

### 4.3 Injection type

- Existing groundwater monitoring wells (GWM) were used for injection of permanganate and for establishing a groundwater circulation (cf. chapter 4.1 Main treatment strategy, Remediation test).
- The distance between injection and extraction wells was about 10 m.
- Permanganate solution was injected into the Bochinger horizon in a depth of about 15-25 m bgl.
- The permanganate was injected continuously over a period of six weeks. After termination of the permanganate injection the test facility was run for another three weeks.
- Sodium permanganate was used as ISCO agent. In total 1.4 t MnO<sub>4</sub> with a concentration of approx. 500-1,000 mg/l were injected.

## 4.4 Radius of influence

The distance between the injection and extraction wells was about 10 m. The establishing of a circulation system was verified by tracer tests using the rising conductivity caused by injection of sodium permanganate.

The radius of influence is regarded higher than 10 m since there are another 10 m radius of influence around the extraction wells.





### 4.5 Control parameters

Before, during and after the test, groundwater samples were analysed monthly for CVOCs, and on-site measurements were performed weekly with regard to the pH value, temperature, conductivity, and redox potential. Manganese, sodium and chloride were analysed as additional parameters twice. Furthermore, samples were taken regularly and examined photometrically in the laboratory for their permanganate content. In tandem, a sample from GWM 3439 was examined for possible by-products by means of LC-MS screening (U.S. DOE., 2000). Initially, the findings showed glyoxylic acid with 0.12 mg/l, hydroxyacetic acid with 0.04 mg/l and oxalic acid with 0.46 mg/l. This screening was repeated 4 months later using another sample from the same monitoring point, in order to be able to assess the potential long-term accumulation of these acids. Here, the concentrations were below the respective determination limit of 0.05 mg/l for glyoxylic acid as well as 0.1 mg/l for hydroxyacetic acid and oxalic acid. With the exception of the existing CVOCs, chlorinated organic compounds, such as trichloroacetic acid, were not detected.





# 5. Full-scale application

### 5.1 Main Reagent

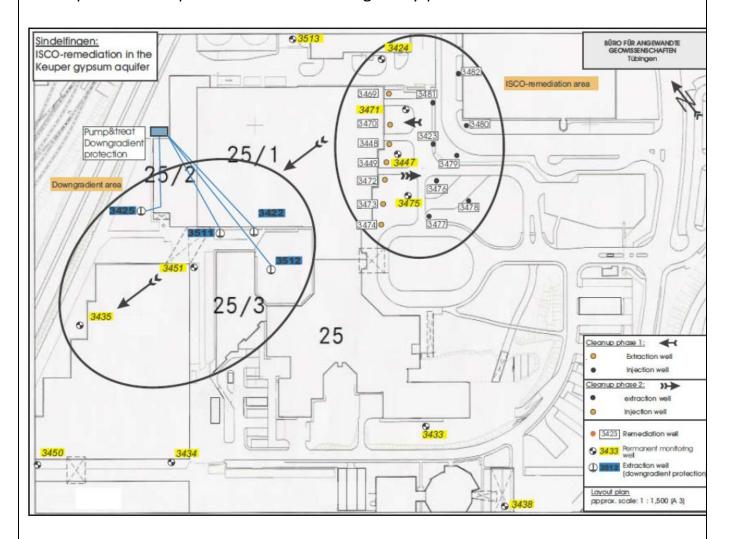
#### Remediation

Due to the positive test results coupled with the cost and time advantages compared with a pump and treat scheme, the cleanup of the contamination centre was implemented using the ISCO method with permanganate. The relevant public authorities deliberately refrained from laying down a specific cleanup target. However, the objective was to employ the ISCO method in order to reduce the contaminant contents by about 80-90% within 2-3 years. The entire contamination zone extends over an area of approx. 20,000 m<sup>2</sup>, with some 5,000 m<sup>2</sup> thereof involving the contaminant source. Due to this large size and the limiting site conditions it was not possible to implement the method, as successfully tested, on a 1:1 basis across the entire contamination zone. The ISCO remediation was therefore restricted to the contamination centre located east of the customer centre, so as to achieve an effective contaminant reduction within a short time period and stop the further spreading of contaminants into the downgradient flow, as well as counter the hazard of migration into deeper groundwater horizons. In order to prevent uncontrolled movement of contaminants and permanganate, a pump-and-treat plant was installed as a hydraulic protection measure. This also served to distribute the oxidant over a large area underneath the building. The ISCO remediation in the Keuper gypsum aguifer proceeded in four successive phases. Initially, the cleanup took place in the area of the contamination source directly in front of the customer centre. Over a period of 15 months sodium permanganate was injected, as a dilution, into the upgradient groundwater wells; it was then transported westwards with the natural groundwater gradient. The transport and distribution of the oxidant were supported by groundwater extraction in the downgradient wells located in front of building 25/1. The extracted groundwater was cleaned and used as process water for diluting the 40% NaMnO<sub>4</sub> solution. Phase 2 The subsequent work, carried out over a period of 15 months, focussed on the contamination area underneath building 25/1. Here, a sodium permanganate dilution was injected through the groundwater monitoring points directly in front of building 25/1, and groundwater was extracted from the downgradient flow southwest behind the building. Phase 3 After termination of the phases 1 and 2, permanganate was once more injected through the wells directly in front of building 25/1, in order to create an oxidant pool for the destruction of the remaining CVOC content. Phase 4 In May 2008 the active measures of the ISCO cleanup project were concluded. Since that time a long-term monitoring programme has been running at the





site. If follow-up monitoring should indicate a rebound at individual wells, this will dealt with by systematic permanganate injections. The groundwater in the downgradient area now contains only comparatively low CVOC concentrations of < 1 mg/l and is being cleaned via a conventional pump-and-treat system. The wells for downgradient protection are sampled at monthly intervals, all the other groundwater monitoring points at the site are sampled every six months and analysed for CVOCs, Mn2+ and Cl-. It is also planned to repeat the LC-MS screening for by-products.



ISCO remediation of the Keuper gypsum aquifer. Layout plan with ISCO cleanup area and downgradient area

#### **Results**

In the period from September 2005 to May 2008, a total of 30 tonnes of oxidant was injected, corresponding to the destruction of about 7,500-10,000 kg of CVOCs. This calculation is due to results of the field tests which indicated a specific oxidant demand of approx. 3-4 kg permanganate per kg CVOCs. The figure below depicts the ISCO plant





technology including a dosing station for sodium permanganate.



ISCO plant technology







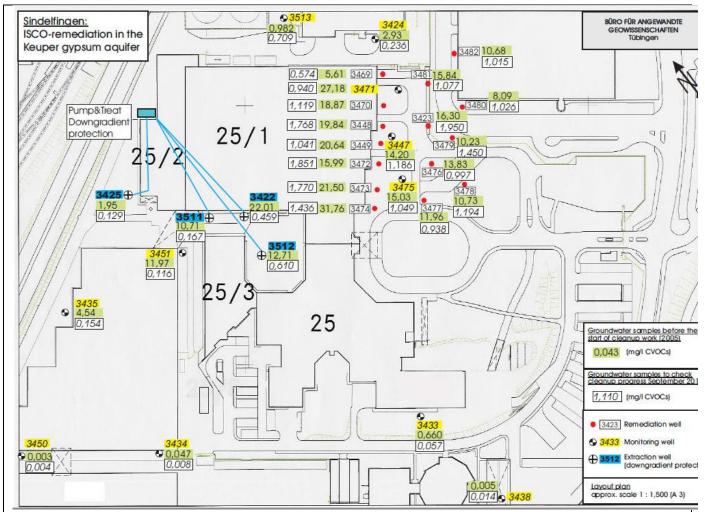
ISCO dosing station

The project achieved a reduction in the CVOC concentrations at the contamination centre down to approx. 0.1-1.9 mg/l, which equals a mean decrease of around 90-95%. As another positive result it should be noted that, with the exception of one well, there has been no significant rebound of the contaminant concentrations. Reduced permeability in the Keuper gypsum aquifer due to  $MnO_2$  precipitation could not be detected.

The ISCO measure at the contamination centre has also had a positive effect on the downgradient flow. Currently (status 12/2020) the CVOC concentrations in the downgradient wells lie below 0.5 mg/l and have thus been reduced by approx. 85-90%. Consequently, in-situ chemical oxidation also offered economic advantages compared with the conventional pump-and-treat method.







Contaminant concentrations before the start and after conclusion of active ISCO remediation

in the Keuper gypsum aquifer (status December 2019)





### 5.3 Injection type

- Existing wells were used and new injection wells in front of building 25/1 were drilled.
- In phase 2 of the remediation works the new wells were used for ISCO treatment of the contamination area underneath building 25/1.
- Permanganate was injected continuously over a period of 15 months in phase 1 and in phase 2. After termination of the phases 1 and 2, permanganate was once more injected through the wells directly in front of building 25/1, in order to create an oxidant pool for the destruction of the remaining CVOC content. A total of 30 tonnes of oxidant with a concentration of 500 –mg/L MnO<sub>4</sub> was injected in the period from September 2005 to May 2008.

#### 5.4 Radius of influence

The transport and distribution of the oxidant were supported by groundwater extraction downstream. The distance of influence was 50 – 100 m verified by monitoring.

### 5.5 Process and performance monitoring

The wells for downgradient protection are sampled at monthly intervals, all the other groundwater monitoring points at the site are sampled every six months and analysed for CVOCs, Mn2+ and Cl-. On-site measurements were performed weekly to monthly with regard to the pH value, temperature, conductivity, and redox potential.





### 6. Post treatment and/or Long Term Monitoring

#### 6.1 Post treatment and/or Long Term Monitoring

Since termination of the active remediation all wells and are sampled twice per year and analysed for CVOCs.

#### 7. Additional information

#### 7.1 Lesson learnt

# Difficulties and weaknesses, successes and strengths, keystones, shortcomings and rooms for improvement

The project achieved a reduction in the CVOC concentrations at the contamination centre down to approx. 0.1-1.9 mg/l (status 12/2020), which equals a mean decrease of around 90-95%. As another positive result it should be noted that, with the exception of one well, there has been no significant rebound of the contaminant concentrations. Reduced permeability in the Keuper gypsum aquifer due to MnO<sub>2</sub> precipitation could not be detected. The ISCO measure at the contamination centre has also had a positive effect on the downgradient flow. Currently (status 12/2020) the CVOC concentrations in the downgradient wells lie below 0.5 mg/l and have thus been reduced by approx. 85-90%. Consequently, in-situ chemical oxidation also offered economic advantages compared with the conventional pump-and-treat method.

#### Outlook

In-situ chemical oxidation is an established and very promising groundwater remediation technology suitable for a wide range of organic contaminants. Among the various in-situ methods, ISCO occupies a prominent market position in Germany and is also increasingly being applied in other European countries. The projects of Züblin Umwelttechnik GmbH implemented so far at more than 40 different contaminated sites have shown that the ISCO method enables a fast reduction of high contamination levels in groundwater. Additionally, the method is also very well suited for minimizing the existing contamination potentials underneath buildings. The direct contact of pollutants and oxidants is the essential prerequisite for a successful application of the ISCO technique. ISCO in a low permeable underground is a challenge but can be realized successfully using specific injection technology (e.g. fixed manchette tubes). However, the ISCO method cannot be applied for all types of contamination involving CVOCs or organic pollutants. Large pools of DNAPL and LNAPL phase cannot be remediates using ISCO. For economic reasons, the method is less suitable for the remediation of extensive





contamination plumes or soils with a very high content of organic substances. The successful application of the ISCO method requires detailed knowledge of the subsurface conditions and the spatial distribution of the contaminants, as well as broad practical experience. Field tests for checking the cleaning efficiency in the aquifer are strongly recommended in case of complex hydrogeological situation.

### 7.3 Training need

Workshops, training on-the job, webinars, e-learning could be an effective training tool

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## 2. Site background

## 2.1 History of the site: Challenges and Solution

The area subjected to the remediation is into the Porto Marghera Site of National Interest - Venice – Italy









### 2.2 Geological and hydrogeological setting

- The area subjected by the remediation has a flat surface, at an approximate altitude of 3 meters above the average sea level.
- The surface morphology of the area is the result of the lagoon landscaping carried out in the past to enlarge the industrial area, with the creation of the peninsula now called "Nuovo Petrolchimico". The sales pitch was aimed at raising and levelling the area to enlarge the industrial area, and is made up of material of a heterogeneous nature, often residues of industrial processing. Within this backfilled material, saturated areas impregnated with water are observed, commonly defined as "groundwater in the backfill" or "backfill impregnation waters". The absolute altitude of the groundwater level is quite variable and cannot be correlated with each other, making it impossible to identify a real direction of groundwater flow.
- From the ground level to a depth of about 3 meters: heterogeneous fill layer, consisting of coarse material (gravel, tout-venant) in fine matrix (sands, silty sands, silts), used in the past for the raising of the ground level and for the localized filling of the most depressed areas, in order to create the new industrial zone.
- Up to a depth of about 5 meters from the ground level: fill made up of red bauxitic mud or blackish mud. Materials of pasty consistency, compact, of variable thickness within the site, used in the past for the artificial filling of the lagoon sandstone area and the raising of the countryside floor, in order to create the new industrial area.
- Depth of the aquifer of the backfill from the ground level 1.2 m





#### 2.3 Contaminants of concern

Given the nature of the fill and the relative "suspended" aquifer characteristic of the Porto Marghera area, the qualitative state of the environmental matrices is not homogeneous within the site. The characteristics of the very limited area, identified by the LEV06 survey, on which the In Situ Chemical Oxidation (ISCO) technology has been applied, are reported.

The area actually affected by the reclamation intervention is equal to 450 m<sup>2</sup>.

The integrative characterization, carried out in January 2016, also showed exceedances in the S2 and S5 surveys which confirmed the presence of heavy hydrocarbons C> 12 and some polycyclic aromatic hydrocarbons (PAHs). The overall portion of contaminated land is between 3 m and 5 m from the ground level

The contamination was detected between 4 m and 4.6 m deep and it is due to the presence of heavy hydrocarbons C>12, detected in a concentration equal to 837 mg/kg and some PAHs (benzo(a)anthracene 16, 1 mg/kg, benzo(b)fluoranthene 22.5 mg/kg, benzo(a)pyrene 15.3 mg/kg and indeno(1,2,3-cd)pyrene 10.3 mg/kg).

There is no NAPL (Non-Aqueous Phase Liquids)

Qualitative status of the groundwater in the LEV06 piezometer, located near the intervention lot with ISCO:

Al: 146 μg/l
As: 44.6 μg/l
Mn: 42.4 μg/l

• Benzene: 1.38 μg/l

• Vinyl chloride monomer (VCM): 2.47 μg/l

• 1,1-dichloroethylene: 0.00914 μg/l

• Sulphates: 6.63 μg/l

#### 2.4 Regulatory framework

- Italian Legislative Decree 152/2006
- Italian Ministerial Decree 31/2015
- "Accordo di Programma per la Bonifica e la Riqualificazione Ambientale del Sito di Interesse Nazionale di Venezia – Porto Marghera e Aree Limitrofe"





### 3. Laboratory-scale application in field

### 3.1 Laboratory scale application

Due to the deep location of the contamination and the heterogeneity of the backfill, no laboratory tests were performed as they were considered not significant.

### 4. Full-scale application

### 4.1 Main Reagent

ISCO technology was chosen for the depth to which the contaminated layer was located and for the ability of oxidizing substances to degrade hydrocarbons into simpler compounds that are generally not critical for the environment.

For the case under examination, two oxidizing compounds produced by Regenesis were chosen.

For the first cycle, it is has been chosen the compound Regenox™, which is a compound designed to treat areas characterized by elevated concentrations of organic contaminants. The main characteristics of the product can be summarized as follows:

- It allows rapid and effective oxidation of a wide range of compounds, such as hydrocarbons (aromatic, aliphatic, polyaromatic, chlorinated);
- It consists of two parts:
  - o Part A: it is the oxidizing complex consisting of a mixture of sodium percarbonate (2Na₂CO₃•3H₂O₂), sodium carbonate (Na₂CO₃), sodium silicate and silica gel. The oxidizing complex appears as a fine white powder.
  - Part B: it is the activator complex consisting of a mixture of sodium silicate, silica gel and ferrous sulphate. It looks like a liquid gel. It has a rather limited longevity and acts only on the desorbed phase.

For the second cycle, it was planned to use an oxidizing product with a greater capacity to permeate the subsoil such as Sodium Persulfate ( $Na_2S_2O_8$ ). In fact, together with a high oxidation potential, Sodium Persulfate has characteristics of high solubility and medium persistence in the subsoil. With a solubility limit equal to 40% w/w it is therefore possible to apply, for the same volume of injected solution, a greater quantity of oxidant. The reagent, suitably activated (thermally or chemically) produces the release of free radicals with high oxidation potential ( $SO_4$  •, OH •,  $O_2$  •) allowing the degradation of a broad spectrum of contaminants including organic compounds such as PAHs.





In consideration of the characteristics of the subsoil and the contaminants to be treated it is provided the use of a solution of 15% Sodium Persulfate, together with an activator based on caustic soda in 25% solution.

The criticalities found in the case in question are due to the low permeability of the soil and the recalcitrant nature of the PAHs

#### **Reactive dosage**

The theoretical dosage of RegenOx provides for an oxidant / hydrocarbon weight ratio equal to

10: 1. In this case, the theoretical dosage requires the use of 140 kg of oxidant. In the case of injections in saturated soils, the yields for this type of intervention vary from 40% to 95% depending on the site specifics and contaminant characteristics. To ensure an adequate safety margin, it was chosen for a double dosage of oxidant, equal to a total of 300 kg. A similar quantity will be provided for the activator Part B. According to the supplier's instructions, in order to obtain an 8% aqueous solution of oxidant, it was necessary add a quantity of water equal to about 10 liters per kilogram of oxidizer and activating agent. Therefore, for the procedure it was necessary to use a volume of water equal to approximately 3 m<sup>3</sup>.

Since it was planned to apply the oxidizer through multiple injection points, the preparation of the solution was carried out by dividing the quantities on the basis of the number of injection points.

As for the Sodium Persulfate, 400 liters of reagent were injected into the soil at each input point., resulting from the mixing of sodium persulfate at 15% and caustic soda at 25% according to the following proportions:

- 50 kg of Sodium Persulfate (Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub>) at 15% in 350 l of water;
- 15 kg of Caustic Soda (NaOH) at 25% in 50 l of water.

#### **Application system**

#### First course of treatment

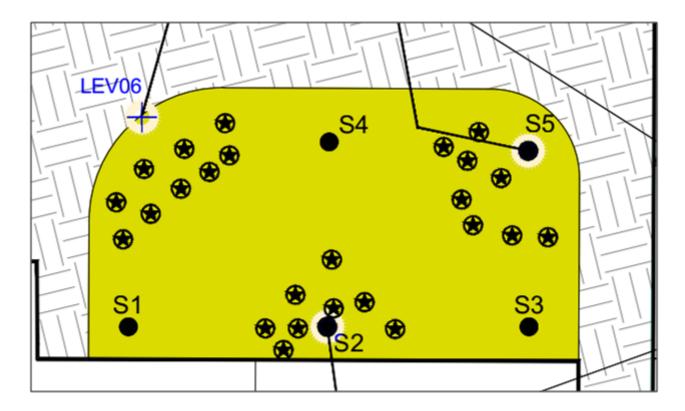
The injection of the solution containing the RegenOx™ in deep soil was carried out by direct injection, with direct push machines like Geoprobe® in order to improve the oxidant distribution and homogenization in the aquifer. Due to the limited soil thickness to be treated, direct injection has been done in bottom-up mode. The injection probe, the final element of the drill rod system, is brought to the maximum depth to be treated. This probe was equipped with a nozzle opening-closing system controlled by surface, integral with the probe or disposable This probe was equipped with a nozzle opening-closing system controlled by surface, integral with the probe or disposable. Once the desired depth is reached, the rod system was connected to the injection pump, which in turn was connected to the tank containing the oxidizing solution. At this





point the injection nozzle was opened, and then the pumping of the solution started. The pumping was continued during the rods system rise, until this reached the minimum depth of treatment. The injection system must have a volumetric counter, to allow dosing of the correct volume of solution per unit of vertical length. To guarantee a safety margin, the injection will be carried out between the depths of 3.8 m and 4.8 m from the ground level. Once the upper end of the injection interval has been reached, the same injection system will be used for the injection of a benthonite mixture during the ascent of the rod system to the surface.

The first injections cycle were performed on 30, 31 May and 1 June 2016, the second one on 11, 12 and 13 July 2016. Overall, during the two campaigns, 27 injections positioned around the polluted points of investigation LEV06, S2 and S5 were performed, where the contamination by heavy hydrocarbons C> 12 and PAH was found. Specifically, around each point 9 injections were done according to the configuration below:

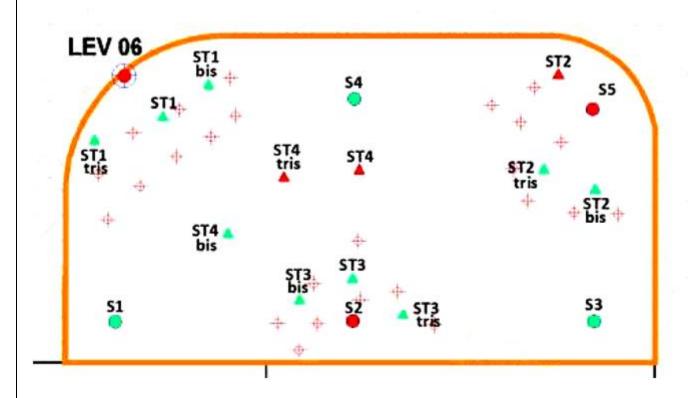


Following the injection of the oxidizing mixture, the remediation monitoring activity was carried out, which included the execution of n. 4 boreholes, with a fortnightly frequency, up to 5 m deep, for a total of 3 survey campaigns from 01/08/2016 to 29/08/2016 (T1, T2 and T3). In this way, the effect of the oxidizing mixture was monitored respect to the portion of soil subject to remediation (between 3.0 m and 5.0 m of depth from the





ground level) As shown in the following figure, the results of the performed monitoring showed a residual contamination by PAHs and heavy hydrocarbons (C> 12) compared to the contamination threshold concentrations (CSC) and the risk threshold concentrations (CSR) of reference provided by the Italian legislation, exclusively between 4.0 m and 5.0 m of depth, therefore in saturated soil, in the ST2 and ST4 boreholes, carried out at verification time T1 and in the ST4btris borehole carried out at verification time T3.



Samples to monitor remediation (ST, STbis, STtris), green means that the target concentration was reached, red means that the target concentration was not reached

In order to better define the extent of the contamination identified in the survey were ST4tris additional investigations, on 15 November 2016, were carried out:

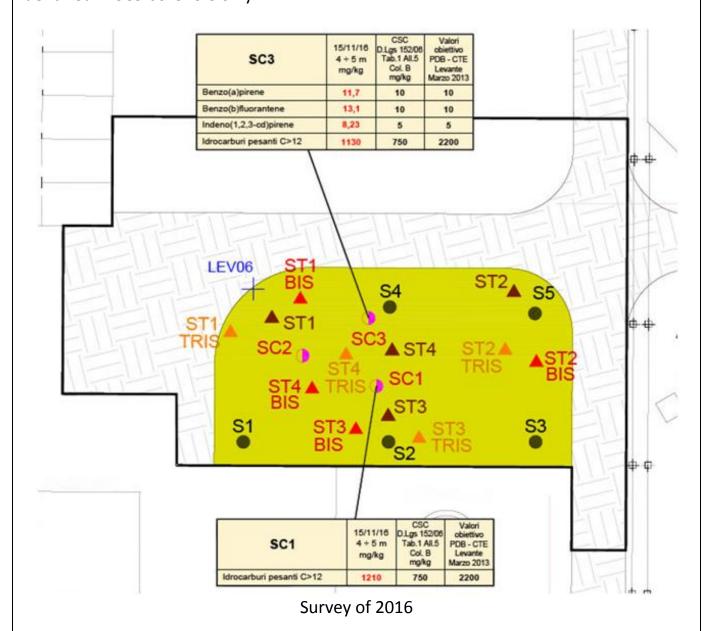
- the perforation of 3 geognostic boreholes, up to a depth of about 5 m by the ground level (SC1, SC2 and SC3), located around the ST4tris borehole, as shown in the following figure;
- the taking of 5 soil samples corresponding to each survey (one sample representative for each meter of depth in accordance with the provisions of the current Italian law "Protocollo operativo per la caratterizzazione dei siti ai sensi del D. Lgs.152/06 e dell'accordo di programma per la chimica di Porto Marghera Revisione Gennaio 2008"). Altogether they were sampled and sent to the laboratory 15 samples;





• laboratory analyzes to research polycyclic aromatic hydrocarbons (PAHs) and heavy hydrocarbons C> 12.

The analytical results relating to the supplementary investigation campaign showed that the CSCs were exceeded for heavy hydrocarbons C > 12, for the SC3 and SC1 boreholes, but with concentrations lower than the target values, equal to 2200 mg/kg. For the PAHs, modest exceedances of the CSCs (corresponding to the target values) are identified in SC3 borehole only.



#### **Second course of treatment**

Considering the specific lithological conditions and the state of contamination found in the area in question it was chosen to use sodium persulfate. ( $Na_2S_2O_8$ ), an oxidizing product with greater ability to permeate the subsoil than the RegenOx already





employed. In addition, in the light of the need to make multiple injection; campaigns or it, it was decided to carry out the application of the oxidizing mixture through a network of injection wells specially realized. The 10 injection points, sequentially named P1 - P10 were distributed according to a regular mesh of about 2.20 m within the polygons identified by the SC3, ST4 and ST4tris probes. They were manufactured through the use of a drilling machine Atlas Copco up to 5.0 m from the ground level, to intercept the layer of soil between 4.0 and 5.0 m in depth where the overcoming of the CSCs were highlighted. At the end of the drilling activity, the boreholes were equipped with a piezometer using 3" HDPE pipes, 5.0 m long. In all piezometers the slotted portion (slot = 0.5 mm) extends for 1.5 m starting from the bottom of the hole, while the remaining 3.5 m part is blind. A drainage mantle with pre-calibrated gravel ( $\emptyset = 2$  mm) from about 0.3 m above the "top" of the filtered section to the bottom of the hole was prepared in the hole/pipe interspace. To ensure a proper insulation by the penetration of surface water and to prevent the possible ascent to the surface of the oxidizing solution during the injection operations, over the drain core were paid, in sequence, a 0.5 m layer of benthonite pellets and cement mortar up to 0.3 m from ground level. At each injection point, 400 I of reagent was injected into the soil, resulting from the

At each injection point, 400 l of reagent was injected into the soil, resulting from the mixing of Sodium Persulfate at 15% and Caustic Soda at 25% according to the following proportions:

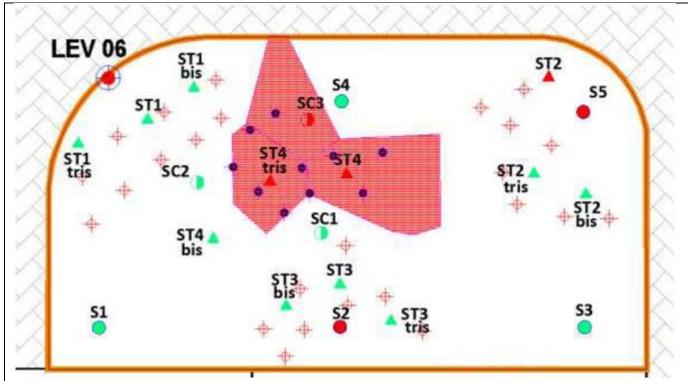
- 50 kg of Sodium Persulfate (Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub>) at 15% in 350 l of water;
- 15 kg of Caustic Soda at 25% in 50 l of water.

The two substances, solid powder the sodium persulfate and liquid the caustic soda, were previously mixed with water in separate tanks in order to minimize the probability of triggering of exothermic reactions dangerous for operators and, only subsequently, mixed together in a common tank. The mixing of the substances was carried out by means of a manual electric mixer with stainless steel stirrers and was continued until their complete or homogenization.

The injection took place by means of a dedicated automatic unit consisting of a piston motor pump and a sealing cap screwed to the wellhead. The injection pressure was constantly monitored and kept less than 2 bar to prevent the flowing back of the substance in neighbouring injection points. Respecting this modality, the only visible effect of the injections was an increase in the local piezometric level of about 0.5 m. In May 2017, 10 injection points were distributed, according to a regular mesh inside of the polygons identified surveys SC3, ST4 and ST4tris as indicated in the following figure.





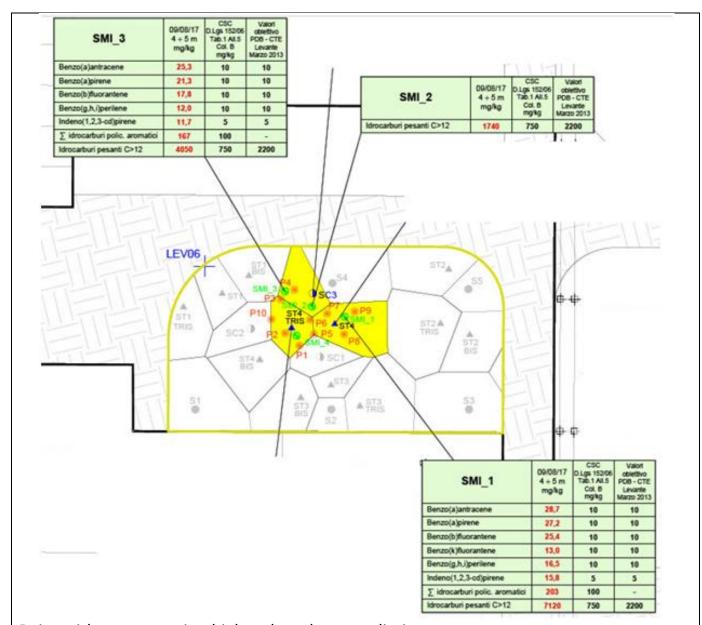


New injection points

Once the new network of injection points had been developed, the first ISCO intervention took place on the 10 injection points on 29, 30 and 31 May 2017 and the second intervention took place on 3 and 4 July 2017. Subsequently, on August 9, 2017 a land monitoring campaign was carried out that showed the presence of PAHs and heavy hydrocarbons (C > 12) in concentrations superior to remediation targets in the range of a depth of only between 4.0 m and 5.0 m in depth only, corresponding to saturated soils, and for the monitoring points SMI\_1 and SMI\_3 only. Instead, at the same depth, the SMI\_2 sample presented concentrations of heavy hydrocarbons in excess respect to the CSC reference but lower than the CSR reference. Finally all the samples from the survey SMI\_4 showed concentrations of the sought parameters lower than the reference CSC.







Point with concentration higher than the remediation target

From the outcomes of the executed campaigns it should be noted that the oxidation intervention resulted only partially effective in the treatment of contamination from heavy hydrocarbons and IPA in the saturated soils, between 4.0 m and 5.0 m in depth. This occurred even if an increase in the reagent dosage and more soluble mixtures were adopted. The verification investigations have in fact highlighted a residual contamination in the soils characterized by the presence of high concentrations of heavy hydrocarbons and PAHs in a localized portion of the subsoil around the monitoring points SMI\_1 and SMI\_3.

In light of the results achieved, it was necessary to evaluate the state of affairs of the area by means of a testing activity aimed at defining the portions of land that have





achieved the objectives and those that still present residual concentrations, higher than those foreseen in the authorized project.

Finally, the persistence of concentrations exceeding the target value was resolved with a risk analysis.

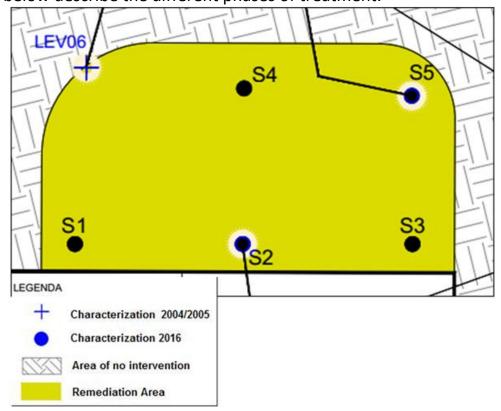
#### 5.2 Additives

In the case of oxidant complex RegenOx<sup>™</sup>, consisting of a mixture of percarbonate of sodium a an activating complex constituted by a mixture of sodium silicate, silica gel and ferrous sulfate. It appears as a liquid gel.

Instead, in the case of Sodium Persulfate (Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub>), Caustic Soda (NaOH) has been added.

### 5.3 Injection type

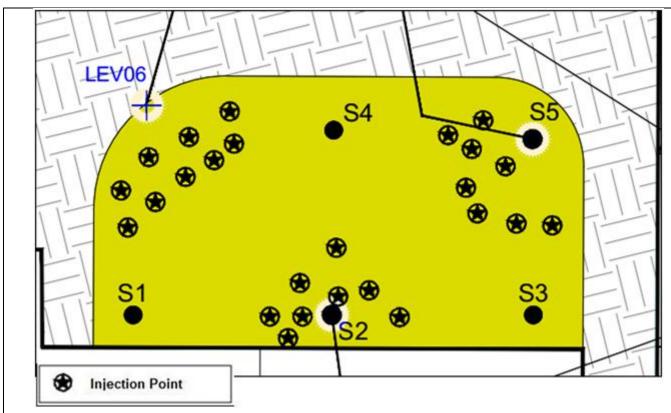
The methods of injection of the oxidant are described in section 4.1 The figures below describe the different phases of treatment.



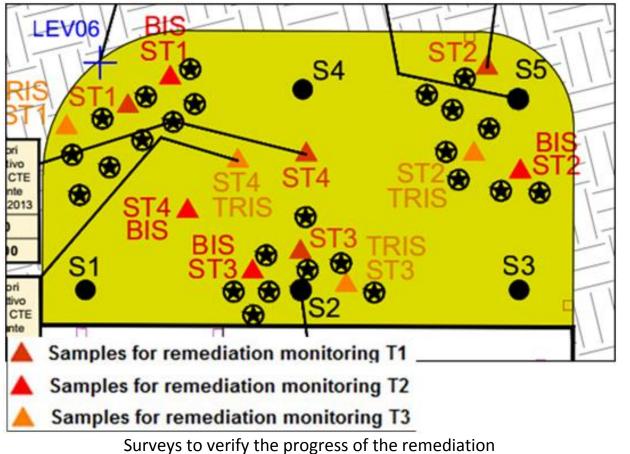
Survey contaminated by LEV06 characterization and integrative surveys to limit the intervention area





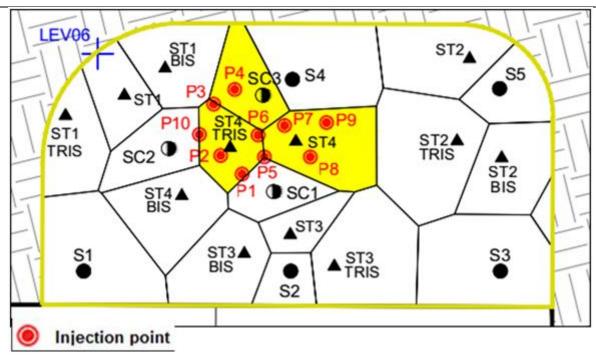


First cycle of treatment with sodium carbonate









Second cycle of treatment with sodium persulfate

- Information on the injection layer is given in paragraph 2.3
- Information about to the number of injection campaigns (how many campaigns, timing, dosages) and the dosage of the ISCO agent are reported in section 4.1
- No injection enhancement system was used.

The following table shows a summary of the activities performed

Summary of the activities performed				
Activity	Task	Date		
January 2016 - November 2016				
Setting up of the	1. Setting up of the	26/01/2016		
construction site area	construction site area			
	2. Laying of the fences			
Additional characterization	1. Drilling activity	26 and 27/01/2016		
of the LEV06 parcel of the	2. Supervision of soil			
area	sampling activities			
	3. Laboratory analysis			
Injection of oxidant	1. Preparation of the	First stage 30+31/05/2016		
(RegenOx)	oxidizing mixture	and 01/06/2016		
	2. Direct push into the soil	Second stage		
		11+13/07/2016		





Monitoring of the	1. Drilling activity	First investigation		
remediation progress	2. Supervision of soil	campaign 01/08/2016		
	sampling activities	Second investigation		
	3. Laboratory analysis	campaign 12/08/2016		
	,	Third investigation		
		campaign 29/08/2016		
Additional characterization	1. Drilling activity			
	2. Supervision of soil			
	sampling activities			
	3. Laboratory analysis			
May 2017 – August 2017				
Location of injection points	1. Identification of	22+23+24/05/2017		
and construction of	underground utilities	29/05/2017		
piezometers	2. Location of injection			
	points			
	3. Execution of drilling			
	4. Installation of			
	piezometers			
	5. Piezometer			
	development			
Injection of oxidant	1. Preparation of the	First stage 29+31/05/2017		
(Sodium Persulfate)	oxidizing mixture	Second stage		
	2. Injection activity	03+04/07/2017		
Monitoring of the	1. Drilling activity	First investigation		
remediation progress	2. Supervision of soil	campaign 09/08/2017		
	sampling activities			
	3. Laboratory analysis			





#### 5.4 Radius of influence

Additional surveys were planned to precisely limit the extent of the contamination of the area, within the intervention area defined on the basis of the Thiessen polygon.

The boreholes were arranged with a regular mesh (15 m x 15 m) around point LEV06. The perforations were pushed up to the "Caranto" (local name of a Pleistocene paleosoil consisting of an extremely compact, silty-sandy clay), the top of which is located in this area at a depth of approximately 4.5 m from the ground level. One sample per meter of thickness crossed was taken, to be analyzed in the laboratory.

Based on the analytical results, the area to be treated was defined and the quantities of contaminants present were estimated.

The range of influence has not been calculated. An estimate It was made from literature data as a function of low permeability of the soil and the test was carried out on a pilot-scale according to the supplementary characterization performed as in the first figure of the previous paragraph 4.3, verifying the effectiveness of the treatment as per the following paragraph

#### 5.5 Process and performance monitoring

To assess the actual degradation of the contaminant in the first injection cycle and for possibly define some corrective manoeuvres, samples of soils within the injection area were carried out fortnightly, during the 45 days following the injection.

In the second treatment cycle, the first check of the remediation progress was performed approximately 30 days after the last injection.





## 6. Post treatment and/or Long Term Monitoring

## 6.1 Post treatment and/or Long Term Monitoring

In the case study described, it was not necessary to provide for long-term monitoring due to the low mobility of the contaminants.

## **Glossary of Terms**

Term (alphabetical order)	Definition
CSC	Contamination threshold concentrations
CSR	Risk threshold concentrations
NAPL	Non-Aqueous Phase Liquids
SIN	Site of National Interest
VCM	Vinyl Chloride Monomer